

Avalon Rare Metals Inc.

RESPONSE TO THE APRIL 16, 2012 CLARIFICATIONS LETTER FROM MACKENZIE VALLEY ENVIRONMENTAL IMPACT REVIEW BOARD FOR THE THOR LAKE RARE EARTH ELEMENT PROJECT DEVELOPER'S ASSESSMENT REPORT

Submitted To: MACKENZIE VALLEY ENVIRONMENTAL IMPACT REVIEW BOARD

May 10, 2012



Avalon Rare Metals Inc. (Avalon) is pleased to provide the following responses to the Mackenzie Valley Environmental Impact Review Board (MVEIRB) letter dated April 16, 2012.

Avalon's responses are found after each information clarification request. For consistency, Avalon has kept the same request and response numbers as previously submitted in the April 2, 2012 submission, except for the new #4 (Footnote), which is a response to the MVEIRB request that appeared as a footnote.

IR Clarification: #1

Source: MVEIRB

To: Avalon Rare Metals Inc.

Preamble

Items #1 and #3 from the 22-March-12 letter also remain outstanding.

MVEIRB IR Clarification Request #1

Please submit results of both sublethal and chronic toxicity tests (as well as the associated acute toxicity test) on a sample of simulated effluent from the March 2012 pilot plant testing Avalon refers to in both Avalon Response #3.1 (in response to AANDC's IR# 3.1) and Avalon Response #23 (Environment Canada's IR#23).

Avalon IR Clarification Response #1

Avalon is pleased to provide its response to the above IR Clarification request and is providing the results of the acute and sub-lethal/chronic toxicity testing. Copies of all recent laboratory test results discussed in this response are provided in Attachment 1.

As stated in the submission of March 28, the flotation pilot plant testing objectives included producing sufficient concentrate for hydrometallurgical testing. Not unexpectedly, reagent consumption used was well above the expected operational consumption to ensure sufficient recovery, and resulted in elevated concentrations of some reagents in the final effluent. Avalon further stated that further flotation testing is ongoing to optimize (reduce) the reagent usage. Preliminary results have identified significant reduction and potential elimination of some reagents.

Using the pilot plant water, Avalon simulated the holding period for the effluent in the tailings management area consistent with the time-frame identified in the DAR. It then subjected the effluent to standard flotation plant treatment methodologies to remove these reagents to the extent practicable by proven technologies. These technologies are being incorporated into the project design, also as per commitments made by Avalon in the DAR.

Avalon reiterates that it continues to be successful in its ongoing work to refine the floatation process with either reduction or elimination of reagent use. Avalon further clarifies that if the ongoing optimization testing results in reagent use that eliminates the need for the water treatment system identified above, it can easily be removed from the design.



Testing Completed

Given this regulatory regime and the MVEIRB request, Avalon tested the following three waters for both acute and sub-lethal/chronic toxicity:

- 1. Treated effluent.
- 2. Treated effluent blended with Drizzle Lake water The amount of Drizzle Lake water was conservatively set at approximately 90% of the expected Lake:treated effluent ratio as per the hydrology model presented in the DAR (i.e., less Drizzle Lake water than is expected to be available for mixing). Drizzle Lake has been discussed as the location within which CCME guidelines parameter concentrations will be targeted.
- 3. Drizzle Lake water It has been reported that natural baseline water quality in Drizzle Lake does not consistently meet CCME guidelines.

All testing was done by AquaTox Testing and Consulting Inc., an independent Canadian laboratory certified for this testing.

The acute toxicity testing was conducted on Rainbow trout and *Daphnia magna* as per the Environment Canada and MMER protocols. An LC_{50} is the concentration of effluent in which 50% of the organisms do not survive over a prescribed time period. Avalon targets 100% survival.

In addition, the following sub-lethal/chronic toxicity tests were completed and assessed using the industry standard IC₂₅ (Inhibition Concentration) test, which is the concentration at which there is a 25% impairment of the parameter (e.g. growth rate or reproduction) being tested.

- 1. Test of larval growth and survival using Fathead minnow. (This also includes an acute toxicity component) The highest score available is an IC₂₅>100%, which indicates that full strength effluent does not result in more than a 25% reduction of growth or reproduction.
- 2. Test of Reproduction and Survival using the invertebrate *Ceriodaphnia dubia* (This also includes an acute toxicity component) The highest score available is an $IC_{25}>100\%$.
- 3. Growth inhibition using freshwater algae *Pseudokirchneiriella subcapitata*. The highest score available in this test is an $IC_{25} > 90.91\%$.
- 4. Growth inhibition using freshwater macrophyte *Lemna minor* (duck weed). The highest score available is >97%.

Acute Toxicity Testing Results

Acute toxicity testing resulted in 100% survival of *Daphnia magna*, rainbow trout, fathead minnow, and *Ceodaphnia dubia* (an invertebrate) in all waters tested. Rainbow Trout are among the most sensitive fish species, especially when compared with the fish species in Murky and Thor Lake, the two lakes immediately downstream of Drizzle Lake. These results are attached.

Acute toxicity testing that took place as part of sublethal/chronic toxicity testing also resulted in 100% survival of fathead minnows and *Cerodaphnia dubia*.

In summary, all acute toxicity testing, including 100% effluent, resulted in 100% survival of test organisms.



Sub-Lethal/Chronic Toxicity Testing

While there are no legal criteria for the sub-lethal/chronic toxicity tests, Avalon's assessments and reporting will be based on results from the IC_{25} tests. The following results were obtained from testing that has been completed to date.

The IC₂₅ for the fathead minnow was >100% in the final effluent, effluent mixed with Drizzle Lake water, and Drizzle Lake. Results in all three tests were very similar.

With respect to the sub-lethal/chronic reproduction test for the invertebrate *Cerodaphnia dubia*, some inhibition of reproduction rate was observed in the treated effluent (IC_{25} =63.5%), less inhibition was noted for the effluent/Drizzle Lake blend (IC_{25} =75%), and about a 15% inhibition was reported for the Drizzle Lake water only, though this equates to an IC_{25} of >100%. It is noted that there were no mortalities in the final effluent test, while there were some in the blended waters and Drizzle Lake water, but interestingly, only in diluted samples and not in the 100% concentration sample. It is also noted that all three waters behaved in a similar manner, with the impairment only in the sample of undiluted solutions.

The *Pseudokirchneiriella subcapitata* (algae) sub-lethal/chronic growth rate tests suggest that growth stimulation may have occurred as opposed to an impairment of the growth rate. All waters tested had IC25 values greater than 90.91%.

Sublethal/chronic toxicity tests involving the aquatic macrophyte *Lemna minor* have resulted in IC₂₅ values greater than 97% for all waters tested.

Water Quality Analysis Results

Water quality analysis results were consistently within federal regulated limits established by the Metal Mining Effluent Regulations (MMER). In addition, results for the treated effluent were also within Canadian Council of Ministers of the Environment (CCME) guidelines for the protection of freshwater aquatic life in the receiving environment for all measured parameters, with the exception of fluorine. This compares with levels of fluorine, suspended solids, cadmium, iron, zinc and ammonia, which exceeded CCME guidelines in background samples of Drizzle Lake water. Fluorine is significantly higher in the natural lake water, such that the effluent marginally improves the lake water quality respecting this element, suggesting that the host rock within the drainage may be a source of the fluorine.



Summary and Conclusions

The reported analytical results do not account for natural amelioration occurring as a result of physical, biological, and chemical processes in the natural environment. Such processes have the potential to further reduce the potential for acute and sub-lethal effects on aquatic organisms.

All acute toxicity tests achieved 100% survival of the test species. Three of the four sublethal/chronic tests showed no observable effect based on IC₂₅ results. The reproduction test showed only minor differences in potential impact between the effluent blended with Drizzle Lake and the Drizzle Lake water and at only the highest concentration tested. Given the available blending in the receiving waters, natural amelioration anticipated upstream of the Drizzle Lake discharge, the natural exceedences of CCME guidelines in Drizzle Lake, and the sensitivity and nature of these tests and test results, this information strongly indicates that there will be no impact on downstream water bodies.



IR Clarification: #2

Source: MVEIRB

To: Avalon Rare Metals Inc.

Preamble

On the subject of water quality objectives: the Review Board notes that environmental assessment is the appropriate forum for establishing water quality objectives.

The Review Board recognizes that Avalon – through its 2-April-12 letter – has proposed site-specific water quality objectives for those parameters that CCME (Canadian Council of Ministers of the Environment) guidelines cover including the location where Avalon proposes to meet the objectives: Drizzle Lake outflow. With regard to Avalon's reference to CCME guidelines as being site-specific water quality objectives (SSWQOs) for the Thor Lake Project mine-site – items #2a and #2b in the Review Board's 22-March-12 letter – the Review Board notes that CCME guidelines do not fully encompass the effluent Avalon would likely discharge downstream. In other words: for Avalon's Thor Lake Project, CCME guidelines represent an incomplete/deficient suite of site-specific water quality objectives. For example, CCME guidelines are silent for lanthanum and cerium: these two parameters are likely to be present in the effluent Avalon proposes to discharge into the downstream environment1. As such, items #2a and #2b from the 22-March-12 letter remain outstanding (including for corresponding rationale) for all elements, parameters, and substances that are likely present in the effluent Avalon proposes to discharge downstream but not covered through CCME guidelines.

MVEIRB IR Clarification Request #2

Please adequately address Aboriginal Affairs and Northern Development Canada's (AANDC) Information Request #1 by submitting:

- a. appropriate site-specific water quality objectives specific to the effluent likely to be discharged from the Thor Lake Rare Earth Element Project; and
- b. specific location(s) where Avalon proposes to meet the site-specific water quality objectives.

Avalon IR Clarification Response #2(a/b)

Avalon is pleased to advise that following receipt of the Board's April 16, 2012 letter on the three remaining pre-requisite items needed for the conduct of the technical sessions, EBA on behalf of Avalon proceeded to make contact with AANDC and Environment Canada (EC) to initiate discussions leading to the eventual development of site-specific water quality objectives (SSWQOs) for the receiving waters located downstream of the proposed Nechalacho Mine Tailings Management Facility.



Based on these contacts with AANDC and EC, an initial SSWQOs teleconference meeting was convened by AANDC on April 20th, 2012 to initiate discussions on this subject. To assist with the initial discussions, EBA provided a brief Technical Memo to AANDC and EC on April 18th, 2012, prior to the meeting, which summarized the existing background receiving water quality conditions of the Nechalacho Mine area. (Attachment 2 to this response).

The teleconference meeting was held on April 20th, 2012, and the notes of that meeting (dated May 1st, 2012) are provided as Attachment 3 to this response. As indicated in the meeting notes, the discussion concluded with a general consensus that the meeting was productive, that the CCME criteria could form an initial starting point for further discussions for the establishment of SSWQOs for this project, and that future meeting(s) would be useful to resolve issues related to the setting of SSWQOs for other parameters for which CCME criteria do not yet exist.

In particular, it was noted that the next round of baseline surface and groundwater sampling was being undertaken currently (April 2012) and that analysis of rare earth metals (REEs) would be undertaken to assist in establishing baseline conditions for these parameters. These results, which are also provided with this response document to the MVEIRB, will be used in subsequent meetings with AANDC and EC to help with the establishment of SSWQOs for these parameters.

In addition, to further facilitate dialogue on this matter, EBA provided a copy of a recently released report prepared by Wilfrid Laurier University entitled Review of Aquatic Effects of Lanthanides and Other Uncommon Elements to AANDC and EC. A copy of this report is provided as Attachment 4 to this response.



IR Clarification: #3

Source: MVEIRB

To: Avalon Rare Metals Inc.

Preamble

Items #1 and #3 from the 22-March-12 letter also remain outstanding. For clarification, Item #3 refers to identification of all water quality parameters present in the Item #1 test-effluents.

MVEIRB IR Clarification Request #3

Please submit the full suite of water quality test results associated with the March 2012 pilot plant testing Avalon refers to in both Avalon Response #3.1 (in response to AANDC's IR# 3.1) and Avalon Response #23 (Environment Canada's IR#23).

Avalon IR Clarification Response #3

As requested, Table 1 provided with this response presents the full suite of water quality testing that was conducted on the three solutions subjected to acute and sublethal testing as referred to in Avalon's response to IR Clarification Response 1:

- PP 22 BZ-MP Treated Process water(final effluent)
- Final effluent mixed with Drizzle Lake water
- Drizzle Lake water.

In providing this new Table 1 it should be noted that all of the metals parameter concentrations included in this table are comparable to or lower than the parameter concentrations previously provided in Table 6.4-1 of the DAR and Table 20 in Avalon's response to EC IR #13.1.

It should also be noted that this table is more extensive than any of the other tables previously provided and it includes total and dissolved concentrations of all of the rare earth elements associated with the mineral deposit.



Table 1: Solution Analyses

Parameter	Unit	MMER		nes for the of Aquatic Life	PP-22 BZ-MP Treated Process Water-2		PP-22 BZ-N	MP Treated	Lake Water	
raiailletei	Onit	2002	Short Term	Long Term			Process + I	Lake Water		
Radionuclides										
²²⁶ Ra	Bq/L	0.37			pending		pending		pending	
²²⁸ Ra	Bq/L				pending		pending		pending	
²¹⁰ Pb	Bq/L				pending		pending		pending	
LIMS					11276-	APR12	11276-	APR12	11276-	APR12
Temp on Rec	°C				15.0		15.0		15.0	
BOD5	mg/L	50 ¹			< 4		< 4		< 4	
BOD5	mg/L	50 ¹								
рН	units	6.0-9.5		6.0-9.5	7.97		8.52		8.45	
Alkalinity	mg/L as CaCO₃				129		257		260	
EMF	mV				220		168		195	
Acidity	mg/L as CaCO₃				< 2		< 2		< 2	
Conductivity	µS/cm				402		468		478	
TDS	mg/L				229		349		338	
TSS	_	45.00		5 mg/L from						24
Cl	mg/L	15.00		ground	2		18		0.0	24
SO ₄	mg/L	1	640	120	42		8.9		8.9	
F	mg/L	1		0.40	6.5		< 0.2		< 0.2	
NO ₂	mg/L as N mg/L			0.12 0.06	0.64 < 0.06		1.71 < 0.06		1.69 < 0.06	
NO ₃	as N mg/L			13	0.07		< 0.05		< 0.05	
NO ₂ +NO ₃	as N mg/L			13	0.07		< 0.05		< 0.05	
Tot.Reactive P	mg/L	<0.004	to >0.1 based	on lake type	0.07		< 0.03		` 0.00	



Table 1: Solution Analyses

Parameter	Unit	MMER		nes for the of Aquatic Life	PP-22 BZ-MP Treated		PP-22 BZ-	MP Treated	Lake Water	
raiailletei	Onit	2002	Short Term	Long Term	Process	Water-2	Process +	Lake Water	Euro Water	
TOC	mg/L				< 1.0		41.5		50.2	
NH ₃ +NH ₄	as N mg/L			0.20	< 0.1		1.4		1.4	
COD	mg/L	150 ¹			< 8		111		109	
Thiosalts	as S ₂ O ₃ mg/L				< 10		< 10		< 10	
Metals					Total	Dissolved	Total	Dissolved	Total	Dissolved
Hg	mg/L			0.000026	< 0.0001	< 0.0001	< 0.0001	< 0.0001	< 0.0001	< 0.0001
Ag	mg/L			0.0001	< 0.00001	< 0.00001	< 0.00001	< 0.00001	< 0.00001	< 0.00001
Al	mg/L			0.1	0.12	0.03	< 0.01	< 0.01	< 0.01	< 0.01
As	mg/L	0.50		0.005	0.0009	0.0010	0.0018	0.0021	0.0017	0.0022
Ва	mg/L				0.0062	0.0069	0.119	0.0815	0.115	0.0823
Ве	mg/L				< 0.00002	< 0.00002	< 0.00002	< 0.00002	< 0.00002	< 0.00002
В	mg/L		29	1.5	0.0084	0.0091	0.0368	0.0345	0.0362	0.0360
Bi	mg/L				< 0.00001	< 0.00001	< 0.00001	< 0.00001	< 0.00001	< 0.00001
Са	mg/L				22.2	22.5	56.0	52.1	54.9	52.8
Cd	mg/L			0.000025	< 0.000003	< 0.000003	0.000040	< 0.000003	0.000029	0.000003
Co	mg/L				0.000071	0.000071	0.000206	0.000127	0.000195	0.000134
Cr	mg/L				< 0.0005	< 0.0005	< 0.0005	< 0.0005	< 0.0005	< 0.0005
Cu	mg/L	0.30		0.002	0.0019	0.0014	0.0014	0.0018	0.0012	0.0038
Fe	mg/L			0.3	0.044	0.010	5.86	0.067	5.69	0.051
K	mg/L				30.8	28.6	3.71	3.52	3.56	3.54
Li	mg/L				0.011	0.010	0.010	0.009	0.010	0.010
Mg	mg/L				4.38	4.39	26.1	24.5	25.5	24.9
Mn	mg/L				0.117	0.125	0.810	0.380	0.790	0.387
Мо	mg/L				0.0062	0.0063	0.0024	0.0022	0.0022	0.0021
Na	mg/L				38.8	38.1	12.0	11.3	11.7	11.5



Table 1: Solution Analyses

Parameter	Unit	MMER		es for the of Aquatic Life	PP-22 BZ-MP Treated Process Water-2		PP-22 BZ-MP Treated		Lake Water	
rarameter	Onit	2002	Short Term	Long Term	Process	Water-2	Process +	Lake Water		
Ni	mg/L	0.50		0.08	0.0020	0.0018	0.0013	0.0012	0.0011	0.0018
Pb	mg/L	0.20		0.002	0.00092	0.00015	0.00038	0.00005	0.00028	0.00010
Sb	mg/L				< 0.0002	< 0.0002	< 0.0002	< 0.0002	< 0.0002	< 0.0002
Se	mg/L			0.001	< 0.001	< 0.001	< 0.001	< 0.001	< 0.001	< 0.001
Si	mg/L				7.61	7.53	6.82	5.83	6.63	5.91
Sn	mg/L				0.00002	< 0.00001	0.00003	< 0.00001	0.00002	0.00011
Sr	mg/L				0.0912	0.0918	0.0999	0.0935	0.0982	0.0936
Th	mg/L				0.000017	< 0.000004	0.000030	< 0.000004	0.000021	< 0.00000
Ti	mg/L				0.0003	0.0002	0.0004	0.0002	0.0003	0.0002
TI	mg/L			0.0008	< 0.0002	< 0.0002	< 0.0002	< 0.0002	< 0.0002	< 0.0002
U	mg/L		0.033	0.015	0.00001	0.00001	0.00015	0.00013	0.00015	0.00014
V	mg/L				0.00019	0.00013	0.00029	0.00020	0.00029	0.00035
Υ	mg/L				0.00025	0.00004	0.00016	0.00002	0.00012	0.00003
Zn	mg/L	0.50		0.03	0.028	0.023	0.084	0.008	0.080	0.008
Ce	mg/L				0.00092	0.00020	0.00082	< 0.00007	0.00056	0.00023
Dy	mg/L				0.000063	0.000015	0.000037	0.000003	0.000027	0.00000
Er	mg/L				0.000022	0.000004	0.000017	0.000002	0.000011	0.00000
Eu	mg/L				0.000014	0.000005	0.000017	0.000006	0.000014	0.00000
Ga	mg/L				0.00005	0.00004	0.00002	0.00004	0.00001	0.00001
Gd	mg/L				0.00011	< 0.00005	0.00006	< 0.00005	< 0.00005	< 0.0000
Hf	mg/L				< 0.000005	< 0.000005	0.000028	< 0.000005	0.000027	< 0.0000
Но	mg/L				0.000010	0.000002	0.000006	0.000001	0.000005	0.00000
La	mg/L				0.00041	0.00009	0.00038	< 0.00004	0.00026	0.00011
Lu	mg/L				0.000002	< 0.000001	0.000001	< 0.000001	0.000001	< 0.00000
Nb	mg/L				0.000045	0.000003	0.000089	0.000001	0.000065	0.00001



Table 1: Solution Analyses

Parameter	Unit	MMER		Guidelines for the Protection of Aquatic Life		MP Treated		MP Treated	Lake Water	
i arameter	Oille	2002	Short Term	Long Term	Process Water-2		Process +	Lake Water		
Nd	mg/L				0.00049	0.00049 0.00012		< 0.00003	0.00028	0.00012
Pr	mg/L				0.00011	0.00003	0.00010	< 0.00001	0.00007	0.00003
Sc	mg/L				0.00082	0.00072	0.00082	0.00062	0.00080	0.00063
Sm	mg/L				0.00011	0.00003	0.00008	< 0.00001	0.00006	0.00002
Та	mg/L				0.000009	0.000001	0.000020	0.000004	0.000021	0.000004
Tb	mg/L				0.000014	0.000005	0.000007	0.000006	0.000003	< 0.000001
Tm	mg/L				0.000003	< 0.000001	0.000002	0.000009	< 0.000001	< 0.000001
Yb	mg/L				0.000012	0.000002	0.000009	< 0.000002	0.000007	0.000002
Zr	mg/L				0.00007	0.00003	0.00013	0.00004	0.00012	0.00006
F2 (C10-C16)	μg/L				< 100		< 100		< 100	
F3 (C16-C34)	μg/L				< 500		< 500		< 500	
F4 (C34-C50)	μg/L				< 500		< 500		< 500	
Baseline at nC50	Yes / No				YES		YES		YES	

^{*}Department of Justice Canada. 2002. Metal Mining Effluent Regulations, Fisheries Act SOR-2002-222.

Available Online: http://laws.justice.gc.ca/en/F-14/SOR-2002-222/119716.html

¹World Bank Environment, Health and Safety Guidelines for Mining, 2007.



IR Clarification: #4 (Footnote)

Source: MVEIRB

To: Avalon Rare Metals Inc.

Preamble

The Review Board notes that in the original DAR Table 6.4-2, the modeled ratio of Plant Discharge to the Thor Lake concentration after 20 years was reported as about 1408:1, but was subsequently corrected to about 10:1 in Avalon's response to MVRB IR #1.2. In Table 1 of Avalon's Response #2.1 to IR AANDC #2, the ratios of Day 5 Decant Concentrations to the Max. Predicted Concentrations Years 1-20 (in Thor Lake) are reported as about 1320:1, apparently reflecting the error in calculation that appeared in the original Table 6.4-2.

MVEIRB IR Clarification Request #4 (Footnote)

Please confirm that Table 1 of Avalon's Response #2.1 to IR AANDC #2 is correct or submit a corrected response.

Avalon IR Clarification Response #4 (Footnote)

The original DAR Table 6.4-2 should be replaced by the following revised table.

Table 6.4-2: Average Concentration of Inert Tracer in The Thor Lake System

Year of Simulation	Plant Discharge	Tailings Pond	Polishing Pond	Drizzle Lake	Murky Lake	Thor Lake
1	1.00000	0.00091	0.00026	0.00004	0.00003	<0.00001
2	1.00001	0.00160	0.00073	0.00021	0.00017	0.00001
3	1.00004	0.00215	0.00119	0.00043	0.00037	0.00004
4	1.00009	0.00260	0.00164	0.00064	0.00058	0.00009
5	1.00016	0.00299	0.00208	0.00092	0.00085	0.00016
6	1.00024	0.00331	0.00241	0.00111	0.00104	0.00024
7	1.00031	0.00360	0.00269	0.00126	0.00119	0.00031
8	1.00038	0.00386	0.00292	0.00138	0.00132	0.00038
9	1.00044	0.00408	0.00313	0.00152	0.00144	0.00044
10	1.00050	0.00423	0.00330	0.00159	0.00152	0.00050
11	1.00057	0.00437	0.00342	0.00178	0.00159	0.00057
12	1.00058	0.00455	0.00355	0.00179	0.00166	0.00058
13	1.00061	0.00466	0.00369	0.00180	0.00171	0.00061
14	1.00063	0.00477	0.00379	0.00185	0.00177	0.00063
15	1.00066	0.00485	0.00387	0.00190	0.00183	0.00066
16	1.00070	0.00492	0.00394	0.00199	0.00186	0.00070
17	1.00068	0.00500	0.00392	0.00194	0.00186	0.00068
18	1.00067	0.00500	0.00389	0.00191	0.00176	0.00067
19	1.00070	0.00504	0.00400	0.00199	0.00186	0.00070
20	1.00071	0.00508	0.00408	0.00207	0.00191	0.00071



The difference between this table and the original table is the second column 'Plant Discharge Concentration', which now includes the effect of recirculating water from Thor Lake through the plant.

The dilution ratio of 1,408 was and still is correct. However, the ratio of 10:1 stated in Avalon's Response #1.2, was stated in error. Therefore, for Table 1 in Avalon's Response #2.1 to AANDC's Information Requests, the dilution ratio that should be used to calculate the maximum predicted concentration for Year 1-20 should be 1,408.

As a result of changes to Table 6.4-2, Table 6.4-3 in the DAR report should also be replaced by the following table, as the effects of background concentration are now included.

Table 6.4-3: Maximum Metal Concentration in The Thor Lake System in Year 20 and Water Quality Guidelines for The Metals of Concern

Metal	Backgro	ound Conce	ntration	Thor	Murky	Drizzle	CCME	MMER Effluent
Species	Thor Lake	Murky Lake	Drizzle Lake	Lake	Lake	Lake	Water Quality Guideline	Criteria
Hg (mg/L)	0.000010	0.000010	0.000005	0.000010	0.000010	0.000005	0.000026	-
Ag (mg/L)	0.00001	0.00001	0.00001	0.0001	0.00001	0.00001	0.0001	-
Al (mg/L)	0.0033	0.0072	0.0083	0.0037	0.0084	0.0096	0.1	-
As (mg/L)	0.00077	0.00129	0.00092	0.00077	0.00129	0.00092	0.005	0.5
Ba (mg/L)	0.0655	0.0636	0.0629	0.0655	0.0636	0.0629	-	-
Be (mg/L)	0.00025	0.00025	0.00025	0.00025	0.00025	0.00025	-	-
B (mg/L)	0.026	0.025	0.026	0.026	0.025	0.026	-	-
Bi (mg/L)	0.00025	0.00025	0.00025	0.00025	0.00025	0.00025	-	-
Ca (mg/L)	35.1	36.6	30.8	35.1	36.7	30.9	-	-
Cd (mg/L)	0.00002	0.00002	0.00001	0.00002	0.00002	0.00001	0.00002 - 0.00013	1
Co (mg/L)	0.00005	0.00009	0.00017	0.00005	0.00009	0.00017	-	-
Cr (mg/L)	0.00028	0.00025	0.00025	0.00028	0.00025	0.00025	0.0089	-
Cu (mg/L)	0.00036	0.00036	0.00025	0.00036	0.00036	0.00025	0.002 -0.004	0.3
Fe (mg/L)	0.070	3.054	1.091	0.070	3.055	1.092	0.3	-
K (mg/L)	2.0	1.8	1.3	2.0	1.9	1.4	-	-
Li (mg/L)	0.0054	0.0051	0.0041	0.0054	0.0052	0.0042	-	-
Mg (mg/L)	18.8	17.8	15.4	18.8	17.8	15.4	-	-
Mn (mg/L)	0.0414	0.2476	0.1872	0.0415	0.2478	0.1874	-	-
Mo (mg/L)	0.00210	0.00142	0.00127	0.00213	0.00151	0.00137	0.073	-
Na (mg/L)	6.8	7.1	6.1	6.8	7.2	6.2	-	-
Ni (mg/L)	0.00025	0.00025	0.00025	0.00025	0.00026	0.00026	0.025 -0.150	0.5
Pb (mg/L)	0.000050	0.000030	0.000028	0.000050	0.000031	0.000029	0.001 -0.007	0.2
Sb (mg/L)	0.00005	0.00005	0.00005	0.00005	0.00006	0.00006	-	-
Se (mg/L)	0.0005	0.0005	0.0005	0.0005	0.00005	0.00005	0.001	-
Si (mg/L)	3.28	2.17	2.37	3.28	2.19	2.39	-	-



Table 6.4-3: Maximum Metal Concentration in The Thor Lake System in Year 20 and Water Quality Guidelines for The Metals of Concern

Sn (mg/L)	0.00008	0.00005	0.00008	0.00008	0.00005	0.00008	-	-
Sr (mg/L)	0.0581	0.0562	0.0523	0.0582	0.0567	0.0528	-	-
Ti (mg/L)	0.005	0.005	0.005	0.005	0.005	0.005	-	-
TI (mg/L)	0.00005	0.00005	0.00005	0.00005	0.00005	0.00005	0.0008	-
U (mg/L)	0.000357	0.000098	0.000077	0.000363	0.000115	0.000095	0.015	-
V (mg/L)	0.0005	0.0005	0.0005	0.0005	0.0005	0.0005	-	-
Zn (mg/L)	0.0014	0.0023	0.0009	0.0014	0.0023	0.0009	0.03	0.5



ATTACHMENTS

Attachment 1: Acute and chronic bioassay test results on simulated Flotation plant TMF

effluent, effluent/Drizzle Lake mix, and Drizzle Lake water

Attachment 2: EBA, A Tetra Tech Company (EBA). April 18, 2012. Background Water

Quality Assessment – Avalon Thor Lake Project. Technical Memo.

Attachment 3: EBA, A Tetra Tech Company (EBA). May 1, 2012. April 20, 2012 Meeting

Notes: Avalon Thor Lake Site Specific Water Quality Objectives.

Attachment 4: Ng, T., D. S. Smith, A. Straus, and J. C. McGeer. March 31, 2011. Review

of Aquatic Effects of Lanthanides and Other Uncommon Elements. Final Project Report prepared for the EC Contribution Agreement with the CNTC

for 2010/2011 by Wilfrid Laurier University.



Attachment 1



Acute Bioassay Test Results: Rainbow Trout



11B Nicholas Beaver Rd. RR 3 Gueiph ON N1H 6H9

Tel: (519) 763-4412 Fax: (519) 763-4419

TOXICITY TEST REPORT **Rainbow Trout**

Page 1 of 2

Work Order: Sample Number: 221256 33957

SAMPLE IDENTIFICATION

Company: Location:

SGS Lakefield Research Limited

Lakefield ON

Substance: PP-22 BZ-MP Treated Process Water-2 Sampling Method:

Grab

Sampled By:

B. Bowman

Sample Description:

Clear, colourless, odourless.

Test Method:

Reference Method for Determining Acute Lethality of Liquid Effluents to Rainbow Trout.

Environment Canada, EPS 1/RM/13 (2nd Edition, December 2000, with May 2007 amendments).

TEST RESULTS

Effect

Value

95% Confidence Limits

Slope

Calculation Method

2012-04-23 2012-04-24

2012-04-25

Time Collected: 15:50

Temp. on arrival: 13.0 °C

Date Collected:

Date Received:

Date Tested:

96-h LC50

>100%

The results reported relate only to the sample tested.

POTASSIUM CHLORIDE REFERENCE TOXICANT DATA

Organism Batch:

T12-04

Date Tested (yyyy-mm-dd):

2012-04-19

Historical Mean LC50:

3817 mg/L

LC50 (95% Confidence Limits):

3975 mg/L (3644 - 4336)

Warning Limits (\pm 2SD):

3245 - 4489 mg/L

Statistical Method:

Spearman-Kärber

Analyst(s):

FS/CN/DK/JGG

TEST FISH

Control Fish Sample Size:

Mean Fish Weight (± 2 SD): Range of Weights:

Fish Loading Rate:

10

 $0.70 \pm 0.12 \text{ g}$

0.64 - 0.82 g

Cumulative stock tank mortality: Mean Fish Fork Length (± 2 SD): 0 % (prev. 7 days) $42.0 \pm 3.3 \text{ mm}$

Range of Fork Lengths:

40 - 45 mm

0.4 g/L

TEST CONDITIONS

Test Organism:

Oncorhynchus mykiss

Volume Tested (L):

16

Sample Treatment: pH Adjustment:

None None Number of Replicates: Organisms Per Replicate: 1 10

Test Aeration:

Yes

Total Organisms Per Test Level:

10

Pre-aeration/Aeration Rate:

 $6.5 \pm 1 \text{ mL/min/L}$

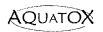
Test Method Deviation(s):

None

Date: 12-05-

Approved by:

Accredited by the Canadian Association for Laboratory Accreditation Inc. (CALA)



TOXICITY TEST REPORT Rainbow Trout

Page 2 of 2

Work Order:	221256
Sample Number:	33957

Total Pre-Acration			pН	D.O.	Cond.	Temp.	O ₂ Sat. (%)*
Time (h)	Initial W	ater Chemistry:	7.4	7.9	396	15.5	_
0:30		y after 30min air:	7.6	8.4	401	15.0	87
0.50	CHEHIISU	y arter Johan au.	0 hours	0.4	401	13.0	07
Date & Time Technician:	2012-04-25 FS	10:10	o nours				
Test Conc. (%)	Mortality	Immobility	рН	D.O.	Cond.	Temp.	O ₂ Sat. (%)*
100	0	0	7.6	8.4	401	15.0	87
50	0	0	8.0	9.0	615	14.5	
25	0	0	8.1	9.2	690	14.5	
12.5	0	0	8.2	9.4	735	14.5	
6.25	0	0	8.2	9.4	755	14.5	
Control	0	0	8.3	9.7	772	14.5	100
Notes:							
			24 hours				
Date & Time	2012-04-26	10:10					
Technician:	FS						
Test Conc. (%)	Mortality	Immobility	pН	D.O.	Cond.	Temp.	
100	0	0	_		_	15.0	
50	0	0	_	atmos.	_	15.0	
25	0	0	_		_	15.0	
12.5	0	0	_	****		15.0	
6.25	0	0	-	****	_	15.0	
Control	0	0	_		-	15.0	
Notes:							
***************************************			48 hours				
Date & Time Technician:	2012-04-27 FS	10:10					
Test Conc. (%)	Mortality	Immobility	pН	D.O.	Cond.	Temp.	
100	0	0			_	14.5	
50	0	0			_	14.5	
25	0	0	_			14.5	
12.5	0	0	_	***		14.5	
6.25	0	0	_			14.5	
Control	0	0	_		_	14.5	
Notes:							
	· · · · · · · · · · · · · · · · · · ·		72 hours				
Date & Time Technician:	2012-04-28 FS	10:10	72 110413				
Test Conc. (%)	Mortality	Immobility	рН	D.O.	Cond.	Temp.	
100	0	0	_	_	_	14.5	
50	0	0	_	_	_	14.5	
25	0	0	-	_	_	14.5	
12.5	0	0		_	_	14.5	
6.25	0	0	- maintings		_	14.5	
Control	0	0			_	14.5	
Notes:							
			96 hours				
Date & Time Technician:	2012-04-29 FS	10:10	, o mouro				
Test Conc. (%)	Mortality	Immobility	рН	D.O.	Cond.	Temp.	
100	0	0	8.4	9.1	406	14.0	
50	0	Ö	8.4	8.9	626	14.0	
25	0	Ö	8.4	9.2	705	14.0	
12.5	0	0	8.4	9.0	722	14.0	
6.25	0	0	8.4	9.1	730	14.0	
Control	0	0	8.3	9.2	740	14.0	
	v	Ü	0.5	7.2	<i>i</i> ⊣r∪	17.0	
Notes:							
" 0							

of control organisms showing stress: Trout Batch #: T12-04

Number immobile does not include number of mortalities.

Test Data Reviewed By: Francisco Date: Constant Office See

^{*} adjusted for actual temp. & barometric pressure

[&]quot;-" = not measured



11B Nicholas Beaver Rd. RR 3 Guelph ON N1H 6H9

Tel: (519) 763-4412 Fax: (519) 763-4419

TOXICITY TEST REPORT Rainbow Trout

Page 1 of 2

Sample Number:

221256 33958

SAMPLE IDENTIFICATION

Company: Location:

Work Order:

SGS Lakefield Research Limited

PP-22 BZ-MP Treated Process + Lake Water

Lakefield ON

Substance: Sampling Method:

Grab

Sampled By:

B. Bowman

Sample Description:

Clear, orange, odourless.

Test Method:

Reference Method for Determining Acute Lethality of Liquid Effluents to Rainbow Trout.

Environment Canada, EPS 1/RM/13 (2nd Edition, December 2000, with May 2007 amendments).

TEST RESULTS

Effect

Value

95% Confidence Limits

Slope

Calculation Method

2012-04-23

2012-04-24

2012-04-25

96-h LC50

>100%

The results reported relate only to the sample tested.

POTASSIUM CHLORIDE REFERENCE TOXICANT DATA

Organism Batch:

T12-04

Date Tested (yyyy-mm-dd): LC50 (95% Confidence Limits): 2012-04-19

Historical Mean LC50:

3817 mg/L

3975 mg/L (3644 - 4336)

Warning Limits (\pm 2SD):

3245 - 4489 mg/L

Statistical Method:

Spearman-Kärber

Analyst(s):

FS/CN/DK/JGG

TEST FISH

Control Fish Sample Size:

Mean Fish Weight (± 2 SD):

0.31 - 0.61 g

Cumulative stock tank mortality:

Mean Fish Fork Length (± 2 SD):

Time Collected: 15:50

Temp. on arrival: 13.0 °C

Date Collected:

Date Received:

Date Tested:

0 % (prev. 7 days) $36.3 \pm 6.1 \text{ mm}$

Range of Weights:

 $0.42 \pm 0.21 \text{ g}$

Range of Fork Lengths:

32 - 41 mm

Fish Loading Rate:

0.4 g/L

TEST CONDITIONS

Test Organism:

Sample Treatment:

Oncorhynchus mykiss None

Volume Tested (L): Number of Replicates: Organisms Per Replicate: 10 1

pH Adjustment: Test Aeration:

None Yes

Total Organisms Per Test Level:

10 10

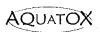
Pre-aeration/Aeration Rate:

 $6.5 \pm 1 \text{ mL/min/L}$

Test Method Deviation(s):

None

Approved by:



Work Order:

Sample Number:

221256

33958

TOXICITY TEST REPORT Rainbow Trout

Page 2 of 2

omipio	20,20						
Total Pre-Aeration			pН	D.O. (mg/L)	Cond. (µmhos/cm)	Temp. (°C)	O ₂ Sat. (%)*
Time (h)	Initial W	ater Chemistry:	8.4	8.3	479	15.0	-
0:30	Chemistr	y after 30min air:	8.5	8.9	481	14.5	93
			0 hours				
Date & Time	2012-04-25	10:10					
Technician:	FS					_	
Test Conc. (%)	•	Immobility	pH	D.O.	Cond.	Temp.	O ₂ Sat. (%)
100	0	0	8.5	8.9	481	14.5	93
50	0	0	8.4	9.2	646	14.5	
25	0	0	8.3	9.3	711	14.5	
12.5	0	0	8.2	9.4	745	14.5	
6.25	0	0	8.2	9.5	753	14.5	
Control	0	0	8.3	9.7	772	14.5	100
Notes:							
			24 hours				
Date & Time	2012-04-26	10:10					
Technician:	FS						
Test Conc. (%)	Mortality		pН	D.O.	Cond.	Temp.	
100	0	0	_			15.0	
50	0	0	_		-	15.0	
25	0	0	_	-		15.0	
12.5	0	0	_	_	****	15.0	
6.25	0	0	_	_	****	15.0	
Control	0	0		_	_	15.0	
Notes:							
		·	48 hours				
Date & Time	2012-04-27	10:10	10 110 111				
Technician:	FS						
Test Conc. (%)	Mortality	Immobility	pН	D.O.	Cond.	Temp.	
100	0	0	_	***	_	14.5	
50	0	0	_		_	14.5	
25	0	0	_		_	14.5	
12.5	0	0	-		_	14.5	
6.25	0	0	-		_	14.5	
Control	0	0			_	14.5	
Notes:							
			72 hours	·		**********	·····
Date & Time	2012-04-28	10:10	/2 mours				
Technician:	FS						
Test Conc. (%)	Mortality	Immobility	pН	D.O.	Cond.	Temp.	
100	0	0	·_	man.	_	14.5	
50	0	0	_		_	14.5	
25	0	0		marks.	_	14.5	
12.5	0	0		_	_	14.5	
6.25	0	0		_	_	14.5	
Control	0	0	disease.		_	14.5	
Notes:							
110005.			96 hours				
Date & Time	2012-04-29	10.10	20 Hours				
Technician:	FS	10.10					
Test Conc. (%)	Mortality	Immobility	pН	D.O.	Cond.	Temp.	
100	0	0	8.5	9.2	491	14.0	
50	0	0	8.5	9.0	650	14.0	
25	0	0	8.5	9.0	733	14.0	
12.5	0	0	8.4	9.1 8.9	763	14.0	
6.25	0	0	8.4	8.9	703 776	14.0	
Control	0	0	8.3	9.0	776 779	14.0	
Control	U	v	0.3	7 ()			
Notes:				3.0	,,,	11.0	

of control organisms showing stress: 0
Trout Batch #: T12-04

Number immobile does not include number of mortalities.

Test Data Reviewed By: State: 2012-014-30

^{*} adjusted for actual temp. & barometric pressure

[&]quot;_" = not measured



11B Nicholas Beaver Rd. RR 3 Guelph ON N1H 6H9

Tel: (519) 763-4412 Fax: (519) 763-4419

TOXICITY TEST REPORT Rainbow Trout

Page 1 of 2

Work Order: Sample Number: 221256 33959

SAMPLE IDENTIFICATION

Company:

SGS Lakefield Research Limited

Time Collected: 15:50 Date Collected:

2012-04-23 2012-04-24

Location: Substance: Lakefield ON Lake Water

Date Received:

Sampling Method: Sampled By:

Grab B. Bowman Date Tested: Temp. on arrival: 13.0 °C

2012-04-25

Sample Description:

Cloudy, brown, odourless.

Test Method:

Reference Method for Determining Acute Lethality of Liquid Effluents to Rainbow Trout.

Environment Canada, EPS 1/RM/13 (2nd Edition, December 2000, with May 2007 amendments).

TEST RESULTS

Effect

Value

95% Confidence Limits

Slope

Calculation Method

96-h LC50

>100%

The results reported relate only to the sample tested.

POTASSIUM CHLORIDE REFERENCE TOXICANT DATA

Organism Batch:

T12-04

Date Tested (yyyy-mm-dd):

2012-04-19

Historical Mean LC50:

3817 mg/L

LC50 (95% Confidence Limits):

3975 mg/L (3644 - 4336)

Warning Limits (± 2SD):

3245 - 4489 mg/L

Statistical Method:

Spearman-Kärber

Analyst(s):

FS/CN/DK/JGG

TEST FISH

Control Fish Sample Size:

Mean Fish Weight (± 2 SD):

10 0.47 ± 0.21 g Cumulative stock tank mortality: Mean Fish Fork Length (± 2 SD): 0 % (prev. 7 days)

Range of Weights:

0.34 - 0.62 g

Range of Fork Lengths:

 $37.6 \pm 4.5 \text{ mm}$

Fish Loading Rate:

0.4 g/L

34 - 40 mm

TEST CONDITIONS

Test Organism:

Oncorhynchus mykiss

Volume Tested (L):

12

Sample Treatment:

None

Number of Replicates:

1

pH Adjustment: Test Aeration:

None

Organisms Per Replicate: Total Organisms Per Test Level: 10 10

Pre-aeration/Aeration Rate:

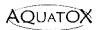
Yes $6.5 \pm 1 \text{ mL/min/L}$

Test Method Deviation(s):

None

Approved by:

Accredited by the Canadian Association for Laboratory Accreditation Inc. (CALA)



TOXICITY TEST REPORT **Rainbow Trout**

Work Order: Sample Number: 221256 33959

Page 2 of 2

ounpro reamour.	55767						
Total Pre-Aeration			рН	D.O. (mg/L)	Cond. (µmhos/cm)	Temp. (°C)	O ₂ Sat. (%)
Time (h)	Initial W	ater Chemistry:	8.5	8.7	479	15.0	_
0:30	Chemistry	y after 30min air:	8.6	9.3	481	15.0	96
			0 hours				
Date & Time Technician:	2012-04-25 FS	10:10	0 110415				
Test Conc. (%)	Mortality	Immobility	рH	D.O.	Cond.	Temp.	O ₂ Sat. (%)
100	0	0	8.6	9.3	481	15.0	96
50	0	0	8.4	9.3	644	14.5	70
25	0	0	8.4	9.3	721	14.5	
12.5	0	0	8.3	9.3	752	14.5	
6.25	0	0	8.4	9.3	759	14.5	
Control	0	0	8.3	9.7	772	14.5	100
Notes:	Ü	Ü	0.5	· · · ·	,,,	11.0	100
110003.			24 hours				
Date & Time	2012-04-26	10.10	24 Hours				
Technician:	FS	10.10					
Test Conc. (%)	Mortality	Immobility	pН	D.O.	Cond.	Temp.	
100	0	0		_	-	15.0	
50	Ö	0	_	_	_	15.0	
25	Ö	Ö	_	_	_	15.0	
12.5	ő	ő	_	_		15.0	
6.25	0	o 0				15.0	
Control	ő	0				15.0	
	Ü	Ü				13.0	
Notes:			40.1				
Data & Time	2012 04 27	10.10	48 hours				
Date & Time Technician:	2012-04-27 FS						
Test Conc. (%)	Mortality	Immobility	pН	D.O.	Cond.	Temp.	
100	0	0	_	_	_	14.5	
50	0	0	_	_	_	14.5	
25	0	0	_	_	_	14.5	
12.5	0	0	-	_	_	14.5	
6.25	0	0	-	_	_	14.5	
Control	0	0	-	_	_	14.5	
Notes:							
			72 hours				
Date & Time Technician:	2012-04-28 FS	10:10					
Test Conc. (%)	Mortality	Immobility	pН	D.O.	Cond.	Temp.	
100	0	0				14.5	
50	ŏ	ő	_	****		14.5	
25	0	ő				14.5	
12.5	ő	ő	***			14.5	
6.25	Ö	0	***	_		14.5	
Control	ő	ő		_		14.5	
	Ü	Ü				14.5	
Notes:					 		·····
Data & Time	2012 04 20	10.10	96 hours				
Date & Time	2012 - 04-29	10:10					
Technician:	FS Montality	Immobility	nII	D.O	Comd	Т	
Test Conc. (%)	Mortality	Immobility	p H	D.O.	Cond.	Temp.	
100	0	0	8.5	8.3	490	14.0	
50	0	0	8.5	8.4	651	14.0	
25	0	0	8.4	8.5	715	14.0	
12.5	0	0	8.4	8.6	753 763	14.0	
6.25	0	0	8.4	8.6	763	14.0	
Control	0	0	8.3	9.0	770	14.0	
Notes:							

0

T12-04 Number immobile does not include number of mortalities. * adjusted for actual temp. & barometric pressure

of control organisms showing stress:

"-" = not measured

Trout Batch #:



Acute Bioassay Test Results: Daphnia Magna



11B Nicholas Beaver Rd.

RR 3

Guelph ON N1H 6H9

Tel: (519) 763-4412 Fax: (519) 763-4419

TOXICITY TEST REPORT

Daphnia magna Page 1 of 2

Work Order: Sample Number: 221256 33957

SAMPLE IDENTIFICATION

Company:

SGS Lakefield Research Limited

Location:

Lakefield ON

Substance:

PP-22 BZ-MP Treated Process Water-2

Sampling Method:

Grab

Sampled By:

B. Bowman

Sample Description: Clear, colourless, odourless.

Test Method:

Reference Method for Determining Acute Lethality of Effluents to Daphnia magna. Environment

Canada EPS 1/RM/14 (Second Edition, December 2000).

TEST RESULTS

Effect

Value

95% Confidence Limits

Slope

Calculation Method

2012-04-23

2012-04-24

2012-04-24

Time Collected: 15:50

Temp. on arrival: 13.0° C

Date Collected:

Date Received:

Date Tested:

48-h LC50

>100%

The results reported relate only to the sample tested.

SODIUM CHLORIDE REFERENCE TOXICANT DATA

Organism Batch:

Dm12-07

Date Tested (yyyy/mm/dd):

2012-04-17

Historical Mean LC50:

6.4 g/L

LC50 (95% Confidence Limits):

6.9 g/L (6.6 - 7.3)

Warning Limits (± 2SD):

5.9 - 7.0 g/L

Statistical Method:

Spearman-Kärber

Analyst(s):

NK

Daphnia magna CULTURE HEALTH DATA

Time to First Brood:

9.2 days

Mean Young Per Brood:

29.7

Culture Mortality:

0% (previous 7 days)

TEST CONDITIONS

Sample Treatment: pH Adjustment:

None None Number of Replicates:

1

Test Aeration:

None

Test Organisms / Replicate: Total Organisms / Test Level: 10 10

Organism Batch:

Dm12-07

Organism Loading Rate:

15.0 mL/organism

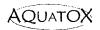
Test Method Deviation(s):

None

Approved by:

Accredited by the Canadian Association for Laboratory Accreditation Inc. (CALA)





Daphnia magna

Page 2 of 2

Work Order: Sample Number: 221256 33957

	Hardness (mg/L as CaCO ₃)	Hardness Adjustment	рН	D.O. (mg/L)	Cond.	Temp.	O ₂ Sat. (%)*	Total Pre-Aeration Time (h) @ 30 mL/min/L
Initial Water Chemistry:	90	None	7.5	7.9	423	21.0	93	0:00
			0 hours				· · · · · · · · · · · · · · · · · · ·	
Date & Time Technician:	2012-04-24 AW	14:20						
Test Conc. (%)	Mortality	Immobility	pН	D.O.	Cond.	Temp.	O ₂ Sat. (%)*	Hardness
100	0	0	7.5	7.9	423	21.0	93	90
50	0	0	7.9	8.3	436	21.0		
25	0	0	8.1	8.3	442	21.0		
12.5	0	0	8.3	8.3	444	21.0		
6.25	0	0	8.3	8.2	445	21.0	0.7	200
Control	0	0	8.3	8.2	445	21.0	97	200
Notes:								
	· · · · · · · · · · · · · · · · · · ·		24 hours					
Date & Time Technician:	2012-04-25 NK	14:20						
Test Conc. (%)	Mortality	Immobility	pН	D.O.	Cond.	Temp.		
100	0	1	_	-		20.5		
50	0	0		_	_	20.5		
25	0	0	_			20.5		
12.5	0	0	_	-	_	20.5		
6.25	0	0	-		_	20.5		
Control	0	0	_	_		20.5		
Notes:								
	· · · · · · · · · · · · · · · · · · ·		48 hours					
Date & Time Technician:	2012-04-26 AW	14:20						
Test Conc. (%)	Mortality	Immobility	pН	D.O.	Cond.	Temp.		
100	0	1	8.5	8.4	424	21.0		
50	0	0	8.5	8.5	440	21.0		
25	0	0	8.5	8.4	446	21.0		
12.5	0	0	8.5	8.5	447	21.0		
6.25	0	0	8.5	8.4	450	21.0		
Control	0	0	8.5	8.5	462	21.0		

Notes:

of control organisms showing stress: 0
Daphnia Batch #: Dm12-07

Number immobile does not include number of mortalities.

⁻ = not measured

^{*} adjusted for actual temp. & barometric pressure



11B Nicholas Beaver Rd. RR 3

Guelph ON N1H 6H9

Tel: (519) 763-4412 Fax: (519) 763-4419

TOXICITY TEST REPORT

Daphnia magna Page 1 of 2

Work Order: Sample Number: 221256 33958

SAMPLE IDENTIFICATION

Company:

SGS Lakefield Research Limited

Location:

Lakefield ON

Substance:

PP-22 BZ-MP Treated Process + Lake Water

Sampling Method: Sampled By:

Grab

Sample Description: Clear, orange, odourless.

B. Bowman

Test Method:

Reference Method for Determining Acute Lethality of Effluents to Daphnia magna. Environment

Canada EPS 1/RM/14 (Second Edition, December 2000).

TEST RESULTS

Effect

Value

95% Confidence Limits

Slope

Calculation Method

2012-04-23

2012-04-24

2012-04-24

Time Collected: 15:50

Temp. on arrival: 13.0° C

Date Collected:

Date Received:

Date Tested:

48-h LC50

>100%

The results reported relate only to the sample tested.

SODIUM CHLORIDE REFERENCE TOXICANT DATA

Organism Batch:

Dm12-07

Date Tested (yyyy/mm/dd):

2012-04-17

Historical Mean LC50:

6.4 g/L

LC50 (95% Confidence Limits):

6.9 g/L (6.6 - 7.3)

Warning Limits (\pm 2SD):

5.9 - 7.0 g/L

Statistical Method:

Spearman-Kärber

Analyst(s):

NK

Daphnia magna CULTURE HEALTH DATA

Time to First Brood:

9.2 days

Mean Young Per Brood:

29.7

Culture Mortality:

0% (previous 7 days)

TEST CONDITIONS

Sample Treatment:

pH Adjustment:

None

Number of Replicates:

1

None

Test Organisms / Replicate:

10

Test Aeration:

None

Total Organisms / Test Level:

10

Organism Batch:

Dm12-07

Organism Loading Rate:

15.0 mL/organism

Test Method Deviation(s):

None

Approved by

Project Mañager

Accredited by the Canadian Association for Laboratory Acereditation Inc. (CALA)



TOXICITY TEST REPORT

Daphnia magna

Page 2 of 2

Work Order: Sample Number: 221256 33958

	Hardness (mg/L as CaCO ₃)	Hardness Adjustment	pH	D.O. (mg/L)	Cond.	Temp.	O ₂ Sat.	Total Pre-Aeration Time (h) @ 30 mL/min/
Initial Water Chemistry:	250	None	8.5	9.1	499	21.0	106	0:30
Date & Time	2012-04-24	14:35	0 hours					
Technician:	AW	14:55						
Test Conc. (%)	Mortality	Immobility	pН	D.O.	Cond.	Temp.	O ₂ Sat. (%)*	Hardness
100	0	0	8.6	8.6	498	21.0	99	250
50	0	0	8.5	8.4	474	21.0		
25	0	0	8.5	8.3	461	21.0		
12.5	0	0	8.4	8.4	454	21.0		
6.25	0	0	8.4	8.3	451	21.0	0.5	200
Control	0	0	8.3	8.2	445	21.0	97	200
Notes:								
			24 hours				, ,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	······································
Date & Time Technician:	2012-04-25 NK	14:35						
Test Conc. (%)	Mortality	Immobility	pН	D.O.	Cond.	Temp.		
100	0	0	_	-	****	20.5		
50	0	0		-	-	20.5		
25	0	0	_		manu.	20.5		
12.5	0	0			_	20.5		
6.25	0	0	_	_		20.5		
Control	0	0	***	*****	_	20.5		
Notes:								
			48 hours					
Date & Time Technician:	2012-04-26 AW	14:35						
Гest Conc. (%)	Mortality	Immobility	pН	D.O.	Cond.	Temp.		
100	0	0	8.7	8.3	501	21.0		
50	0	0	8.6	8.4	475	21.0		
25	0	0	8.6	8.6	463	21.0		
2.5	0	0	8.6	8.5	459	21.0		
5.25	0	0	8.5	8.4	455	21.0		
Control	0	0	8.5	8.5	454	21.0		

Notes:

of control organisms showing stress: 0

Daphnia Batch #: Dm12-07

Number immobile does not include number of mortalities.

⁻ = not measured

^{*} adjusted for actual temp. & barometric pressure



11B Nicholas Beaver Rd. RR 3 Guelph ON N1H 6H9

Tel: (519) 763-4412 Fax: (519) 763-4419

TOXICITY TEST REPORT

Daphnia magna Page 1 of 2

Work Order: Sample Number: 221256 33959

SAMPLE IDENTIFICATION

Company:

SGS Lakefield Research Limited

Location:

Lakefield ON

Substance:

Lake Water

Sampling Method: Sampled By:

Grab

B. Bowman

Test Method:

Sample Description: Cloudy, brown, odourless.

Reference Method for Determining Acute Lethality of Effluents to Daphnia magna. Environment

Canada EPS 1/RM/14 (Second Edition, December 2000).

TEST RESULTS

Effect

Value

95% Confidence Limits

Slope

Calculation Method

2012-04-23

2012-04-24

2012-04-24

Time Collected: 15:50

Temp. on arrival: 13.0° C

Date Collected:

Date Received:

Date Tested:

48-h LC50

>100%

The results reported relate only to the sample tested.

SODIUM CHLORIDE REFERENCE TOXICANT DATA

Organism Batch:

Dm12-07

Date Tested (yyyy/mm/dd):

2012-04-17

Historical Mean LC50:

6.4 g/L

LC50 (95% Confidence Limits):

6.9 g/L (6.6 - 7.3)

Warning Limits $(\pm 2SD)$:

5.9 - 7.0 g/L

Statistical Method:

Spearman-Kärber

Analyst(s):

NK

Daphnia magna CULTURE HEALTH DATA

Time to First Brood:

9.2 days

Mean Young Per Brood:

29.7

Culture Mortality:

0% (previous 7 days)

TEST CONDITIONS

Sample Treatment:

None

Number of Replicates:

1

pH Adjustment: Test Aeration:

None None

Test Organisms / Replicate: Total Organisms / Test Level: 10 10

Organism Batch:

Dm12-07

Organism Loading Rate:

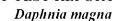
15.0 mL/organism

Test Method Deviation(s):

None

Accredited by the Canadian Association for Laboratory Accreditation Inc. (CALA)





Page 2 of 2

Work Order: Sample Number:

AQUATOX

221256 33959

	Hardness	Hardness	pН	D.O.	Cond.	Temp.	O ₂ Sat.	Total Pre-Aeration
	(mg/L as CaCO ₃)	Adjustment		(mg/L)	(µmhos/cm)	(°C)	(%)*	Time (h) @ 30 mL/min/I
Initial Water Chemistry:	250	None	8.6	9.1	498	21.0	105	0:30
			0 hours					
Date & Time Technician:	2012-04-24 AW	14:50						
Test Conc. (%)	Mortality	Immobility	pН	D.O.	Cond.	Temp.	O ₂ Sat. (%)*	Hardness
100 50	0 0	0	8.6	8.6	498 466	21.0	100	250
25	0	0	8.5 8.5	8.4 8.3	460 460	21.0 21.0		
12.5	0	0	8.5	8.3 8.3	456	21.0		
6.25	0	0	8.5	8.2	454	21.0		
Control	0	0	8.3	8.2	445	21.0	97	200
Notes:								
			24 hours					
Date & Time Technician:	2012-04-25 NK	14:50						
Test Conc. (%)	Mortality	Immobility	pН	D.O.	Cond.	Temp.		
100	0	0	_	_	-	20.5		
50	0	0	****		*****	20.5		
25	0	0	-	_	-	20.5		
12.5	0	0		****		20.5		
6.25	0	0	_	_	_	20.5		
Control	0	0	actes:	prince.	ARR	20.5		
Notes:								
			48 hours	· · · · · · · · · · · · · · · · · · ·				
Date & Time Technician:	2012-04-26 AW	14:50						
Test Conc. (%)	Mortality	Immobility	pН	D.O.	Cond.	Temp.		
100	0	0	8.7	8.3	502	21.0		
50	0	0	8.6	8.5	476	21.0		
25	0	0	8.6	8.5	463	21.0		
12.5	0	0	8.5	8.4	458	21.0		
6.25	0	0	8.5	8.5	452	21.0		
Control	0	0	8.5	8.5	450	21.0		

Notes:

of control organisms showing stress: 0

Daphnia Batch #: Dm12-07

Number immobile does not include number of mortalities.

Test Data Reviewed By: The Date: 2004 Oct 30

⁻ = not measured

^{*} adjusted for actual temp. & barometric pressure



Chronic Bioassay Test Results: Fathead Minnow



11B Nicholas Beaver Rd. RR 3 Guelph ON N1H 6H9

Tel: (519) 763-4412 Fax: (519) 763-4419

Larval Fathead Minnow Test Report

Survival and Growth 1 of 5

2012-04-23

2012-04-24

2012-04-24

15:50

10:30

SAMPLE IDENTIFICATION

Company:

Work Order:

Sample Number:

SGS Lakefield Research Limited

Location:

Lakefield ON

Substance:

PP-22 BZ-MP Treated Process Water-2

Sampling Method:

Grab

221257

33960

Sampled By: Temp. on arrival: B. Bowman 13.0°C

Sample Description:

Clear, colourless, odourless.

Test Method:

Effect

Test of Larval Growth and Survival Using Fathead Minnows. Environment Canada, Conservation and Protection. Ottawa, Ontario. Report EPS 1/RM/22, February 1992 (including Nov. 1997 and

Sept. 2008 amendments).

TEST RESULTS

95% Confidence Limits

Statistical Method

IC25 (Growth from Biomass)

>100%

Value

Date Collected:

Time Collected:

Date Received:

Time Received:

Date Tested:

LC50

>100%

The results reported relate only to the sample tested.

POTASSIUM CHLORIDE REFERENCE TOXICANT DATA

Date Tested:

2012-04-25

Analyst(s):

RD/MR/CL

Organism Batch:

Fm12-04

Test Duration:

7 days

IC25 Growth (from Biomass): 0.99 g/L

LC50:

1.09 g/L

95% Confidence Limits:

0.83 - 1.05 g/L

95% Confidence Limits:

0.87 - 1.36 g/L

Statistical Method:

Non-Linear Regression (CETIS)^a

Statistical Method:

Probit (Stephan)^c

Historical Mean IC25:

0.97 g/L

Historical Mean LC50:

1.12 g/L

Warning Limits (\pm 2SD):

0.82 - 1.14 g/L

Warning Limits (± 2SD):

1.00 - 1.25 g/L

The reference toxicity test was performed under the same experimental conditions as those used with the test sample.

TEST CONDITIONS

Test Organism:

Pimephales promelas

Test Type:

Static Renewal

Organism Batch:

Fm12-04

Control/Dilution Water:

Well water (no chemicals added)

Organism Age:

 \sim 07:00 - 23:45 h at start of test

Test Volume / Replicate:

300 mL

Source:

In-house culture

Test Vessel:

420 mL polystyrene beaker

Culture Mortality/Diseased:

0.07 % (previous 7 days)

Depth of Test Solution:

8 cm

pH Adjustment: Sample Filtration:

None

Organisms per Replicate: Number of Replicates:

Hardness Adjustment:

None None

Daily Renewal Method:

80-85% syphoned and replaced

Test Aeration: None

Test Method Deviation(s): None

10

COMMENTS

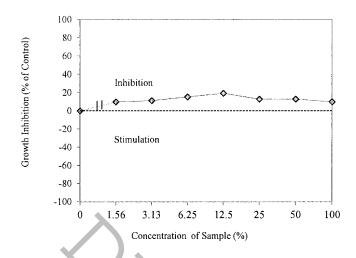
- •All test validity criteria as specified in the test method cited above were satisfied.
- •No organisms exhibiting unusual appearance, behaviour, or undergoing unusual treatment were used in the test.
- •Inflated swim bladders were confirmed in all test organisms used in this test.

Survival and Growth 2 of 5

Work Order:
Sample Number:

221257 33960

Fathead Minnow Growth Inhibition (based on Biomass)



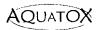
REFERENCES

Date :		Approved By:	
	yyyy-mm-dd		Project Manager

^a CETIS, © 2001-2007. Comprehensive Environmental Toxicity Information System. Tidepool Scientific Software, McKinleyville, Calif. 95519 [Program on disk and printed User's Guide].

^bGrubbs, F.E., 1969. Procedures for detecting outlying observations in samples. *Technometrics*, 11:1-21.

^c Stephan, C. E. 1977. Methods for calculating an LC50. pp 65-84 in: P. L. Mayer and J. L. Hamelink (eds.), Aquatic Toxicology and Hazard Evaluation. Amer. Soc. Testing and Materials, Philadelphia PA. ASTM STP 634.



Survival and Growth

3 of 5

Work Order: Sample Number: 221257 33960

CUMULATIVE DAILY CONTROL MORTALITY AND IMPAIRMENT (±SD)

Date:

2012-04-24

2012-04-25 2012-04-26 2012-04-27

2012-04-28

2012-04-29

2012-04-30

2012-05-01

 $0.00\% \ (\pm 0.0) \quad 0.00\% \ (\pm 0.0)$

FATHEAD MINNOW CUMULATIVE DAILY MORTALITY

Initiation Time:

14:15

Initiation Date: Completion Date: 2012-04-24 2012-05-01

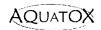
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	Completion E	are .	2012 03																
Marker M			Day	0	Da	y 1	Da	y 2	Da	y 3	Da	y 4			Day	y 6		_	
Marily M																			•
																			, ,
Control B		DE																	
Control B	(%)		^								Λ	0	Λ		^			0.	
C O O O O O O O O O	Control																		0.00 (+0.00)
A	Control						_												0.00 (±0.00)
1.56 B 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0																			
C 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0									-						-			_	0.00 (±0.00)
A 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	1.56									-							-	-	0.00 (±0.00)
3.13 B O O O O O O O O O O O O							-												
C 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0									•			_							0.00 (10.00)
A 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	3.13									U		_							0.00 (±0.00)
6.25 B 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0										· · · · ·									
C 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0			0	0	0	0	0		`			0							
12.5	6.25			0	0	0	0			•	-	0			-			-	0.00 (±0.00)
12.5 B 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0		С										0							,
C 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0		A	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
A 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	12.5	В	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0.00 (±0.00)
25 B 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0		С	0	0	0	0	0	0	0	0	0	0	0	0	0				
C 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0		A	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
A 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	25	В	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	$0.00 (\pm 0.00)$
50 B 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0		С	0	0	0	0	0	0_	0	0	0	0	0	0	0	0	0	0	
C 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0		A	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
A 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	50	В	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	$0.00 (\pm 0.00)$
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		C	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
		A	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
	100	В	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	$0.00~(\pm 0.00)$
			0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	

Aberrant behaviour or swimming impairment:

None

Data Reviewed By:

Date : <u>2012</u> 95-03



Larval Fathead Minnow Test Report

Survival and Growth

4 of 5

Work Order: 221257 Sample Number: 33960

FATHEAD MINNOW DRY WEIGHT AND BIOMASS DATA

Concentration	Replicate	Number of Larvae Exposed	Replicate Mean Dry Weight (mg)	Treatment Mean Biomass (mg)	Standard Deviation	
(%)						
	A	10	0.985			
Control	В	10	0.936	0.934	0.052	
	C	10	0.881			
	Α	10	0.810			
1.56	В	10	0.803	0.842	0.062	
	С	10	0.914			
	A	10	0.879			
3.13	В	10	0.792	0.830	0.045	
	C	10	0.818			
	A	10	0.779			
6.25	В	10	0.790	0.791	0.013	
	С	10	0.805			
	A	10	0.743			
12.5	В	10	0.694	0.755	0.068	
•	C	10	0.828			
	A	10	0.817			
25	В	10	0.782	0.816	0.033	
	C	10	0.848			
	A	10	0.871			
50	В	10	0.783	0.814	0.049	
,	С	10	0.789			
	A	10	0.807			
100	В	10	0.821	0.843	0.050	
	С	10	0.900			

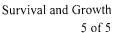
NOTES:

•No outlying data points were detected according to Grubbs Test ^b.

• Control average dry weight per surviving organism = 0.934 mg

Data Reviewed By:

Date: 2012 - 05 - 013



AQUATOX

Fathead Minnow Water Chemistry Data

		Initial Chemistry:		DO (mg/L)	рН	Conductivity (µmhos/cm)		dness s CaCO3)
			25.0	7.7	7.6	421	9	00
		Day 0 - 1	Day 1 - 2	Day 2 - 3	Day 3 - 4	Day 4 - 5	Day 5 - 6	Day 6 - 7
		2012-04-24	2012-04-25	2012-04-26	2012-04-27	2012-04-28	2012-04-29	2012-04-30
Sub-sample Used		1	1	1	2	2	3	3
Temperature (°C)		25.0	24.0	24.0	24.0	24.0	24 .0	24.0
Dissolved Oxygen (mg/L)	7.7	7.2	8.2	7.7	8.1	8.5	8.7
Dissolved Oxygen %	% Sat.¹	100	91	103	95	99	104	108
pН		7.6	7.6	7.6	7.6	7.6	7.7	7.7
Pre-aeration Time	(min)²	0	0	20	0	0	20	20
Analyst(s): Initial	CL	CL	AW	AW	JGG	CL	AW
• •	Final	CL	CL	AW	MR	CL	AW	HL(RD)
Control (0%)								
Temp.(°C)	Initial	24.0	24.0	24.0	24.0	25.0	24.0	24.0
/	Final	24.0	24.0	24.5	24.0	24.0	24.0	24.5
DO % Sat.	Initial	95	89	98	95	97	100	97
DO (mg/L)	Initial	7.6	7.2	7.8	7.8	8.0	8.2	8.1
	Final	7.5	7.1	6.3	6.8	6.4	6.8	6.6
pН	Initial	8.3	8.3	8.4	8.2	8.3	8.4	8.3
	Final	8.2	8.1	8.1	8.0	7.9	8.0	7.7
Cond. (µmhos)	Initial	501	499	555	483	481	476	479
1.56 %								
Temp.(°C)	Initial	24.0	24.0	24.0	24.0	25.0	24.0	24.0
	Final	24.0	24.0	24.5	24.0	24.0	24.0	24.5
DO (mg/L)	Initial	7.6	7.8	7.9	8.0	8.3	8.2	8.1
	Final	7.4	7.0	6.3	6.7	5.7	6.8	6.8
pН	Initial	8.4	8.3	8.4	8.3	8.4	8.4	8.3
	Final	8.2	8.1	8.0	8.0	7.8	8.0	7.8
Cond. (µmhos)	Initial	502	501	485	482	481	477	475
25 %								
Temp.(°C)	Initial	24.0	24.0	24.0	24.0	25.0	24.0	24.0
	Final	24.0	24.0	24.5	24.0	24.0	24.0	24.5
DO (mg/L)	Initial	7.7	7.8	7.9	8.1	8.0	8.3	8.2
	Final	7.6	7.1	6.0	7.1	6.3	7.0	5.9
pН	Initial	8.2	8.2	8.3	8.2	8.3	8.3	8.3
	Final	8.2	8.1	7.9	8.0	7.9	8.0	7.8
Cond. (µmhos)	Initial	485	484	472	471	454	462	460
100 %								
Temp.(°C)	Initial	24.0	24.0	24.0	24.0	25.0	24.0	24.0
•	Final	24.0	24.0	24.5	24.0	24.0	24.0	24.5
DO (mg/L)	Initial	7.6	7.7	8.0	7.9	8.3	8.4	8.4
•	Final	7.3	7.1	5.9	7.1	6.2	7.2	6.4
pН	Initial	7.8	7.9	8.0	7.9	8.0	8.1	8.0
	Final	8.1	8.1	7.9	8.0	7.9	8.1	7.8
Cond. (µmhos)	Initial	425	426	423	427	429	405	405

[&]quot;_" = not measured

¹% saturation (adjusted for actual temperature and barometric pressure)

² \le 100 bubbles/minute



AquaTox Testing & Consulting Inc.

11B Nicholas Beaver Rd. RR 3 Guelph ON N1H 6H9

Tel: (519) 763-4412 Fax: (519) 763-4419

Larval Fathead Minnow Test Report

Survival and Growth 1 of 5

2012-04-23

2012-04-24

2012-04-24

15:50

10:30

Work Order: 221257 Sample Number: 33961

SAMPLE IDENTIFICATION

Company:

SGS Lakefield Research Limited

Location:

Lakefield ON

Substance:

PP-22 BZ-MP Treated Process + Lake Water

Sampling Method: Sampled By:

Grab B. Bowman

Temp. on arrival:

13.0°C

Sample Description: Test Method:

Effect

LC50

Clear, orange, odourless

Test of Larval Growth and Survival Using Fathead Minnows. Environment Canada, Conservation

and Protection. Ottawa, Ontario. Report EPS 1/RM/22, 2nd ed. (February 2011).

TEST RESULTS

95% Confidence Limits

Statistical Method

Date Collected:

Time Collected:

Date Received:

Time Received:

Date Tested:

IC25 (Growth from Biomass)

>100%

Value

>100%

The results reported relate only to the sample tested.

POTASSIUM CHLORIDE REFERENCE TOXICANT DATA

Date Tested: 2012-04-25

Fm12-04

IC25 Growth (from Biomass): 0.99 g/L

95% Confidence Limits:

0.83 - 1.05 g/L

Statistical Method:

Organism Batch:

Non-Linear Regression (CETIS)^a

Historical Mean IC25:

 $0.97 \, g/L$

Warning Limits (\pm 2SD):

0.82 - 1.14 g/L

Statistical Method:

95% Confidence Limits:

Analyst(s):

LC50:

Test Duration:

Historical Mean LC50: 1.12 g/L

Warning Limits (\pm 2SD):

1.00 - 1.25 g/L

RD/MR/CL

7 days

1.09 g/L

0.87 - 1.36 g/L Probit (Stephan)^c

The reference toxicity test was performed under the same experimental conditions as those used with the test sample.

TEST CONDITIONS

Test Organism:

Pimephales promelas

Test Type: Control/Dilution Water: Static Renewal

Organism Batch:

Fm12-04

Test Volume / Replicate:

Well water (no chemicals added)

Organism Age:

~07:00 - 23:30 h at start of test

Source:

In-house culture

Test Vessel:

420 mL polystyrene beaker

Culture Mortality/Diseased:

0.07 % (previous 7 days)

Depth of Test Solution:

8 cm

pH Adjustment:

None

Organisms per Replicate:

10

300 mL

Sample Filtration: Hardness Adjustment: None

Number of Replicates:

80-85% syphoned and replaced

None

Daily Renewal Method:

Test Aeration: None Test Method Deviation(s): None

COMMENTS

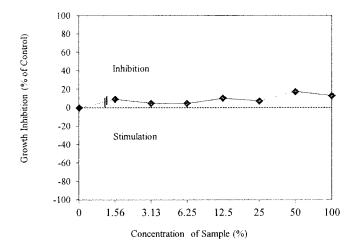
•All test validity criteria as specified in the test method cited above were satisfied.

•No organisms exhibiting unusual appearance, behaviour, or undergoing unusual treatment were used in the test.

•Inflated swim bladders were confirmed in all test organisms used in this test.



Fathead Minnow Growth Inhibition (based on Biomass)



REFERENCES

Date .

vvvv-mm-dd

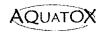
Approved By

Project Manager

^a CETIS, © 2001-2007. Comprehensive Environmental Toxicity Information System. Tidepool Scientific Software, McKinleyville, Calif. 95519 [Program on disk and printed User's Guide].

^b Grubbs, F.E., 1969. Procedures for detecting outlying observations in samples. *Technometrics*, 11:1-21.

^c Stephan, C. E. 1977. Methods for calculating an LC50. pp 65-84 in: P. L. Mayer and J. L. Hamelink (eds.), Aquatic Toxicology and Hazard Evaluation. Amer. Soc. Testing and Materials, Philadelphia PA. ASTM STP 634.



Work Order:

221257

Sample Number: 33961

CUMULATIVE DAILY CONTROL MORTALITY AND IMPAIRMENT (±SD)

Date:

2012-04-24 2012-04-25

2012-04-26

2012-04-27 2012-04-28

2012-04-29

2012-04-30

2012-05-01

 $0.00\% \ (\pm 0.0) \quad 0.00\% \ (\pm 0.0)$

FATHEAD MINNOW CUMULATIVE DAILY MORTALITY

Initiation Time:

14:00

Initiation Date : Completion Date : 2012-04-24 2012-05-01

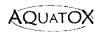
Date : Analyst(s):		Day 2012-0 CN	14-24	Da 2012- X	•	Da 2012- X		2012-	y 3 04-27 C	2012-	y 4 04-28	2012-	y 5 04-29 S	Day 2012-0 A'	04-30	Da 2012- JC	05-01	Treatment Mean Mortality (± SD)
Concentration		Number	· %	Number	%	Number	%	Number	%	Number	%	Number	%	Number	%	Number	%	%
(%)	Replicate	Dead	Dead	Dead	Dead	Dead	Dead	Dead	Dead	Dead	Dead	Dead	Dead	Dead	Dead	Dead	Dead	
	A	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
Control	В	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0.00 (±0.00)
	С	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
	A	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
1.56	В	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0.00 (±0.00)
	С	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
	A	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
3.13	В	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0.00 (±0.00)
	С	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
	A	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
6.25	В	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0.00 (±0.00)
	С	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
	A	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
12.5	В	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0.00 (±0.00)
	C	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
	A	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
25	В	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0.00 (±0.00)
	С	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
	A	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
50	В	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0.00 (±0.00)
	С	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
	Α	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
100	В	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0.00 (±0.00)
	С	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	

Aberrant behaviour or swimming impairment:

None

Data Reviewed By: 500

Date: 2012-05-02



Survival and Growth 4 of 5

Work Order:

221257

Sample Number: 33961

FATHEAD MINNOW DRY WEIGHT AND BIOMASS DATA

Concentration	Replicate	Number of Larvae Exposed	Replicate Mean Dry Weight (mg)	Treatment Mean Biomass (mg)	Standard Deviation
(%)		·			
	A	10	0.918		
Control	В	10	1.021	0.961	0.054
	С	10	0.944		
	A	10	0.889		
1.56	В	10	0.884	0.873	0.024
	С	10	0.846		
	A	10	0.898		
3.13	В	10	0.898	0.916	0.031
	С	10	0.951		
	A	10	0.914		
6.25	В	10	0.947	0.916	0.030
	C	10	0.888		
	Α	10	0.877		·
12.5	В	10	0.913	0.863	0.058
	C	10	0.800		
	A	10	0.871		
25	В	10	0.999	0.891	0.100
	С	10	0.803		
	A	10	0.750		
50	В	10	0.813	0.794	0.039
	С	10	0.820		
	A	10	0.909		
100	В	10	0.842	0.838	0.073
	C	10	0.764		

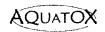
NOTES:

•No outlying data points were detected according to Grubbs Test^b.

• Control average dry weight per surviving organism = 0.961 mg

Data Reviewed By: 5m

Date: 2012-05-02



Fathead Minnow Water Chemistry Data

		Initial Chemistry:	•	DO (mg/L)	рН	Conductivity (µmhos/cm)	(mg/L as	dness s CaCO ₃)
			25.0	8.4	8.5	502	2	50
		Day 0 - 1 2012-04-24	Day 1 - 2 2012-04-25	Day 2 - 3 2012-04-26	Day 3 - 4 2012-04-27	Day 4 - 5 2012-04-28	Day 5 - 6 2012-04-29	Day 6 - 7 2012-04-30
Sub-sample Used		1	1	1	2	2	3	3
Temperature (°C)		25.0	24.0	24.0	24.5	24.0	24.0	24.0
Dissolved Oxygen (8.4	8.0	8.9	8.7	8.9	8.8	8.8
Dissolved Oxygen 9	% Sat.1	110	101	111	108	110	108	108
рH		8.5	8.4	8.4	8.5	8.4	8.4	8.4
Pre-aeration Time	(min) ²	20	20	20	20	20	20	20
Analyst(s): Initial	CL	CL	AW	AW	MR	CL	AW
	Final	CL	AW	CL	MR	CL	AW	HL(RD)
Control (0%)								
Temp.(°C)	Initial	24.0	24.0	24.0	24.0	24.0	24.0	24.0
	Final	24.0	24.0	24.0	24.0	24.0	24.0	24.5
DO % Sat.	Initial	95	98	97	97	97	100	97
DO (mg/L)	Initial	7.5	7.9	7.8	8.0	8.1	8.2	8.0
	Final	7.4	6.8	6.1	6.6	6.3	6.0	6.2
pН	Initial	8.3	8.3	8.3	8.4	8.4	8.3	8.3
	Final	8.2	8.0	7.9	8.1	7.9	7.8	7.7
Cond. (µmhos)	Initial	502	502	485	493	505	476	475
1.56 %								
Temp.(°C)	Initial	24.0	24.0	24.0	24.0	24.0	24.0	24.0
	Final	24.0	24.0	24.0	24.0	24.0	24.0	24.5
DO (mg/L)	Initial	7.6	7.9	7.8	8.0	8.1	8.2	8.1
**	Final	7.1	6.4	6.1	7.1	6.4	6.3	7.2
pН	Initial	8.4	8.3	8.3	8.4	8.4	8.4	8.0
Cond (umhoo)	Final	8.2	7.9	7.9	8.1	7.9 484	7.8 479	7.9
Cond. (µmhos)	Initial	503	504	486	483	404	4/9	477
25 %								
Temp.(°C)	Initial	24.0	24.0	24.0	24.0	24.0	24.0	24.0
	Final	24.0	24.0	24.0	24.0	24.0	24.0	24.5
DO (mg/L)	Initial	7.7	7.9	7.9	8.0	8.2	8.3	8.1
	Final	7.1	6.4	6.2	7.0	6.3	6.4	6.5
pН	Initial	8.4	8.4	8.4	8.4	8.4	8.4	8.4
Cond. (µmhos)	Final Initial	8.3 506	8.0 505	8.0 491	8.2 489	8.0 492	8.0 486	7.9 485
condi (primos)	*********	•	200					
100 %								
Temp.(°C)	Initial	24.0	24.0	24.0	24.0	24.0	24.0	24.0
5 0 / -:	Final	24.0	24.0	24.0	24.0	24.0	24.0	24.5
DO (mg/L)	Initial	8.2	8.0	8.4	8.4	8.3	8.5	8.3
**	Final	7.0	6.4	5.9	6.8	6.7	6.1	5.9
рН	Initial	8.5	8.5	8.5	8.5	8.5	8.5	8.5
Cand (tan)	Final	8.5	8.3	8.3	8.4 506	8.4	8.3	8.2 504
Cond. (µmhos)	Initial	507	507	505	506	505	506	504

[&]quot;_" = not measured

 $^{^{\}rm 1}\,\%$ saturation (adjusted for actual temperature and barometric pressure)

 $^{^2 \}le 100$ bubbles/minute



AquaTox Testing & Consulting Inc.

11B Nicholas Beaver Rd. RR 3 Guelph ON N1H 6H9

Tel: (519) 763-4412 Fax: (519) 763-4419

Larval Fathead Minnow Test Report

Survival and Growth 1 of 5

2012-04-23

2012-04-24

2012-04-25

15:50

10:30

Sample Number: 33962

221257

SAMPLE IDENTIFICATION

Company:

Work Order:

SGS Lakefield Research Limited

Location: Substance: Lakefield ON Lake Water

Sampling Method: Sampled By:

Grab B. Bowman 13.0°C

Temp. on arrival: Sample Description:

Cloudy, orange, odourless

Test Method:

Effect

LC50

Test of Larval Growth and Survival Using Fathead Minnows. Environment Canada, Conservation

and Protection. Ottawa, Ontario. Report EPS 1/RM/22, 2nd ed. (February 2011).

TEST RESULTS 95% Confidence Limits

Statistical Method

Date Collected:

Time Collected:

Date Received:

Time Received:

Date Tested:

IC25 (Growth from Biomass)

>100%

Value

>100%

The results reported relate only to the sample tested.

POTASSIUM CHLORIDE REFERENCE TOXICANT DATA

Date Tested:

2012-04-25

Analyst(s):

RD/MR/CL

Organism Batch:

Fm12-04

Test Duration:

7 days

IC25 Growth (from Biomass): 0.99 g/L

LC50:

1.09 g/L

95% Confidence Limits:

0.83 - 1.05 g/L

95% Confidence Limits:

0.87 - 1.36 g/L

Statistical Method:

Non-Linear Regression (CETIS)^a

Statistical Method:

Probit (Stephan)^c

Historical Mean IC25:

0.97 g/L

Historical Mean LC50:

1.12 g/L

Warning Limits (\pm 2SD):

0.82 - 1.14 g/L

Warning Limits (\pm 2SD):

1.00 - 1.25 g/L

The reference toxicity test was performed under the same experimental conditions as those used with the test sample.

TEST CONDITIONS

Test Organism:

Pimephales promelas

Test Type:

Static Renewal

Organism Batch:

Fm12-04

Control/Dilution Water:

Well water (no chemicals added)

Organism Age:

 $\sim 07:00 - 22:05$ h at start of test

Test Volume / Replicate:

300 mL

Source:

In-house culture

Test Vessel:

420 mL polystyrene beaker

Culture Mortality/Diseased:

0.07 % (previous 7 days)

Depth of Test Solution:

8 cm

pH Adjustment:

None

Organisms per Replicate:

3

Sample Filtration:

None

Number of Replicates: Daily Renewal Method:

80-85% syphoned and replaced

Hardness Adjustment:

None

Test Aeration: None Test Method Deviation(s): None

10

COMMENTS

- •All test validity criteria as specified in the test method cited above were satisfied.
- •No organisms exhibiting unusual appearance, behaviour, or undergoing unusual treatment were used in the test.
- •Inflated swim bladders were confirmed in all test organisms used in this test.



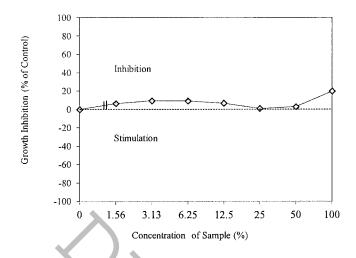
Work Order:

221257

Sample Number:

33962

Fathead Minnow Growth Inhibition (based on Biomass)



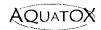
REFERENCES

Date :		Approved By:	
	yyyy-mm-dd		Project Manager

^a CETIS, © 2001-2007. Comprehensive Environmental Toxicity Information System. Tidepool Scientific Software, McKinleyville, Calif. 95519 [Program on disk and printed User's Guide].

^bGrubbs, F.E., 1969. Procedures for detecting outlying observations in samples. *Technometrics*, 11:1-21.

^c Stephan, C. E. 1977. Methods for calculating an LC50. pp 65-84 in: P. L. Mayer and J. L. Hamelink (eds.), Aquatic Toxicology and Hazard Evaluation. Amer. Soc. Testing and Materials, Philadelphia PA. ASTM STP 634.



Work Order: Sample Number: 221257 33962

CUMULATIVE DAILY CONTROL MORTALITY AND IMPAIRMENT (±SD)

Date:

2012-04-25

2012-04-26 2012-04-27 2012-04-28 2012-04-29 2012-04-30

2012-05-01

2012-05-02

 $0.00\% \ (\pm 0.0) \quad 0.00\% \ (\pm 0.0)$

FATHEAD MINNOW CUMULATIVE DAILY MORTALITY

Initiation Time:

14:05

Initiation Date: Completion Date: 2012-04-25

2012-05-02

Date : Analyst(s):		Day 2012-0 VC	4-25	Day 2012-0 V	04-26	Da; 2012-	04-27	Day 2012-	04-28	Da 2012- F	04-29	Da 2012- X	04-30	Da 2012- A	05-01	Day 2012-0 RJ	05-02	Treatment Mean Mortality (± SD)
Concentration		Number	%	Number	%	Number	%	Number	%	Number Dead	9/0	Number Dead	% Dead	Number Dead	% Dead	Number Dead	% Dead	%
(%)	Replicate	Dead	Dead	Dead	Dead	Dead	Dead	Dead	Dead	Dead	Dead	Dead	Dead	Dead	Dead	Dead	Dead	
	A	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
Control	В	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	$0.00 (\pm 0.00)$
	C	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
-	A	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
1.56	В	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	$0.00 (\pm 0.00)$
	C	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
	A	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
3.13	В	0	0	0	0	0	0	0	0	0	0	0	0	0	0	1	10	3.33 (±5.77)
	С	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
	A	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
6.25	В	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0.00 (±0.00)
	С	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
	A	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
12.5	В	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0.00 (±0.00)
	С	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
	A	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
25	В	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0.00 (±0.00)
~~	C	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	,
	A	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
50	В	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0.00 (±0.00)
20	C	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
Section 1981	A	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
100	В	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0.00 (±0.00)
200	C	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	(")

Aberrant behaviour or swimming impairment:

None

Data Reviewed By:

Date: 2012-05-03

Survival and Growth 4 of 5

Work Order: Sample Number: 33962

221257

FATHEAD MINNOW DRY WEIGHT AND BIOMASS DATA

Concentration	Replicate	Number of Larvae Exposed	Replicate Mean Dry Weight (mg)	Treatment Mean Biomass (mg)	Standard Deviation
(%)					
	Α	10	0.976		
Control	В	10	0.997	1.004	0.032
	С	10	1.039		
	Α	10	0.933		
1.56	В	10	0.987	0.939	0.045
	С	10	0.898		
	Α	10	0.893		
3.13	В	10	0.855	0.908	0.062
	C	10	0.976		
	A	10	0.8851		
6.25	В	10	0.924	0.911	0.023
	С	10	0.925		
	A	10	0.966		
12.5	В	10	0.969	0.936	0.055
	С	10	0.872^{1}		
	A	10	0.996		
25	В	10	1.036	0.994	0.044
	С	10	0.949		
	A	10	0.858		
50	В	10	0.861	0.974	0.199
	C	10	1.204		
	A	10	0.674		*****
100	В	10	0.898	0.805	0.117
	C	10	0.844		

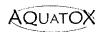
NOTES:

Data Reviewed By:

Date: 2012-05-03

^{• &}lt;sup>1</sup>Outlier according to Grubbs Test^b. Outlying data points were not excluded from statistical analysis, since they could not be attributed to error.

[•] Control average dry weight per surviving organism = 1.004 mg



Fathead Minnow Water Chemistry Data

		Initial Chemistry:		DO (mg/L)	pH	Conductivity (µmhos/cm)	(mg/L as	dness s CaCO ₃)
			24.0	8.1	8.5	500	2	60
		Day 0 - 1 2012-04-25	Day 1 - 2 2012-04-26	Day 2 - 3 2012-04-27	Day 3 - 4 2012-04-28	Day 4 - 5 2012-04-29	Day 5 - 6 2012-04-30	Day 6 - 7 2012-05-01
Sub-sample Used		1	1	1	2	2	3	3
Temperature (°C)		24.0	24.0	24.5	24.0	24.0	24.0	24.5
Dissolved Oxygen (8.1	8.6	8.8	8.8	8.9	9.0	8.9
Dissolved Oxygen	% Sat.1	102	108	110	108	109	110	112
рН		8.5	8.4	8.4	8.4	8.4	8.4	8.4
Pre-aeration Time	(min) ²	20	20	20	20	20	20	20
Analyst(s): Initial	CL	AW	AW	JGG	CL	AW	AW
```	Final	CL	CL	JGG	CL	AW	HL(RD)	HL(RD)
Control (0%)								
Temp.(°C)	Initial	24.0	24.0	24.0	25.0	24.0	24.0	24.5
/	Final	24.0	24.0	24.5	24.0	24.0	24.5	25.0
DO % Sat.	Initial	97	97	97	96	100	98	97
DO (mg/L)	Initial	7.8	7.7	7.9	8.0	8.2	8.0	7.7
	Final	7.5	6.3	7.2	6.7	6.6	6.2	5.6
pН	Initial	8.4	8.4	8.4	8.3	8.4	8.4	8.3
	Final	8.2	7.9	8.1	8.1	8.1	8.1	7.6
Cond. (µmhos)	Initial	500	484	482	478	476	477	474
1.56 %				7				
Temp.(°C)	Initial	24.0	24.0	24.0	25.0	24.0	24.0	24.5
	Final	24.0	24.0	24.5	24.0	24.0	24.5	25.0
DO (mg/L)	Initial	7.9	7.8	7.9	8.0	8.2	8.1	8.3
	Final	7.6	6.4	7.3	6.8	6.3	6.2	5.4
pН	Initial	8.3	8.4	8.4	8.4	8.4	8.3	8.3
	Final	8.2	7.9	8.1	8.0	7.9	8.0	7.7
Cond. (µmhos)	Initial	503	484	483	482	479	479	477
25 %								
Temp.(°C)	Initial	24.0	24.0	24.0	25.0	24.0	24.0	24.5
•	Final	24.0	24.0	24.5	24.0	24.0	24.5	25.0
DO (mg/L)	Initial	7.9	7.9	8.0	8.0	8.3	8.1	8.0
	Final	7.3	6.4	7.1	6.5	6.2	7.3	5.2
pН	Initial	8.3	8.4	8.4	8.5	8.5	8.3	8.4
	Final	8.3	8.1	8.2	8.1	8.0	8.0	7.8
Cond. (µmhos)	Initial	502	491	489	489	487	490	485
100 %								
Temp.(°C)	Initial	24.0	24.0	24.0	25.0	24.0	24.0	24.5
- • •	Final	24.0	24.0	24.5	24.0	24.0	24.5	25.0
DO (mg/L)	Initial	8.1	8.3	8.3	8.4	8.5	8.2	8.6
	Final	7.2	6.7	7.0	6.3	5.9	7.6	5.7
pН	Initial	8.5	8.5	8.5	8.5	8.5	8.5	8.5
	Final	8.5	8.4	8.4	8.4	8.3	8.3	8.1
Cond. (µmhos)	Initial	506	504	505	<b>5</b> 06	506	507	506

[&]quot;-" = not measured

¹% saturation (adjusted for actual temperature and barometric pressure)

² ≤100 bubbles/minute



# Chronic Bioassay Test Results: Ceriodaphnia (crustacean)



AquaTox Testing & Consulting Inc.

11B Nicholas Beaver Rd. RR 3

Guelph ON N1H 6H9

Tel: (519) 763-4412 Fax: (519) 763-4419

Ceriodaphnia dubia Test Report

2012-04-23

2012-04-24

2012-04-24

15:50

10:30

Survival and Reproduction 1 of 4

SAMPLE IDENTIFICATION

Company:

Work Order:

Sample Number:

SGS Lakefield Research Limited

Location:

Lakefield ON

Substance:

PP-22 BZ-MP Treated Process Water-2

Sampling Method: Sampled By:

Grab B. Bowman

221257

33960

Temp. on arrival:

13.0°C

Sample Description:

Clear, colourless, odourless.

Test Method:

Test of Reproduction and Survival using the Cladoceran Ceriodaphnia dubia. Environment Canada,

Date Collected:

Time Collected:

Date Received:

Time Received:

Date Tested:

Conservation and Protection. Ottawa, Ontario. Report EPS 1/RM/21, 2nd ed. (February 2007).

TEST RESULTS

Effect Value 95% Confidence Limits

Statistical Method

LC50

>100%

63.5% IC25 (Reproduction)

56.9-68.0

Linear Interpolation (CETIS) a

The results reported relate only to the sample tested.

SODIUM CHLORIDE REFERENCE TOXICANT DATA

Date Tested:

2012-04-25

Analyst(s): Test Duration: VC/MR/RD 7 days

Organism Batch:

Cd12-04

2.09 g/L

IC25 Reproduction:

0.86 g/L

LC50:

95% Confidence Limits:

0.76 - 1.11 g/L

95% Confidence Limits: Statistical Method:

0.65 - 3.00 g/L

Statistical Method:

Linear Interpolation (CETIS)^a

Historical Mean LC50:

Nonlinear Interpolation (Stephan)^c

Historical Mean IC25:

1.04 g/L

2.17 g/L

Warning Limits ( $\pm$  2SD):

0.60 - 1.81 g/L

Warning Limits ( $\pm$  2SD):

1.45 - 3.24 g/L

The reference toxicity test was performed under the same experimental conditions as those used with the test sample.

#### TEST CONDITIONS

Sample Filtration:

None

Test Volume per Replicate: 15 mL

Test Aeration:

None

Depth of Test Solution:

22 mL polystyrene vial

pH Adjustment:

None

Organisms per Replicate:

4.0 cm

Hardness Adjustment:

None

10

Daily Renewal Method: Control/Dilution Water: Transferred to fresh solutions Well water (no chemicals added)

Number of Replicates: Test Method Deviation(s): None

Test Vessel:

#### **COMMENTS**

•All test validity criteria as specified in the test method cited above were satisfied.

•Statistical analysis could not be performed using non linear regression, since a suitable model could not be found. Therefore, test results were calculated using Linear Interpolation (CETIS)^a.

### Ceriodaphnia dubia Test Report



Survival and Reproduction

2 of 4

Work Order:

221257

Sample Number:

33960

TEST	$\Omega \mathbf{D}$	CA	NI	CN	A C
1651	UK	A	IVI		115

Test Organism: Ceriodaphnia dubia

Organism Batch:

Cd12-04

Organism Origin:

Single in-house mass culture

Test Organism Origin: Individual in-house cultures

Range of Age (at start of test):

06:00 h - 12:10 h

Mean Brood Organism Mortality:

3.3%

Ephippia in Culture:

No

**Brood Organism Neonate Production** 

Replicate:	1	2
Total (third or subsequent brood):	18	16
Total (first three broods):	27	24

3 5 15 17 15 26 28 21

6 14 17 23 25

9 13 27

10 Mean 14 15.3 25.3 25

No organisms exhibiting unusual appearance, behaviour, or undergoing unusual treatment were used in the test.

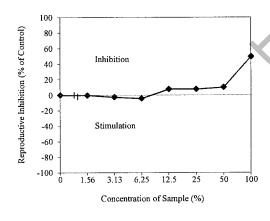
#### **TEST DATA**

#### Ceriodaphnia dubia Reproductive Inhibition

#### Cumulative Daily Test Organism Mortality (%)

14

27



				Concen	tration (	of Samp	le (%)		
Date	Test Day	Control	1.56	3.13	6.25	12.5	25	50	100
2012-04-25	1	0	0	0	0	0	0	0	0
2012-04-26	2	0	0	0	0	0	0	0	0
2012-04-27	3	0	0	0	0	0	0	0	0
2012-04-28	4	0	0	0	0	0	0	0	0
2012-04-29	5	0	0	0	0	0	0	0	0
2012-04-30	6	0	0	0	0	0	0	0	0
2012-05-01	7	0	0	0	0	0	0	0	0
Total Mo	tality (%)	0	0	0	0	0	0	0	0

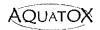
#### REFERENCES

Date :		Approved By:		
	yyyy-mm-dd		Project Manager	

^a CETIS, © 2001-2007. Comprehensive Environmental Toxicity Information System. Tidepool Scientific Software, McKinleyville, Calif. 95519[Program on disk and printed User's Guide].

^c Stephan, C. E. 1977. Methods for calculating an LC50. pp 65-84 in: P. L. Mayer and J. L. Hamelink (eds.), Aquatic Toxicology and Hazard Evaluation. Amer. Soc. Testing and Materials, Philadelphia PA. ASTM STP 634.

3 of 4



Work Order: 221257 Sample Number: 33960

#### Ceriodaphnia dubia Survival and Reproduction

Test Initiation Date : 2012-04-24 Initiation Time : 15:30 Test Completion Date : 2012-05-01

Concentration (	%)					Rep	olicate	!				Mean Young	Analyst(	s) Concentratio	n (%)					Rep	olicate					Mean Young
Control	Day	1	2	3	4	5	6	7	8	9	10	(±SD)		12.5	Day	1	2	3	4	5	6	7	8	9	10	(±SD)
2012-04-25	I	0	0	0	0	0	0	0	0	0	0	0	CL	2012-04-25	1	0	0	0	0	0	0	0	0	0	0	0
2012-04-26	2	0	0	0	0	0	0	0	0	0	0	0	CL	2012-04-26	2	0	0	0	0	0	0	0	0	0	0	0
2012-04-27	3	0	0	0	0	0	0	0	0	0	0	0	CL	2012-04-27	3	0	0	0	0	0	0	0	0	0	0	0
2012-04-28	4	7	7	6	6	6	6	6	6	7	6	6.3	VC	2012-04-28	4	6	6	5	6	7	8	6	3	6	6	5.9
2012-04-29	5	11	13	13	14	13	15	17	14	11	14	13.5	VC	2012-04-29	5	13	11	13	13	15	13	11	6	13	12	12
2012-04-30	6	0	14	0	0	11	0	9	0	0	20	5.4	XD	2012-04-30	6	17	1	12	0	0	0	0	0	0	15	4.5
2012-05-01	7	20	0	17	19	0	19	0	22	18	0	11.5	RD	2012-05-01	7	0	17	0	17	21	18	15	11	16	0	11.5
Total		38	34	36	39	30	40	32	42	36	40	36.7 (±3.8	)	Total		36	35	30	36	43	39	32	20	35	33	33.9 (±6.1)

Concentration (	(%)					Re	plicate	,				Mean Young	Concentration	n (%)					Re	plicate				·	Mean Young
1.56	Day	1	2	3	4	5	6	7	8	9	10	(±SD)	25	Day	1	2	3	4	5	6	7	8	9	10	(±SD)
2012-04-25	1	0	0	0	0	0	0	0	0	0	0	0	2012-04-25	1	0	0	0	0	0	0	0	0	0	0	0
2012-04-26	2	0	0	0	0	0	0	0	0	0	0	0	2012-04-26	2	0	0	0	0	0	0	0	0	0	0	0
2012-04-27	3	0	0	0	0	0	0	0	0	0	0	0	2012-04-27	3	0	0	0	0	0	0	0	0	0	0	0
2012-04-28	4	6	5	7	8	6	7	8	8	4	6	6.5	2012-04-28	4	7	4	3	7	6	7	7	6	5	7	5.9
2012-04-29	5	12	13	15	16	14	15	10	17	13	16	14.1	2012-04-29	5	12	10	15	12	12	12	15	13	14	14	12.9
2012-04-30	6	4	0	18	0	0	0	0	0	0	19	4.1	2012-04-30	6	17	0	16	15	15	0	17	0	5	19	10.4
2012-05-01	7	0	18	0	16	18	19	21	17	12	0	12.1	2012-05-01	7	0	14	0	0	0	18	0	15	0	0	4.7
Total		22	36	40	40	38	41	39	42	29	41	36.8 (±6.4)	Total		36	28	34	34	33	37	<b>3</b> 9	34	24	40	33.9 (±4.8)

Concentration (	(%)					Re	plicate	;				Mean Young	7	Concentration	ı (%)					Rep	olicate					Mean Young
3.13	Day	1	2	3	4	5	6	7	8	9	10	(±SD)	7	50	Day	1	2	3	4	5	6	7	8	9	10	(±SD)
2012-04-25	1	0	0	0	0	0	0	0	0	0	0	0		2012-04-25	1	0	0	0	0	0	0	0	0	0	0	0
2012-04-26	2	0	0	0	0	0	0	0	0	0	0	0		2012-04-26	2	0	0	0	0	0	0	0	0	0	0	0
2012-04-27	3	0	0	0	0	0	0	0	0	0	0	0		2012-04-27	3	0	0	0	0	0	0	0	0	0	0	0
2012-04-28	4	6	6	7	6	7	8	7	6	5	6	6.4		2012-04-28	4	6	6	7	7	7	6	6	5	6	8	6.4
2012-04-29	5	15	13	15	14	15	15	16	9	15	15	14.2		2012-04-29	5	11	11	13	14	12	11	12	9	11	11	11.5
2012-04-30	6	14	0	7	0	0	0	0	0	0	26	4.7		2012-04-30	6	16	12	19	0	0	15	13	0	0	11	8.6
2012-05-01	7	0	19	0	19	24	20	7	13	22	0	12.4		2012-05-01	7	0	0	0	17	15	0	0	14	19	0	6.5
Total		35	38	29	39	46	43	30	28	42	47	37.7 (±7.0)		Total		33	29	39	38	34	32	31	28	36	30	33.0 (±3.7)

Concentration (	%)					Re	olicate	;				Mean Young	Concentration	1 (%)					Rej	olicate					Mean Young
6.25	Day	1	2	3	4	5	6	7	8	9	10	(±SD)	100	Day	1	2	3	4	5	6	7	8	9	10	(±SD)
2012-04-25	1	0	0	0	0	0	0	0	0	0	0	0	2012-04-25	1	0	0	0	0	0	0	0	0	0	0	0
2012-04-26	2	0	0	0	0	0	0	0	0	0	0	0	2012-04-26	2	0	0	0	0	0	0	0	0	0	0	0
2012-04-27	3	0	0	0	0	0	0	0	0	0	0	0	2012-04-27	3	0	0	0	0	0	0	0	0	0	0	0
2012-04-28	4	8	6	7	6	6	5	9	3	6	8	6.4	2012-04-28	4	4	0	0	3	0	4	5	0	0	1	1.7
2012-04-29	5	15	14	15	12	14	15	15	15	12	14	14.1	2012-04-29	5	6	0	4	0	0	6	9	7	0	4	3.6
2012-04-30	6	19	18	0	0	0	0	0	0	18	16	7.1	2012-04-30	6	0	6	0	5	5	0	0	0	6	0	2.2
2012-05-01	7	0	0	17	15	20	19	18	18	0	0	10.7	2012-05-01	7	11	10	10	11	9	11	10	13	10	15	11
Total		42	38	<b>3</b> 9	33	40	39	42	36	36	38	38.3 (±2.8)	Total		21	16	14	19	14	21	24	20	16	20	18.5 (±3.3)

NOTES: *All young produced by a test organism during its fourth and subsequent broods were discarded and not included in the above counts. The presence of two or more neonates in any test chamber, during any given day of the test, constitutes a brood.

Data Reviewed By:

Date: 2012-08-08

[•]No outlying data points were detected according to Grubbs Test (CETIS)^a.

[&]quot;x"= test organism mortality

[&]quot;*"= accidental test organism mortality

[&]quot;-"=4th brood (see 'NOTES')



### Ceriodaphnia dubia Water Chemistry Data

		Initial Chemistry:	Temp. (°C)	DO (mg/L) 7.7	<b>рН</b> 7.6	Conductivity (µmhos/cm) 421	Hardness (mg/L as CaCO ₃ )	
			2010		,			
Date :		Day 0 - 1 2012-04-24	Day 1 - 2 2012-04-25	Day 2 - 3 2012-04-26	Day 3 - 4 2012-04-27	Day 4 - 5 2012-04-28	Day 5 - 6 2012-04-29	Day 6 - 7 2012-04-30
Sub-sample Use	d	1	1	1	2	2	3	3
Temperature (°		25.0	24.0	24.0	24.0	24.0	24.0	24.0
Dissolved Oxyg	*	7.7	7.2	8.2	7.7	8.1	8.5	8.7
Dissolved Oxyg		100	91	103	95	99	104	108
рН		7.6	7.6	7.6	7.6	7.6	7.7	7.7
Pre-aeration Ti	me (min) ⁴	0	0	20	0	0	20	20
Analyst(s)	Initial	$\operatorname{CL}$	CL	AW	AW	JGG	CL	AW
	Final	CL	CL	CL	MR	CL	AW	HL(RD)
Control (0%)	١							
Temp. (°C)	, Initial	24.0	24.0	24.0	24.0	25.0	24.0	24.0
	Final	24.0	24.0	24.0	24.0	24.0	24.0	24.5
DO % Sat.3	Initial	95	89	98	95	97	100	97
DO (mg/L)	Initial	7.6	7.2	7.8	7.8	8.0	8.2	8.1
,	Final	7.2	7.0	7.2	7.3	7.4	7.5	7.2
pН	Initial	8.3	8.3	8.4	8.2	8.3	8.4	8.3
	Final	8.1	8.0	8.1	8.1	8.2	8.3	7.8
Cond. (µmhos)	Initial	501	499	555	483	481	476	479
1.56 %				<b>*</b> ./				
Temp. (°C)	Initial	24.0	24.0	24.0	24.0	25.0	24.0	24.0
	Final	24.0	24.0	24.0	24.0	24.0	24.0	24.5
DO (mg/L)	Initial	7.6	7.8	7.9	8.0	8.3	8.2	8.1
	Final	7.0	7.1	7.1	7.3	7.6	7.5	7.1
pН	Initial	8.4	8.3	8.4	8.3	8.4 8.2	8.4 8.2	8.3 8.0
Cond. (µmhos)	Final Initial	8.1 502	8.0 501	8.1 485	8.2 482	8.2 481	6.2 477	475
Cond. (µminos)	Illitiai	302	301	403	402	401	7//	4/3
25 %								
Temp. (°C)	Initial	24.0	24.0	24.0	24.0	25.0	24.0	24.0
DO ( // // )	Final	24.0	24.0	24.0	24.0	24.0	24.0	24.5
DO (mg/L)	Initial Final	7.7 7.1	7.8 7.1	7.9 7.3	8.1 7.3	8.0 7.6	8.3 7.6	8.2 7.3
pН	Initial	8.2	8.2	8.3	8.2	8.3	8.3	8.3
pri	Final	8.1	8.1	8.1	8.2	8.2	8.2	8.0
Cond. (µmhos)	Initial	485	484	472	471	454	462	460
100.0/								
100 %	Initial	24.0	24.0	24.0	24.0	25.0	24.0	24.0
Temp. (°C)	Initial Final	24.0	24.0	24.0	24.0	24.0	24.0	24.5
DO (mg/L)	Initial	7.6	7.7	8.0	7.9	8.3	8.4	8.4
= = (	Final	7.2	7.2	7.3	7.4	7.4	7.5	7.3
pН	Initial	7.8	7.9	8.0	7.9	8.0	8.1	8.0
-	Final	8.2	8.1	8.2	8.2	8.2	8.2	8.1
Cond. (µmhos)	Initial	425	426	423	427	429	405	405

[&]quot;-" = not measured

 $^{^3\ \%}$  saturation (adjusted for actual temperature and barometric pressure)

⁴ ≤100 bubbles/minute



AquaTox Testing & Consulting Inc.

11B Nicholas Beaver Rd. RR 3

Guelph ON N1H 6H9

Tel: (519) 763-4412 Fax: (519) 763-4419

Ceriodaphnia dubia Test Report Survival and Reproduction

2012-04-23

2012-04-24

2012-04-24

15:50

10:30

1 of 4

SAMPLE IDENTIFICATION

Company:

Work Order: Sample Number:

SGS Lakefield Research Limited

Location:

Lakefield ON

221257

33961

Substance:

PP-22 BZ-MP Treated Process + Lake Water

Sampling Method: Grab

B. Bowman Sampled By: 13.0°C Temp. on arrival:

Sample Description:

Clear, orange, odourless

Test Method:

Test of Reproduction and Survival using the Cladoceran Ceriodaphnia dubia. Environment Canada,

Date Collected:

Time Collected:

Date Received:

Time Received:

Date Tested:

Conservation and Protection. Ottawa, Ontario. Report EPS 1/RM/21, 2nd ed. (February 2007).

TEST RESULTS

Effect Value

95% Confidence Limits

Statistical Method

LC50

>100%

74.8% IC25 (Reproduction)

65.2-84.7

Linear Interpolation (CETIS) a

The results reported relate only to the sample tested.

SODIUM CHLORIDE REFERENCE TOXICANT DATA

Date Tested:

2012-04-25 Organism Batch:

Cd12-04

Analyst(s): Test Duration: VC/MR/RD 7 days

IC25 Reproduction:

0.86 g/L

LC50:

95% Confidence Limits:

0.76 - 1.11 g/L

95% Confidence Limits:

2.09 g/L 0.65 - 3.00 g/L

Statistical Method:

Linear Interpolation (CETIS)^a

Statistical Method:

Nonlinear Interpolation (Stephan)^c

Historical Mean IC25:

1.04 g/L

Historical Mean LC50:

Warning Limits ( $\pm 2SD$ ):

0.60 - 1.81 g/L

Warning Limits ( $\pm$  2SD):

2.17 g/L 1.45 - 3.24 g/L

The reference toxicity test was performed under the same experimental conditions as those used with the test sample.

#### **TEST CONDITIONS**

Sample Filtration:

None

Test Volume per Replicate: 15 mL Test Vessel:

Test Aeration:

None None

Depth of Test Solution:

22 mL polystyrene vial 4.0 cm

pH Adjustment: Hardness Adjustment:

None

Organisms per Replicate:

1 10

Daily Renewal Method: Control/Dilution Water:

Transferred to fresh solutions

Number of Replicates:

Well water (no chemicals added) Test Method Deviation(s): None

#### **COMMENTS**

•All test validity criteria as specified in the test method cited above were satisfied.

•Statistical analysis could not be performed using non linear regression, since a suitable model could not be found. Therefore, test results were calculated using Linear Interpolation (CETIS)^a.





Work Order:

Test Organism:

Organism Batch:

Organism Origin:

221257

Cd12-04

Test Organism Origin: Individual in-house cultures

Ceriodaphnia dubia

Single in-house mass culture

Sample Number: 33961

**TEST ORGANISMS** 

Range of Age (at start of test):

12:15 h - 22:50 h

Mean Brood Organism Mortality:

3.3%

Ephippia in Culture:

No

**Brood Organism Neonate Production** 

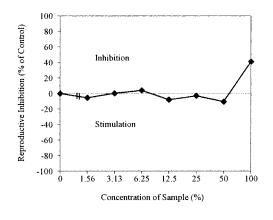
Replicate:	1	2	3	4	5	6	7	8	9	10	Mean
Total (third or subsequent brood):	18	15	17	17	20	16	18	17	19	17	17.4
Total (first three broods):	25	24	26	28	27	25	22	26	26	19	24.8

No organisms exhibiting unusual appearance, behaviour, or undergoing unusual treatment were used in the test.

#### TEST DATA

#### Ceriodaphnia dubia Reproductive Inhibition

#### Cumulative Daily Test Organism Mortality (%)



				Concer	itration	of Samp	le (%)		
Date	Test Day	Control	1.56	3.13	6.25	12.5	25	50	100
2012-04-25	1	0	0	0	0	0	0	0	0
2012-04-26	2	0	0	0	0	0	0	0	0
2012-04-27	3	0	0	0	0	0	0	0	0
2012-04-28	4	0	0	0	0	0	0	0	0
2012-04-29	5	0	0	10	10	0	10	0	0
2012-04-30	6	0	0	10	10	0	10	0	0
Total Mor	tality (%)	0	0	10	10	0	10	0	0

#### REFERENCES

Approved By:

^a CETIS, © 2001-2007. Comprehensive Environmental Toxicity Information System. Tidepool Scientific Software, McKinleyville, Calif. 95519[Program on disk and printed User's Guide].

^c Stephan, C. E. 1977. Methods for calculating an LC50. pp 65-84 in: P. L. Mayer and J. L. Hamelink (eds.), Aquatic Toxicology and Hazard Evaluation. Amer. Soc. Testing and Materials, Philadelphia PA. ASTM STP 634.



Work Order: Sample Number: 221257 33961

Ceriodaphnia dubia Survival and Reproduction

Test Initiation Date:

2012-04-24 15:35

Initiation Time : Test Completion Date :

2012-04-30

Concentration (	%)					Re	plicate	:				Mean Young	Analyst(	s) Concentration	1 (%)					Rep	olicate					Mean Young
Control	Day	1	2	3	4	5	6	7	8	9	10	(±SD)		12.5	Day	1	2	3	4	5	6	7	8	9	10	(±SD)
2012-04-25	1	0	0	0	0	0	0	0	0	0	0	0	XD	2012-04-25	1	0	0	0	0	0	0	0	0	0	0	0
2012-04-26	2	0	0	0	0	0	0	0	0	0	0	0	XD	2012-04-26	2	0	0	0	0	0	0	0	0	0	0	0
2012-04-27	3	0	0	0	0	0	0	0	0	0	0	0	RD	2012-04-27	3	0	0	0	0	0	0	0	0	0	0	0
2012-04-28	4	7	6	6	6	6	6	6	6	5	6	6.0	MR	2012-04-28	4	7	6	6	6	5	6	6	7	6	6	6.1
2012-04-29	5	11	13	12	9	12	15	11	12	12	14	12.1	CL	2012-04-29	5	15	13	12	12	8	12	12	14	11	14	12.3
2012-04-30	6	21	19	14	22	17	18	11	0	16	25	16.3	XD	2012-04-30	6	21	17	19	20	17	19	20	19	18	17	18.7
Total		39	38	32	37	35	39	28	18	33	45	34.4 (±7.4)	)	Total		43	36	37	38	30	37	38	40	35	37	37.1 (±3.3)

Concentration (	%)					Rej	plicate	,				Mean Young	Concentration	1 (%)					Rep	olicate	:				Mean Young
1.56	Day	1	2	3	4	5	6	7	8	9	10	(±SD)	25	Day	1	2	3	4	5	6	7	8	9	10	(±SD)
2012-04-25	1	0	0	0	0	0	0	0	0	0	0	0	2012-04-25	1	0	0	0	0	0	0	0	0	0	0	0
2012-04-26	2	0	0	0	0	0	0	0	0	0	0	0	2012-04-26	2	0	0	0	0	0	0	0	0	0	0	0
2012-04-27	3	0	0	0	0	0	0	0	0	0	0	0	2012-04-27	3	0	0	0	0	0	0	0	0	0	0	0
2012-04-28	4	7	7	6	7	6	6	5	6	7	6	6.3	2012-04-28	4	7	6	6	6	7	6	6	5	5	6	6.0
2012-04-29	5	11	12	10	13	11	11	10	8	12	13	11.1	2012-04-29	5	14 >	c 13	11	10	12	12	15	11	10	12	12.0
2012-04-30	6	23	22	21	18	18	23	21	14	17	12	18.9	2012-04-30	6	0	21	17	18	19	19	22	19	19	20	17.4
Total		41	41	37	38	35	40	36	28	36	31	36.3 (±4.2)	Total		21	40	34	34	38	37	43	35	34	38	35.4 (±5.9)

Concentration (	%)					Re	plicate					Mean Young	Concentration	n (%)					Rep	olicate					Mean Young
3.13	Day	1	2	3	4	5	6	7	8	9	10	(±SD)	50	Day	1	2	3	4	5	6	7	8	9	10	(±SD)
2012-04-25	1	0	0	0	0	0	0	0	0	0	0	0	2012-04-25	1	0	0	0	0	0	0	0	0	0	0	0
2012-04-26	2	0	0	0	0	0	0	0	0	0	0	0	2012-04-26	2	0	0	0	0	0	0	0	0	0	0	0
2012-04-27	3	0	0	0	0	0	0	0	0	0	0	0	2012-04-27	3	0	0	0	0	0	0	0	0	0	0	0
2012-04-28	4	7	6	6	5	6	7	6	4	6	6	5.9	2012-04-28	4	7	7	6	7	6	7	6	6	5	4	6.1
2012-04-29	5	15	13	12	12	13	14	12	11	0	x 12	11.4	2012-04-29	5	14	11	14	14	12	14	12	13	10	14	12.8
2012-04-30	6	21	20	20	19	18	15	17	19	0	21	17.0	2012-04-30	6	21	20	20	17	23	18	19	13	19	21	19.1
Total		43	39	38	36	37	36	35	34	6 ¹	39	34.3 (±10.3)	Total		42	38	40	38	41	39	37	32	34	39	38.0 (±3.1)

Concentration (	%)					Re	plicate					Mean Young	Concentration	n (%)					Rej	olicate					Mean Young
6.25	Day	1	2	3	4	5	6	7	8	9	10	(±SD)	100	Day	. 1	2	3	4	5	6	7	8	9	10	(±SD)
2012-04-25	1	0	0	0	0	0	0	0	0	0	0	0	2012-04-25	1	0	0	0	0	0	0	0	0	0	0	0
2012-04-26	2	0	0	0	0	0	0	0	0	0	0	0	2012-04-26	2	0	0	0	0	0	0	0	0	0	0	0
2012-04-27	3	0	0	0	0	0	0	0	0	0	0	0	2012-04-27	3	0	0	0	0	0	0	0	0	0	0	0
2012-04-28	4	6	6	4	6	7	7	6	6	6	4	5.8	2012-04-28	4	6	5	4	4	7	6	5	4	5	5	5.1
2012-04-29	5	15	8	1	x 12	13	13	11	12	11	15	11.1	2012-04-29	5	10	13	9	10	10	9	10	10	11	10	10.2
2012-04-30	6	17	19	0	19	20	18	15	19	12	22	16.1	2012-04-30	6	7	14	1	0	0	0	0	11	16	0	4.9
Total		38	33	5	37	40	38	32	37	29	41	33.0 (±10.5)	Total		23	32	14	14	17	15	15	25	32	15	20.2 (±7.3)

NOTES: •All young produced by a test organism during its fourth and subsequent broods were discarded and not included in the above counts. The presence of two or more neonates in any test chamber, during any given day of the test, constitutes a brood.

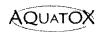
Data Reviewed By : <u>5</u>M Date : 2012-05-02

[•]¹ Outlier according to Grubbs Test (CETIS)^a. Outlying data points were not excluded from statistical analysis, since they could not be attributed to error.

[&]quot;x"= test organism mortality

[&]quot;*"= accidental test organism mortality

[&]quot;-"=4th brood (see 'NOTES')



#### Ceriodaphnia dubia Water Chemistry Data

		Initial Chemistry:	Temp. (°C) 25.0	DO (mg/L) 8.4	<b>рН</b> 8.5	Conductivity (µmhos/cm) 502	Hardness (mg/L as CaCO ₃ ) 250
		Day 0 - 1	Day 1 - 2	Day 2 - 3	Day 3 - 4	Day 4 - 5	Day 5 - 6
Date:		2012-04-24	2012-04-25	2012-04-26	2012-04-27	2012-04-28	2012-04-29
Sub-sample Use		1	1	1	2	2	3
Temperature (°C	*	25.0	24.0	24.0	24.5	24.0	24.0
Dissolved Oxyge		8.4	8.0	8.9	8.7	8.9	8.8
Dissolved Oxyge	en % Sat.	110	101	111	108	110	108
рH		8.5	8.4	8.4	8.5	8.4	8.4
Pre-aeration Tir	me (min)⁴	20	20	20	20	20	20
Analyst(s)	Initial	CL	CL	AW	AW	MR	CL
• , ,	Final	CL	CL	CL	MR	CL	AW
Control (0%)							
Temp. (°C)	Initial	24.0	24.0	24.0	24.0	24.0	24.0
	Final	24.0	24.0	24.0	24.0	24.0	24.0
DO % Sat.3	Initial	95	98	97	97	97	100
DO (mg/L)	Initial	7.5	7.9	7.8	8.0	8.1	8.2
	Final	7.2	7.1	7.4	7.5	7.1	7.8
pН	Initial	8.3	8.3	8.3	8.4	8.4	8.3
	Final	8.1	8.1	8.1	8.3	8.2	8.2
Cond. (µmhos)	Initial	502	502	485	493	505	476
1.56 %							
Temp. (°C)	Initial	24.0	24.0	24.0	24.0	24.0	24.0
	Final	24.0	24.0	24.0	24.0	24.0	24.0
DO (mg/L)	Initial	7.6	7.9	7.8	8.0	8.1	8.2
	Final	7.2	7.1 8.3	7.3 8.3	7.4 8.4	7.1	7.3
pН	Initial Final	8.4 8.1	8.1	8.1	8.2	8.4 8.1	8.4 8.1
Cond. (µmhos)	r mai Initial	503	504	486	483	484	479
,	muai	303	304	400	403	***	479
25 %							• • •
Temp. (°C)	Initial	24.0	24.0	24.0	24.0	24.0	24.0
DO (/L)	Final	24.0	24.0	24.0	24.0	24.0	24.0
DO (mg/L)	Initial Final	7.7 7.2	7.9 7.1	7.9 7.1	8.0 7.4	8.2 7.1	8.3 7.3
pН	Initial	8.4	8.4	8.4	8.4	8.4	8.4
pri	Final	8.2	8.2	8.2	8.3	8.2	8.2
Cond. (µmhos)	Initial	506	505	491	489	492	486
100 %							
Temp. (°C)	Initial	24.0	24.0	24.0	24.0	24.0	24.0
- 2b. ( O)	Final	24.0	24.0	24.0	24.0	24.0	24.0
DO (mg/L)	Initial	8.2	8.0	8.4	8.4	8.3	8.5
/	Final	6.9	6.8	7.0	7.2	6.7	7.0
pН	Initial	8.5	8.5	8.5	8.5	8.5	8.5
	Final	8.4	8.4	8.4	8.3	8.4	8.4
Cond. (µmhos)	Initial	507	507	505	506	505	506

[&]quot;_" = not measured

 $^{^3\ \%}$  saturation (adjusted for actual temperature and barometric pressure)

⁴ ≤100 bubbles/minute



AquaTox Testing & Consulting Inc.

11B Nicholas Beaver Rd. RR 3

Guelph ON N1H 6H9

Tel: (519) 763-4412 Fax: (519) 763-4419

Ceriodaphnia dubia Test Report Survival and Reproduction

2012-04-23

2012-04-24

2012-04-25

15:50

10:30

1 of 4

SAMPLE IDENTIFICATION

Company:

Work Order: Sample Number:

SGS Lakefield Research Limited

Location:

Lakefield ON Lake Water

Substance: Sampling Method:

Grab

221257

33962

Sampled By: Temp. on arrival:

B. Bowman 13.0°C

Sample Description:

Cloudy, orange, odourless

Test Method:

Test of Reproduction and Survival using the Cladoceran Ceriodaphnia dubia. Environment Canada,

Conservation and Protection. Ottawa, Ontario. Report EPS 1/RM/21, 2nd ed. (February 2007).

**TEST RESULTS** 

Effect 95% Confidence Limits Value Statistical Method

LC50 >100%

IC25 (Reproduction) >100%

The results reported relate only to the sample tested.

SODIUM CHLORIDE REFERENCE TOXICANT DATA

Date Tested:

1C25 Reproduction:

Statistical Method:

95% Confidence Limits:

2012-04-25 Organism Batch:

Cd12-04

0.86 g/L

0.76 - 1.11 g/L

Linear Interpolation (CETIS)^a

1.04 g/L

Historical Mean 1C25: Warning Limits ( $\pm$  2SD): 0.60 - 1.81 g/L Analyst(s):

Test Duration:

LC50: 2.09 g/L 0.65 - 3.00 g/L

95% Confidence Limits: Statistical Method:

Nonlinear Interpolation (Stephan)^c

Historical Mean LC50: 2.17 g/L Warning Limits ( $\pm$  2SD): 1.45 - 3.24 g/L

Date Collected:

Time Collected:

Date Received:

Time Received:

Date Tested:

The reference toxicity test was performed under the same experimental conditions as those used with the test sample.

**TEST CONDITIONS** 

Sample Filtration:

Test Aeration:

None None Test Volume per Replicate: 15 mL

Test Vessel: Depth of Test Solution: 22 mL polystyrene vial

VC/MR/RD

7 days

pH Adjustment: Hardness Adjustment: None None

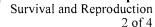
Organisms per Replicate:

4.0 cm 1 10

Daily Renewal Method: Control/Dilution Water: Transferred to fresh solutions Well water (no chemicals added) Number of Replicates: Test Method Deviation(s): None

**COMMENTS** 

•All test validity criteria as specified in the test method cited above were satisfied.





Work Order:

221257

Sample Number:

33962

TEST	OB	CA	NI	SM	C
11.51	U/K	1 v A		-> IVI	

Test Organism:

Ceriodaphnia dubia

Organism Batch:

Cd12-04

Organism Origin:

Total (first three broods):

Single in-house mass culture

Test Organism Origin: Individual in-house cultures

Range of Age (at start of test):

19:30 h - 20:40 h

Mean Brood Organism Mortality:

0%

Ephippia in Culture:

No

**Brood Organism Neonate Production** 

Replicate: Total (third or subsequent brood):

1 19 28

3 20 31

5 6 20 18 30 33

8 22 23 29 32

9 16 27

10 Mean 19.4 18 30 30.2

No organisms exhibiting unusual appearance, behaviour, or undergoing unusual treatment were used in the test.

2

18

31

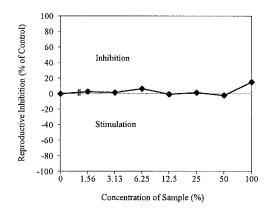
#### **TEST DATA**

20

31

#### Ceriodaphnia dubia Reproductive Inhibition

#### Cumulative Daily Test Organism Mortality (%)



				Concer	tration	of Samp	le (%)		
Date	Test Day	Control	1.56	3.13	6.25	12.5	25	50	100
2012-04-26	1	0	0	0	0	0	0	0	0
2012-04-27	2	0	0	0	0	0	0	0	0
2012-04-28	3	0	0	0	0	0	0	0	0
2012-04-29	4	0	0	0	0	0	0	0	0
2012-04-30	5	0	0	0	0	0	10	0	0
2012-05-01	6	0	0	0	. 0	0	10	0	0
Total Mor	rtality (%)	0	0	0	0	0	10	0	0

#### REFERENCES

^a CETIS, © 2001-2007. Comprehensive Environmental Toxicity Information System. Tidepool Scientific Software, McKinleyville, Calif. 95519[Program on disk and printed User's Guide].

^bGrubbs, F.E., 1969. Procedures for detecting outlying observations in samples. *Technometrics*, 11:1-21.

^c Stephan, C. E. 1977. Methods for calculating an LC50. pp 65-84 in: P. L. Mayer and J. L. Hamelink (eds.), Aquatic Toxicology and Hazard Evaluation. Amer. Soc. Testing and Materials, Philadelphia PA. ASTM STP 634.



#### Ceriodaphnia dubia Survival and Reproduction

Test Initiation Date:

2012-04-25

rest intitution isute.	2012-07-23
Initiation Time:	I1:00
Test Completion Date:	2012-05-01

Concentration (	%)					Re	olicate					Mean Young	Analyst(	s) Concentration	n (%)					Re	plicate					Mean Young
Control	Day	1	2	3	4	5	6	7	8	9	10	(±SD)		12.5	Day	1	2	3	4	5	6	7	8	9	10	(±SD)
2012-04-26	1	0	0	0	0	0	0	0	0	0	0	0	VC	2012-04-26	1	0	0	0	0	0	0	0	0	0	0	0
2012-04-27	2	0	0	0	0	0	0	0	0	0	0	0	VC	2012-04-27	2	0	0	0	0	0	0	0	0	0	0	0
2012-04-28	3	6	7	6	4	6	6	6	6	5	6	5.8	MR	2012-04-28	3	7	6	6	5	4	5	6	6	6	7	5.8
2012-04-29	4	0	0	0	0	0	0	0	0	0	0	0	JGG	2012-04-29	4	0	0	0	0	0	0	0	0	0	0	0
2012-04-30	5	11	13	13	11	12	11	12	11	11	10	11.5	ΑW	2012-04-30	5	12	14	12	9	10	12	10	10	14	13	11.6
2012-05-01	6	18	18	17	16	17	17	17	17	17	15	16.9	AW	2012-05-01	6	18	18	19	12	18	20	16	16	19	14	17.0
Total		35	38	36	31	35	34	35	34	33	31	34.2 (±2.1)	)	Total		<b>3</b> 7	38	<b>3</b> 7	26	32	<b>3</b> 7	32	32	39	34	34.4 (±4.0)

Concentration (	%)					Rep	olicate	:				Mean Young	Concentration	n (%)					Rej	plicate					Mean Young
1.56	Day	1	2	3	4	5	6	7	8	9	10	(±SD)	25	Day	1	2	3	4	5	6	7	8	9	10	(±SD)
2012-04-26	1	0	0	0	0	0	0	0	0	0	0	0	2012-04-26	1	0	0	0	0	0	0	0	0	0	0	0
2012-04-27	2	0	0	0	0	0	0	0	0	0	0	0	2012-04-27	2	0	0	0	0	0	0	0	0	0	0	0
2012-04-28	3	7	6	5	6	6	7	6	3	5	5	5.6	2012-04-28	3	5	7	6	4	6	8	6	6	7	6	6.1
2012-04-29	4	0	0	0	0	0	0	0	0	18	0	1.8	2012-04-29	4	0	0	0	0	0	0	0	0	0	0	0
2012-04-30	5	12	12	12	12	11	13	11	0	0	14	9.7	2012-04-30	5	11	13	10	12	12	13	14	12 2	(11	14	12.2
2012-05-01	6	17	17	17	18	17	17	15	10	16	18	16.2	2012-05-01	6	17	16	20	16	18	19	15	0	16	17	15.4
Total		36	35	34	36	34	<b>3</b> 7	32	13 ¹	39	37	33.3 (±7.4)	Total		33	36	36	32	36	40	35	18 ¹	34	<b>3</b> 7	33.7 (±5.9)

Concentration (	%)					Rep	olicate					Mean Young	Concentratio	n (%)					Re	olicate					Mean Young
3.13	Day	1	2	3	4	5	6	7	8	9	10	(±SD)	50	Day	1	2	3	4	5	6	7	8	9	10	(±SD)
2012-04-26	1	0	0	0	0	0	0	0	0	0	0	0	2012-04-26	1	0	0	0	0	0	0	0	0	0	0	0
2012-04-27	2	0	0	0	0	0	0	0	0	0	0	0	2012-04-27	2	0	0	0	0	0	0	0	0	0	0	0
2012-04-28	3	6	7	7	5	6	5	5	6	3	5	5.5	2012-04-28	3	7	6	7	6	4	5	6	6	6	6	5.9
2012-04-29	4	0	0	0	0	0	0	0	0	0	0	0	2012-04-29	4	0	0	0	0	0	0	0	0	0	0	0
2012-04-30	5	12	11	13	12	9	13	13	12	12	10	11.7	2012-04-30	5	13	10	10	11	11	12	12	12	13	10	11.4
2012-05-01	6	16	18	18	19	17	17	14	16	15	15	16.5	2012-05-01	6	16	17	16	18	20	18	17	21	15	18	17.6
Total		34	36	38	36	32	35	32	34	30	30	33.7 (±2.7)	Total		36	33	33	35	35	35	35	39	34	34	34.9 (±1.7)

Concentration (	%)					Re	plicate					Mean Young	Concentration	n (%)					Re	olicate					Mean Young
6.25	Day	1	2	. 3	4	. 5	6	7	8	9	10	(±SD)	100	Day	1	2	3	4	5	6	7	8	9	10	(±SD)
2012-04-26	1	0	0	0	0	0	0	0	0	0	0	0	2012-04-26	1	0	0	0	0	0	0	0	0	0	0	0
2012-04-27	2	0	0	0	0	0	0	0	0	0	0	0	2012-04-27	2	0	0	0	0	0	0	0	0	0	0	0
2012-04-28	3	4	5	6	6	6	2	6	4	5	7	5.1	2012-04-28	3	6	5	3	4	5	5	5	4	4	6	4.7
2012-04-29	4	0	0	0	0	0	0	0	0	0	0	0	2012-04-29	4	0	0	0	0	0	0	0	0	0	0	0
2012-04-30	5	12	12	12	12	11	8	11	11	0	11	10.0	2012-04-30	5	6	10	8	9	10	10	7	8	8	11	8.7
2012-05-01	6	19	17	15	16	18	13	18	17	17	19	16.9	2012-05-01	6	16	18	16	15	16	17	10	14	17	17	15.6
Total		35	34	33	34	35	23	35	32	22	37	32.0 (±5.2)	Total		28	33	27	28	31	32	22	26	29	34	29.0 (±3.6)

NOTES: •All young produced by a test organism during its fourth and subsequent broods were discarded and not included in the above counts. The presence of two or more neonates in any test chamber, during any given day of the test, constitutes a brood.

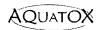
Data Reviewed By : 500 Date : 202-05-03

[•]¹ Outlier according to Grubbs Test^b. Outlying data points were not excluded from statistical analysis, since they could not be attributed to error.

[&]quot;x"= test organism mortality

[&]quot;*"= accidental test organism mortality

[&]quot;-"=4th brood (see 'NOTES')



#### Ceriodaphnia dubia Water Chemistry Data

		Initial Chemistry:	Temp. (°C) 24.0	DO (mg/L) 8.1	рН 8.5	Conductivity (µmhos/cm) 500	Hardness (mg/L as CaCO ₃ ) 260
		Day 0 - 1	Day 1 - 2	Day 2 - 3	Day 3 - 4	Day 4 - 5	Day 5 - 6
Date:		2012-04-25	2012-04-26	2012-04-27	2012-04-28	2012-04-29	2012-04-30
Sub-sample Use		1	1	1	2	2	3
Temperature (°	•	24.0	24.0	24.5	24.0	24.0	24.0
Dissolved Oxyge		8.1	8.6	8.8	8.8	8.9	9.0
Dissolved Oxyge	en % Sat.3	102	108	110	108	109	110
pН		8.5	8.4	8.4	8.4	8.4	8.4
Pre-aeration Ti	me (min) ⁴	20	20	20	20	20	20
Analyst(s)	Initial Final	CL CL	AW AW	AW JGG	JGG CL	CL AW	AW HL(RD)
Control (0%)	)						
Temp. (°C)	Initial	24.0	24.0	24.0	25.0	24.0	24.0
,	Final	24.0	24.0	24.0	24.0	24.0	24.5
DO % Sat.3	Initial	97	97	97	96	100	98
DO (mg/L)	Initial	7.8	7.7	7.9	8.0	8.2	8.0
	Final	7.2	7.3	7.4	7.5	7.4	7.4
pН	Initial	8.4	8.4	8.4	8.3	8.4	8.4
	Final	8.1	8.1	8.1	8.2	8.2	7.4
Cond. (µmhos)	Initial	500	484	482	478	476	477
1.56 %							
Temp. (°C)	Initial	24.0	24.0	24.0	25.0	24.0	24.0
50( 5)	Final	24.0	24.0	24.0	24.0	24.0	24.5
DO (mg/L)	Initial	7.9	7.8	7.9	8.0	8.2	8.1
pН	Final Initial	7.3 8.3	7.3 8.4	7.4 8.4	7.6	7.5	7.3
рп	Final	8.2	8.4 8.1	8.4 8.2	8.4 8.3	8.4 8.1	8.3 8.0
Cond. (µmhos)	Initial	503	484	483	482	479	479
25 %							
Temp. (°C)	Initial	24.0	24.0	24.0	25.0	24.0	24.0
p· ()	Final	24.0	24.0	24.0	24.0	24.0	24.5
DO (mg/L)	Initial	7.9	7.9	8.0	8.0	8.3	8.1
	Final	7.2	7.2	7.3	7.4	7.4	7.4
pН	Initial	8.3	8.4	8.4	8.5	8.5	8.3
	Final	8.2	8.2	8.3	8.3	8.2	8.1
Cond. (µmhos)	Initial	502	491	489	489	487	490
100 %							
Temp. (°C)	Initial	24.0	24.0	24.0	25.0	24.0	24.0
	Final	24.0	24.0	24.0	24.0	24.0	24.5
DO (mg/L)	Initial	8.1	8.3	8.3	8.4	8.5	8.2
	Final	7.0	7.0	7.2	7.4	7.3	6.9
pH	Initial Final	8.5 8.4	8.5	8.5	8.5	8.5	8.5
Cond. (µmhos)	Finai Initial	8.4 506	8.4 504	8.5 505	8.5 506	8.4 506	8.3 507
Cond. (µiiiios)	minai	300	JU4	202	500	500	507

[&]quot;_" = not measured

 $^{^{\}rm 3}$  % saturation (adjusted for actual temperature and barometric pressure)

⁴ ≤100 bubbles/minute



# Chronic Bioassay Test Results: *Lemna Minor* (duck weed)

Growth Inhibition 1 of 4



AquaTox Testing & Consulting Inc.

11B Nicholas Beaver Rd. RR3

Guelph ON N1H 6H9 Tel: (519) 763-4412 Fax: (519) 763-4419

Work Order: 221257 Sample Number: 33960

SAMPLE IDENTIFICATION

Company:

SGS Lakefield Research Limited

Location:

Lakefield ON

Date Collected:

2012-04-23

Substance:

PP-22 BZ-MP Treated Process Water-2

Time Collected: Date Received:

15:50 2012-04-24

Sampling Method: Sampled By:

B. Bowman

Time Received:

10:30

Temp. on arrival:

13.0°C

Grab

Date Tested:

2012-04-26

Sample Description:

Clear, colourless, odourless.

Test Method:

Test for Measuring the Inhibition of Growth using the Freshwater Macrophyte, Lemna minor. Method Development and Application Section, Environmental Technology Centre, Environment

Canada. Ottawa, Ontario. Report EPS 1/RM/37, 2nd ed. (January 2007).

TEST RESULTS 95% Confidence Limits

Effect Value

Statistical Method

IC25 (Weight)

>97.0%

IC25 (Frond Production)

>97.0%

The results reported relate only to the sample tested.

POTASSIUM CHLORIDE REFERENCE TOXICANT DATA

Date Tested:

Statistical Method:

Organism Batch:

Historical Geometric Mean IC25:

Test Duration:

Warning Limits  $(\pm 2SD)$ :

IC25 (Frond Production):

Growth Medium:

95% Confidence Limits:

Analyst(s):

The reference toxicant test was performed under the same experimental conditions as those used with the test sample.

**TEST CONDITIONS** 

Test Organism:

Lemna minor L., Strain 7730

Test Type:

Static (no sub-samples required)

Organism Batch:

Lm12-04

Control/Dilution Medium:

Culture Origin:

**UTCC 492** 

Medium Preparation Water:

Modified APHA Distilled Water

Test Organism Source:

Axenic in-house culture

Source of Water:

Fernbrook Springs

Culture Medium:

Modified Hoaglands E+

Medium Preparation Chemicals:

Modified APHA stocks A, B, C (10 mL/L)

Age (on Test Day 0):

9 days

Nutrient Spiking of Sample:

Modified APHA stocks A, B, C (10 mL/L)

Health Criteria (in APHA): 15.7-fold frond increase in 7 days Replicates per Concentration:

4

Organism Acclimation:

20:50 h in APHA medium

Test Volume per Replicate:

100 mL

Inoculum (Test Day 0):

2 plants (3 fronds per plant)

Test Vessel:

250 mL glass Erlenmeyer flask

Sample Filtration:

1 μm (Whatman GF/C)

Depth of Test Solution: Photoperiod/Light Intensity:

Continuous, 4313 - 5005 lux

Sample Pre-aeration: pH Adjustment:

20 min. at ≤100 bubbles/min.

Test Method Deviation(s):

4.0 cm

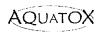
None

None Hardness Adjustment: None

**COMMENTS** 

[•]All test validity criteria as specified in the test method cited above were satisfied.

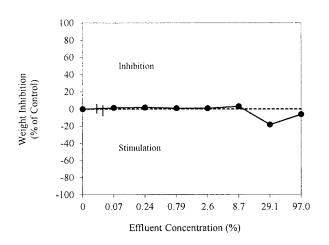


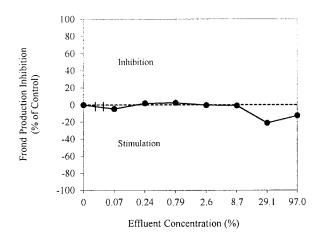


Work Order : Sample Number :

221257 33960

#### Lemna minor Growth Inhibition





#### **TEST MONITORING**

Initiation Date: 2012-04-26
Intitiation Time: 12:05
Initiated By: SM

Termination Date: 2012-05-03
Termination Time: 15:00
Terminated By: AW

#### **Temperature Monitoring Test Day** Date Temperature (°C) 0 (unmodified sample) 25.0 2012-04-26 0 2012-04-26 26.0 1 2012-04-27 26.0 2 2012-04-28 26.5 3 2012-04-29 25.5 4 2012-04-30 26.0

2012-05-01

2012-05-02

2012-05-03

pH Mon	itoring	
Concentration (%)	Day 0	Day 7
100 (unmodified sample)	7.3	_
Control	8.2	8.3
0.07	8.2	8.3
0.24	-	uname
0.79		_
2.6	8.2	8.3
8.7		
29.1		_
97.0	7.9	8.7

5

6

7

#### REFERENCES

26.0

26.0

25.5

Date :		Approved By:	
	yyyy-mm-dd		Project Manager

[&]quot;-" = not required

^a CETIS, © 2001-2007. Comprehensive Environmental Toxicity Information System. Tidepool Scientific Software, McKinleyville, Calif. 95519 [Program on disk and printed User's Guide].

3 of 4



Work Order: Sample Number:

221257 33960

#### Lemna minor Frond Increase

Concentration (%)	Replicate	Frond Count Day 0*	Frond Count Day 7	Frond Increase		Standard Deviation	CV (%)	Stimulation (%)**	Frond/Root Appearance (Day 7)
(70)	A	6	88	82	Hierease	Deviation		( /0)	to the should all the state of
Control	В	6	106	100	90.75	8.22	9.1	_	Fronds healthy, appearance normal in
Control	C	6	92	86	70.75	0.22	7.1		all replicates.
	D	6	101	95					an representation
-	A	6	99	93					
0.07	В	6	100	94	95.00	3.37	3.5	_	Fronds healthy, appearance normal in
0,07	C	6	99	93					all replicates.
	D	6	106	100					1
	A	6	83	77					
0.24	В	6	92	86	89.00	10.80	12.1	_	Fronds healthy, appearance normal in
	С	6	96	90					all replicates.
	D	6	109	103					•
	A	6	99	93					
0.79	В	6	96	90	88.50	4.65	5.3	_	Fronds healthy, appearance normal in
	C	6	95	89					all replicates.
	D	6	88	82					•
	A	6	99	93					
2.6	В	6	93	87	91.00	4.90	5.4	_	Fronds healthy, appearance normal in
	С	6	93	87					all replicates.
	D	6	103	97					
	A	6	105	99					
8.7	В	6	101	95	91.50	8.70	9.5	_	Fronds healthy, appearance normal in
	C	6	99	93		/			all replicates.
	D	6	85	79					
	A	6	94	88	-				
29.1	В	6	132	126	110.00	16.08	14.6	21.2	Fronds healthy, appearance normal in
	C	6	122	116					all replicates.
	D	6	116	110					
	A	6	108	102					
97.0	В	6	96	90	102.25	8.73	8.5	-	Fronds healthy, appearance normal in
	C	6	114	108					all replicates.
	D	6	115	109					

**NOTES:** *No unusual appearance or treatment of culture prior to testing. Test inoculated with healthy plants.

"-" = not available/not required

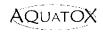
Test Data Reviewed By:

^{**}Significant stimulation (ANOVA - Dunnett's Test (CETIS)a,  $\alpha$ =0.05) compared to control.

[•]A 16.1-fold increase in frond number was observed in the control over the testing period.

[•]No outlying data points were detected according to Grubbs Test (CETIS)^a.





#### Lemna minor Frond Weight Data

Concentration (%)	Replicate	Dry Weight of Fronds (mg)	Treatment Mean Dry Weight (mg)	Standard Deviation	Stimulation (%)**
	A	9.29			
Control	В	10.59	9.85	0.73	_
	С	9.17			
	D	10.36			
	A	9.78			
0.07	В	9.99	9.71	0.22	_
	С	9.52			
	D	9.56			
	A	7.98			
0.24	В	9.53	9.67	1.30	-
	C	10.04			
	D	11.11			
	A	10.97			***************************************
0.79	В	9.85	9.75	0.89	_
	C	8.97			
	D	9.21			
	A	9.98			
2.6	В	9.53	9.76	0.46	_
	C	9.25			
	D	10.28			
	A	10.35			
8.7	В	9.97	9.55	0.74	_
	C	9.13		•	
	D	8.75			
	A	9.64		-	
29.1	В	13.81	11.65	1.70	18.27
	С	11.57			
	D	11.59			
	A	10.62			
97.0	В	9.56	10.46	0.73	_
	С	11.33			
	D	10.33			

NOTES:

^{**}Significant stimulation (ANOVA - Dunnett's Test (CETIS)a,  $\alpha$ =0.05) compared to control.

[•]No outlying data points were detected according to Grubbs Test (CETIS)^a.

[&]quot;_" = not available/not required

#### Lemna minor Test Report

2012-04-23

2012-04-24

2012-04-26

15:50

10:30

Growth Inhibition





AquaTox Testing & Consulting Inc.

11B Nicholas Beaver Rd. RR 3 Guelph ON N1H 6H9

Tel: (519) 763-4412 Fax: (519) 763-4419

Work Order: 221257 Sample Number: 33961

SAMPLE IDENTIFICATION

SGS Lakefield Research Limited Company:

Location: Lakefield ON

PP-22 BZ-MP Treated Process + Lake Water Substance:

Sampling Method:

Grab Sampled By: B. Bowman

Temp. on arrival: 13.0°C

Sample Description: Clear, orange, odourless

Test Method: Test for Measuring the Inhibition of Growth using the Freshwater Macrophyte, Lemna minor.

Method Development and Application Section, Environmental Technology Centre, Environment

Date Collected:

Time Collected:

Date Received:

Time Received:

Date Tested:

Canada, Ottawa, Ontario. Report EPS 1/RM/37, 2nd ed. (January 2007).

**TEST RESULTS** Effect Value 95% Confidence Limits Statistical Method IC25 (Weight) >97.0% IC25 (Frond Production) >97.0% The results reported relate only to the sample tested.

POTASSIUM CHLORIDE REFERENCE TOXICANT DATA

Statistical Method: Date Tested:

Organism Batch: Historical Geometric Mean IC25:

Warning Limits ( $\pm$  2SD): Test Duration: IC25 (Frond Production): Growth Medium:

95% Confidence Limits: Analyst(s):

The reference toxicant test was performed under the same experimental conditions as those used with the test sample.

#### **TEST CONDITIONS**

Test Organism: Lemna minor L., Strain 7730 Test Type: Static (no sub-samples required)

Organism Batch: Lm12-04 Modified APHA Control/Dilution Medium: Culture Origin: **UTCC 492** Medium Preparation Water: Distilled Water Test Organism Source: Axenic in-house culture Source of Water: Fernbrook Springs

Culture Medium: Modified Hoaglands E+ Medium Preparation Chemicals: Modified APHA stocks A, B, C (10 mL/L) Age (on Test Day 0): 9 days Nutrient Spiking of Sample: Modified APHA stocks A, B, C (10 mL/L)

Health Criteria (in APHA): 15.7-fold frond increase in 7 days Replicates per Concentration:

Organism Acclimation: 21:05 h in APHA medium Test Volume per Replicate: 100 mL

Inoculum (Test Day 0): 2 plants (3 fronds per plant) Test Vessel: 250 mL glass Erlenmeyer flask

4.0 cm Sample Filtration: 1 μm (Whatman GF/C) Depth of Test Solution:

Continuous, 4950 - 5442 lux Sample Pre-aeration: 20 min. at  $\leq 100$  bubbles/min. Photoperiod/Light Intensity:

Test Method Deviation(s): pH Adjustment: None None

Hardness Adjustment: None

#### COMMENTS

[•]All test validity criteria as specified in the test method cited above were satisfied.

7

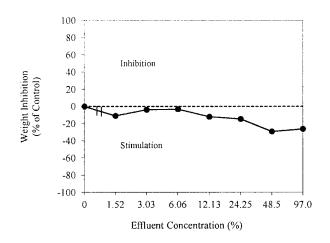


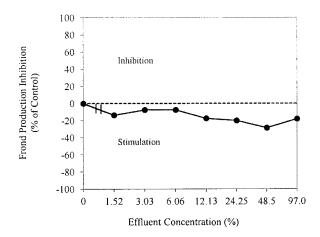
Work Order:

221257

Sample Number: 33961







#### **TEST MONITORING**

Initiation Date: 2012-04-26 Intitiation Time: 12:20

Initiated By: AS Termination Date: 2012-05-03 Termination Time: 13:30 Terminated By: AS

pH Monitoring

## **Temperature Monitoring**

rompera		~~8	Pinnisia	I				
<b>Test Day</b>	Date	Temperature	Concentration (%)	Day 0	Day 7			
		(°C)						
0 (unmodified sample)	2012-04-26	25.0	100 (unmodified sample)	8.4	_			
0	2012-04-26	25.0	Control	8.2	8.4			
1	2012-04-27	25.0	1,52	8.2	8.4			
2	2012-04-28	24.5	3.03	-	-			
3	2012-04-29	24.5	6.06	-	-			
4	2012-04-30	24.5	12.13	8.2	8.6			
5	2012-05-01	24.5	24.25	-	-			
6	2012-05-02	24.5	48.5	-	-			
7	2012-05-03	24.5	97.0	8.3	8.9			

[&]quot;-" = not required

#### REFERENCES

Date :		Approved By:			
	yyyy-mm-dd		Project Manager		

^a CETIS, © 2001-2007. Comprehensive Environmental Toxicity Information System. Tidepool Scientific Software, McKinleyville, Calif. 95519 [Program on disk and printed User's Guide].



Work Order: Sample Number:

221257 33961

#### Lemna minor Frond Increase

Concentration (%)	Replicate	Frond Count Day 0*	Frond Count Day 7	Frond Increase		Standard Deviation	CV (%)	Stimulation (%)**	Frond/Root Appearance (Day 7)
	А	6	97	91					
Control	В	6	90	84	90.00	9.49	10.5	_	Fronds healthy, appearance normal in
	C	6	88	82					all replicates.
	D	6	109	103					
	Α	6	108	102					
1.52	В	6	126	120¹	102.25	12.50	12.2	_	Fronds healthy, appearance normal in
	C	6	100	94					all replicates.
	D	6	99	93					
	A	6	96	90					
3.03	В	6	99	93	96.75	6.40	6.6		Fronds healthy, appearance normal in
	C	6	106	100					all replicates.
	D	6	110	104					
	A	6	110	104					
6.06	В	6	100	94	96.75	8.46	8.7		Fronds healthy, appearance normal in
	C	6	92	86					all replicates.
	D	6	109	103					
	А	6	97	91					
12.13	В	6	122	116	105.75	10.81	10.2	_	Fronds healthy, appearance normal in
	C	6	117	111					all replicates.
	D	6	111	105					
	Α	6	109	103					
24.25	В	6	120	114	108.00	9.70	9.0	20.0	Fronds healthy, appearance normal in
	C	6	124	118		/			all replicates.
	D	6	103	97					
	A	6	124	118	<b>X</b>				
48.5	В	6	109	103	115.75	9.91	8.6	28.6	Fronds healthy, appearance normal in
	C	6	121	115					all replicates.
	D	6	133	127					•
	A	6	112	106				,	Possida hashdan anna anna anna anna
97.0	В	6	107	101	106.25	4.11	3.9	18.1	Fronds healthy, appearance normal in
	С	6	117	111					all replicates. Solutions slightly green
	D	6	113	107					in colour.

NOTES: *No unusual appearance or treatment of culture prior to testing. Test inoculated with healthy plants.

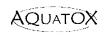
"-" = not available/not required

Test Data Reviewed By: 14H
Date: 2012-15-07

^{**}Significant stimulation (ANOVA - Dunnett's Test (CETIS)a,  $\alpha$ =0.05) compared to control.

[•]A 16.0-fold increase in frond number was observed in the control over the testing period.

[•] ¹Outlier according to Grubbs Test (CETIS)ª. Outlying data points were not excluded from statistical analysis, since they could not be attributed to error.



#### Lemna minor Frond Weight Data

Concentration (%)	Replicate	Dry Weight of Fronds (mg)	Treatment Mean Dry Weight (mg)	Standard Deviation	Stimulation (%)**	
· · · · · · · · · · · · · · · · · · ·	A	9.96				
Control	В	8.45	9.28	0.65	-	
	C	9.56				
	D	9.14				
	A	10.29				
1.52	В	$12.01^{2}$	10.29	1.26	_	
	С	9.81				
	D	9.05				
	A	9.00				
3.03	В	9.46	9.63	0.61	_	
	С	10.45				
	D	9.62				
	A	10.08				
6.06	В	9.74	9.57	0.42		
	С	9.29				
	D	9.16				
	Α	9.67	7			
12.13	В	11.84	10.39	1.11	paymen.	
	С	10.67				
	D	9.39				
	Α	9.94				
24.25	В	11.22	10.63	0.83	, allegania.	
	С	11.47				
	D	9.90				
	Α	11.84				
48.5	В	10.80	11.99	0.90	29.18	
	С	12.41				
	D	12.89				
	Α	10.95				
97.0	В	11.22	11.71	0.75	26.22	
	С	12.15				
	D	12.52				

NOTES:

Test Data Reviewed By: 41
Date: 2012-0507

^{**}Significant stimulation (ANOVA - Dunnett's Test (CETIS)a,  $\alpha$ =0.05) compared to control.

^{• &}lt;sup>2</sup>Outlier according to Grubbs Test (CETIS)^a. Outlying data points were not excluded from statistical analysis, since they could not be attributed to error.

[&]quot;_" = not available/not required

2012-04-23

2012-04-24

2012-04-26

15:50

10:30

Statistical Method



AquaTox Testing & Consulting Inc.

11B Nicholas Beaver Rd. RR 3 Guelph ON N1H 6H9

Tel: (519) 763-4412 Fax: (519) 763-4419

221257 Sample Number: 33962

SAMPLE IDENTIFICATION

Company:

Work Order:

SGS Lakefield Research Limited

Location: Substance: Lakefield ON

Lake Water

Sampling Method: Sampled By: Temp. on arrival:

Test Method:

B. Bowman 13.0°C

Grab

Sample Description:

Cloudy, orange, odourless

Test for Measuring the Inhibition of Growth using the Freshwater Macrophyte, Lemna minor.

Method Development and Application Section, Environmental Technology Centre, Environment

Date Collected:

Time Collected:

Date Received:

Time Received:

Date Tested:

Canada. Ottawa, Ontario. Report EPS 1/RM/37, 2nd ed. (January 2007).

TEST RESULTS 95% Confidence Limits

Effect Value

IC25 (Weight) >97.0%

IC25 (Frond Production) >97.0%

The results reported relate only to the sample tested.

POTASSIUM CHLORIDE REFERENCE TOXICANT DATA

Date Tested:

Statistical Method:

Organism Batch:

Historical Geometric Mean IC25:

Test Duration: IC25 (Frond Production): Warning Limits ( $\pm 2SD$ ):

Growth Medium:

95% Confidence Limits:

Analyst(s):

The reference toxicant test was performed under the same experimental conditions as those used with the test sample.

**TEST CONDITIONS** 

Test Organism:

Lemna minor L., Strain 7730

Test Type:

Static (no sub-samples required)

Organism Batch:

Lm12-04

Control/Dilution Medium:

Modified APHA

Culture Origin:

**UTCC 492** 

Medium Preparation Water:

Distilled Water

Test Organism Source:

Axenic in-house culture

Source of Water:

Fernbrook Springs Modified APHA stocks A, B, C (10 mL/L)

Culture Medium: Age (on Test Day 0): Modified Hoaglands E+ 9 days

Medium Preparation Chemicals: Nutrient Spiking of Sample:

Modified APHA stocks A, B, C (10 mL/L)

Health Criteria (in APHA): 15.7-fold frond increase in 7 days Replicates per Concentration:

4

Organism Acclimation:

22:25 h in APHA medium

Test Volume per Replicate:

250 mL glass Erlenmeyer flask

Inoculum (Test Day 0): Sample Filtration:

2 plants (3 fronds per plant) 1 μm (Whatman GF/C)

Depth of Test Solution:

Sample Pre-aeration:

20 min. at ≤100 bubbles/min.

Photoperiod/Light Intensity:

Continuous, 4159 - 4941 lux

pH Adjustment:

None

Test Method Deviation(s):

None

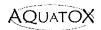
100 mL

Hardness Adjustment: None

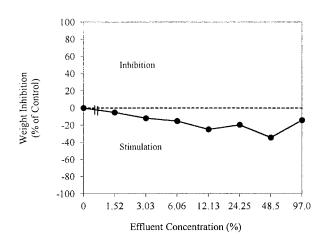
#### COMMENTS

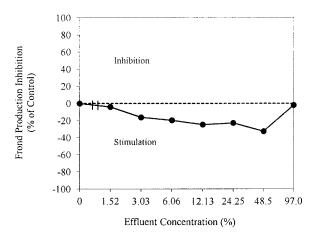
Test Vessel:

[•]All test validity criteria as specified in the test method cited above were satisfied.



#### Lemna minor Growth Inhibition





#### **TEST MONITORING**

Initiation Date: 2012-04-26
Intitiation Time: 13:40
Initiated By: SM

"-" = not required

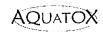
Termination Date: 2012-05-03
Termination Time: 13:15
Terminated By: KEH/RD

#### Temperature Monitoring pH Monitoring Date Temperature **Test Day** Concentration (%) Day 0 Day 7 (°C) 0 (unmodified sample) 2012-04-26 25.0 00 (unmodified sample) 8.5 0 2012-04-26 26.0 Control 8.2 8.4 2012-04-27 8.2 1 26.0 1.52 8.4 2 2012-04-28 26.5 3.03 3 2012-04-29 25.5 6.06 12.13 4 2012-04-30 26.0 8.3 8.6 5 2012-05-01 26.0 24.25 6 2012-05-02 26.0 48.5 7 2012-05-03 25.5 97.0 8.4 8.8

#### **REFERENCES**

Date:	type personal and the second s	Approved By:			
	vvvv-mm-dd		Project Manager		

^a CETIS, © 2001-2007. Comprehensive Environmental Toxicity Information System. Tidepool Scientific Software, McKinleyville, Calif. 95519 [Program on disk and printed User's Guide].



Work Order:
Sample Number:

221257 33962

#### Lemna minor Frond Increase

Concentration (%)	Replicate	Frond Count Day 0*	Frond Count Day 7	Frond Increase		Standard Deviation	CV (%)	Stimulation (%)**	Frond/Root Appearance (Day 7)
	A	6	95	89					
Control	В	6	96	90	94.50	5.80	6.1	_	Fronds healthy, appearance normal in
	С	6	105	99					all replicates.
	D	6	106	100					
	A	6	102	96					
1.52	В	6	117	111	98.25	13.67	13.9		Fronds healthy, appearance normal in
	C	6	86	80					all replicates.
	D	6	112	106					
	А	6	108	102					
3.03	В	6	120	114	109.75	16.01	14.6		Fronds healthy, appearance normal in
	C	6	99	93					all replicates.
	D	6	136	$130^{1}$					
	A	6	126	120					
6.06	В	6	124	118	113.00	7.70	6.8	19.6	Fronds healthy, appearance normal in
	C	6	109	103					all replicates.
	D	6	117	111					
	Α	6	111	105					
12.13	В	6	124	118	117.75	9.29	7.9	24.6	Fronds healthy, appearance normal in
	C	6	133	127					all replicates.
	D	6	127	121					
	Α	6	112	106					
24.25	В	6	135	129	116.00	9.56	8.2	22.8	Fronds healthy, appearance normal in
	C	6	121	115					all replicates.
	D	6	120	114					
	A	6	132	126	*				
48.5	В	6	131	125	125.25	4.11	3.3	32.5	Fronds healthy, appearance normal in
	C	6	136	130					all replicates.
	D	6	126	120				·	
	A	6	93	87					
97.0	В	6	99	93	96.25	11.47	11.9	-	Fronds healthy, appearance normal in
	C	6	98	92				all replicates.	all replicates.
	D	6	119	113					

NOTES: *No unusual appearance or treatment of culture prior to testing. Test inoculated with healthy plants.

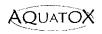
"_" = not available/not required

Test Data Reviewed By:

^{**}Significant stimulation (ANOVA - Dunnett's Test (CETIS)a,  $\alpha$ =0.05) compared to control.

[•]A 16.8-fold increase in frond number was observed in the control over the testing period.

[•] ¹Outlier according to Grubbs Test (CETIS)ª. Outlying data points were not excluded from statistical analysis, since they could not be attributed to error.



Work Order: 221257 Sample Number: 33962

# Lemna minor Frond Weight Data

Concentration (%)	Replicate	Dry Weight of Fronds (mg)	Treatment Mean Dry Weight (mg)	Standard Deviation	Stimulation (%)**
	A	9.18			
Control	В	8.88	9.12	0.17	_
	C	9.27			
	D	9.15			
	A	8.85			
1.52	В	10.56	9.58	1.14	
	C	8.37			
	D	10.55			
	A	9.51			
3.03	В	10.92	10.20	1.40	_
	С	8.61 ²			
	D	11.74			
	A	11.77			
6.06	В	10.03	10.50	0.96	_
	C	9.55			
	D	10.64			
······································	A	11.07	7		
12.13	В	10.86	11.38	0.56	24.75
	C	11.44			
	D	12.14			
	A	10.43			
24.25	В	12.13	10.90	0.83	19.49
	С	10.42			
	D	10.61			
<del></del>	A	12.26			
48.5	В	12.74	12.25	0.41	34.35
	C	11.74			
	D	12.27			
	A	9.61			
97.0	В	11.15	10.40	0.64	_
	C	10.32			
	D	10.53			

NOTES:

Test Data Reviewed By: 1447
Date: 2012-05-08

^{**}Significant stimulation (ANOVA - Dunnett's Test (CETIS)a,  $\alpha$ =0.05) compared to control.

^{• &}lt;sup>2</sup>Outlier according to Grubbs Test (CETIS)^a. Outlying data points were not excluded from statistical analysis, since they could not be attributed to error.

[&]quot;-" = not available/not required



# Chronic Bioassay Test Results: Pseodokirchneriella Subcapitata (algae)



### AquaTox Testing & Consulting Inc.

11B Nicholas Beaver Rd. RR 3 Guelph ON N1H 6H9

Tel: (519) 763-4412 Fax: (519) 763-4419

# Pseudokirchneriella subcapitata

Growth Inhibition

2012-04-23

2012-04-24

2012-04-24

15:50

10:30

1 of 2

Work Order: 221257 Sample Number: 33960

SAMPLE IDENTIFICATION

Company:

SGS Lakefield Research Limited

Location:

Lakefield ON

Substance:

PP-22 BZ-MP Treated Process Water-2

Sampling Method: Grab

Sampled By: B. Bowman

Temp. on arrival: 13.0°C

Sample Description:

Clear, colourless, odourless.

Test Method:

Growth Inhibition Test Using a Freshwater Alga. Environment Canada, Conservation and

Protection. Ottawa, Ontario. Report EPS 1/RM/25, 2nd ed. (March 2007).

**TEST RESULTS** 

Effect Value 95% Confidence Limits

Statistical Method

IC25 (Growth)

>90.91%

The results reported relate only to the sample tested.

ZINC (AS ZINC SULPHATE) REFERENCE TOXICANT DATA

Date Tested:

2012-04-24

Statistical Method:

Nonlinear Regression (CETIS)^a

Organism Batch:

Ps12-04

Historical Mean IC25:

 $11.0 \, \mu g/L$ 

Test Duration:

72 hours

Warning Limits (± 2SD):

4.3 - 28.6 μg/L

IC25 Growth:

 $11.5 \mu g/L$ 

Analyst(s):

VC/SM

Date Collected:

Time Collected:

Date Received:

Time Received:

Date Tested:

95% Confidence Limits: 7.9 - 15.5 μg/L

The reference toxicity test was performed under the same experimental conditions as those used with the test sample.

# **TEST CONDITIONS**

Test Organism:

Pseudokirchneriella subcapitata

Control/Dilution Water:

Millipore Milli-Q (no chemicals added)

Organism Batch:

Ps12-04

Test Vessel:

U-shaped polystyrene microplate

Strain Number:

UTEX (1648)

Volume per Replicate:

Source:

In-house culture

Number of Control Replicates:

220 μL

Culture Origin:

10 4

Age (at start of test):

University of Texas, Austin TX, USA Number of Test Replicates:

Concentrations Tested:

10 + Control

pH Adjustment:

4 days (in exponential growth) None

Photoperiod / Light Intensity:

Continuous light, 4196 - 4368 lux

Hardness Adjustment:

None

Mean Test Temperature (± SD):

 $23.0^{\circ}C (\pm 0.0)$ 

Sample Pre-aeration: Sample Filtration:

None 0.45 µm preconditioned filter Test Duration: Test Method Deviation(s): 72 hours None

Volume Filtered:

≥10 mL

# COMMENTS

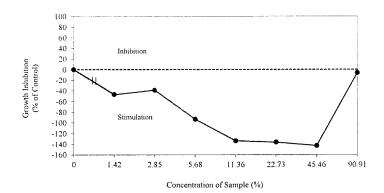
- •All test validity criteria as specified in the test method cited above were satisfied.
- •No unusual appearance or treatment of culture prior to testing.
- •Algal growth curve is determined at least twice per year as required by the test method cited above.



AQUATOX

Work Order: 221257 Sample Number: 33960

## Pseudokirchneriella subcapitata Growth Inhibition



# **CELL ENUMERATION AT 72-HOURS**

Initiation Date: 2012-04-24 Sample pH (at 0 hours): 7.3
Initiated By: VC/SM Control pH (at 0 hours): 6.5
Completion/Enumeration Date: 2012-04-27 Control pH (at 72 hours): 7.0

Enumerated By: KEH Initial Cell Density at 0-h: 10636 cells/mL per microplate well
Enumeration Technique: Manual (hemocytometer) Inoculum Prepared: 00:35 h prior to test initiation

Control Cell Increase Factor: 37.1 times growth

# Cell Concentration (x 10000 cells/mL)

# Cell Yield (x 10000 cells/mL)

Concentration	1			Replicate					Mean	Standard	CV (%)	Stimulation
(%)	1	2	3	4	7	8	9	10		Deviation		(% of control)**
Control	41.5	44.5	35.5	40.5	40.5	43.5	35.0	35.0	38.44	3.85	10.00	_
0.18	_	_							_	Anton	_	<del></del>
0.35	_	_		-					_		_	-
0.71	_	_		_			· ·		_		_	
1.42	65.5	49.5	57.5	_					56.44	8.00	14.2	46.8
2.85	54.5	53.0	55.5	_					53.27	1.26	2.4	_
5.68	79.0	74.5	72.5	_					74.27	3.33	4.5	93.2
11.36	76.5	90.5	105.5	_					89.77	14.50	16.2	133.6
22.73	76.5	94.5	104.5	_					90.77	14.19	15.6	136.2
45.46	82.0	105.5	95.5	name.					93.27	11.79	12.6	142.7
90.91	45.0	51.5	29.0						40.77	11.58	28.4	

NOTES: **Significant stimulation compared to control, according to ANOVA - Dunnett's Test (CETIS)^a,  $\alpha$ =0.05.

- •Control replicates 5 and 6 used for pH measurement.
- •The Mann-Kendall test shows that there is no inhibitory gradient ( $\alpha$ =0.05).
- •No outlying data points were detected according to Grubbs Test (CETIS)^a.

"_" = not enumerated/not required	Data Reviewed By:
	Date: <u>2012-05-03</u>

# REFERENCES

Date :		Approved By:	
	yyyy- <b>mm-</b> dd		Project Manager

^a CETIS, © 2001-2007. Comprehensive Environmental Toxicity Information System. Tidepool Scientific Software, McKinleyville CA 95519. [Program on disk and printed User's Guide].



# AquaTox Testing & Consulting Inc.

11B Nicholas Beaver Rd. RR 3 Guelph ON N1H 6H9

Tel: (519) 763-4412 Fax: (519) 763-4419

# Pseudokirchneriella subcapitata

Growth Inhibition 1 of 2

2012-04-23

2012-04-24

2012-04-24

15:50

10:30

Work Order: Sample Number: 221257 33961

SAMPLE IDENTIFICATION

Company:

SGS Lakefield Research Limited

Location:

Lakefield ON

Substance:

PP-22 BZ-MP Treated Process + Lake Water

Sampling Method:

Grab

Sampled By: Temp. on arrival:

B. Bowman 13.0°C

Sample Description:

Clear, orange, odourless

Test Method:

Growth Inhibition Test Using a Freshwater Alga. Environment Canada, Conservation and

Protection. Ottawa, Ontario. Report EPS 1/RM/25, 2nd ed. (March 2007).

TEST RESULTS

Effect Value 95% Confidence Limits

Statistical Method

IC25 (Growth)

>90.91%

Linear Interpolation (CETIS) a

The results reported relate only to the sample tested.

ZINC (AS ZINC SULPHATE) REFERENCE TOXICANT DATA

Date Tested:

2012-04-24

Statistical Method:

Nonlinear Regression (CETIS)^a

Organism Batch:

Ps12-04

Historical Mean IC25:

11.0 µg/L

Test Duration:

72 hours

Warning Limits (± 2SD):

4.3 - 28.6 µg/L

IC25 Growth:

 $11.5 \mu g/L$ 

Analyst(s):

Test Vessel:

VC/SM

Date Collected:

Time Collected:

Date Received:

Time Received:

Date Tested:

95% Confidence Limits: 7.9 - 15.5 µg/L

The reference toxicity test was performed under the same experimental conditions as those used with the test sample.

**TEST CONDITIONS** 

Test Organism:

Pseudokirchneriella subcapitata

Control/Dilution Water:

Millipore Milli-O (no chemicals added)

Organism Batch:

Ps12-04

Volume per Replicate:

U-shaped polystyrene microplate

Strain Number:

UTEX (1648)

220 µL

Source:

In-house culture

Number of Control Replicates:

10

Culture Origin:

University of Texas, Austin TX, USA Number of Test Replicates:

Concentrations Tested:

10 + Control

Age (at start of test):

4 days (in exponential growth)

0.45 µm preconditioned filter

Photoperiod / Light Intensity:

pH Adjustment:

None

Mean Test Temperature (± SD):

Continuous light, 4196 - 4368 lux  $23.0^{\circ}C (\pm 0.0)$ 

Hardness Adjustment:

None

Test Duration:

Sample Pre-aeration: Sample Filtration:

None

Test Method Deviation(s):

72 hours None

4

Volume Filtered:

≥10 mL

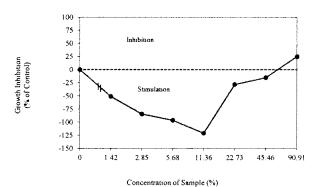
# **COMMENTS**

- •Statistical analysis could not be performed using non linear regression, since a suitable model could not be found. Therefore, test results were calculated using Linear Interpolation (CETIS)^a. In test concentrations where cell yield was stimulated (greater than the control), data were replaced with control values for the purposes of statistical analysis, as recommended by Environment Canada (2005).
- •All test validity criteria as specified in the test method cited above were satisfied.
- •No unusual appearance or treatment of culture prior to testing.
- •Algal growth curve is determined at least twice per year as required by the test method cited above.



Work Order: 221257 Sample Number: 33961

# Pseudokirchneriella subcapitata Growth Inhibition



### **CELL ENUMERATION AT 72-HOURS**

Sample pH (at 0 hours):

Initiation Date: 2012-04-24 Initiated By: VC/SM

Control pH (at 0 hours): 6.5

Completion/Enumeration Date: 2012-04-27

Control pH (at 72 hours): 6.5

Enumerated By: AS

Initial Cell Density at 0-h: 10636 cells/mL per microplate well

Enumeration Technique:

Manual (hemocytometer)

Inoculum Prepared: 00:45 h prior to test initiation

8.3

Control Cell Increase Factor: 28.9 times growth

### Cell Concentration (x 10000 cells/mL)

# Cell Yield (x 10000 cells/mL)

Concentration	n			Replicate					Mean	Standard	CV (%)	Stimulation
(%)	1	2	3	4	7	8	9	10		Deviation		(% of control)**
Control	35.0	29.0	29.5	37.5	28.0	34.5	25.5	26.5	29.62	4.40	14.85	_
0.18	_	_	_	_					_	_		
0.35	_	_							_	-	****	store
0.71		_	_	_					_	_	_	_
1.42	49.5	48.0	40.0						44.77	5.11	11.4	51.1
2.85	57.0	52.5	57.5	_		;			54.60	2.75	5.0	84.3
5.68	54.0	64.0	59.5						58.10	5.01	8.6	96.1
11.36	71.0	55.5 ¹	73.0						65.44	9.58	14.6	120.9
22.73	35.0	40.5	41.5						37.94	3.50	9.2	_
45.46	35.0	40.5	30.0						34.10	5.25	15.4	_
90.91	23.0	24.0	23.0	_					22.27	0.58	2.6	

- NOTES: **Significant stimulation compared to control, according to ANOVA Dunnett's Test (CETIS)a,  $\alpha$ =0.05.
  - •Control replicates 5 and 6 used for pH measurement.
  - •The Mann-Kendall test shows that there is no inhibitory gradient ( $\alpha$ =0.05).
  - •Negative cell yield indicates no growth.
  - Outlier according to Grubbs Test (CETIS)^a. Outlying data points were not excluded from statistical analysis, since they could not be attributed to error.

2012-04-27: Algal cells appear slightly deformed, yellow, and spherical in shape in the 45.46% and 90.91% test concentrations. AS

"-" = not enumerated/not required

Date: 2012 -05

# REFERENCES

Environment Canada. 2005. Guidance Document on Statistical Methods for Environmental Toxicity Tests. Environment Canada, Method Development and Application Section, Environmental Technology Centre, Environmental Protection Service. Ottawa, Ontario. EPS 1/RM/46, December 2005.

Approved By:

^a CETIS, © 2001-2007. Comprehensive Environmental Toxicity Information System. Tidepool Scientific Software, McKinleyville CA 95519. [Program on disk and printed User's Guide].



### AquaTox Testing & Consulting Inc.

11B Nicholas Beaver Rd. Guelph ON N1H 6H9

Tel: (519) 763-4412 Fax: (519) 763-4419

# Pseudokirchneriella subcapitata

Date Collected:

Time Collected:

Date Received: Time Received:

Date Tested:

Growth Inhibition 1 of 2

2012-04-23

2012-04-25

15:50 2012-04-24

10:30

Sample Number: 33962

SGS Lakefield Research Limited Company:

221257

Location: Lakefield ON Substance: Lake Water Sampling Method: Grab

Sampled By: BB 13.0°C Temp. on arrival:

Work Order:

Sample Description: Cloudy, orange, odourless

Test Method: Growth Inhibition Test Using a Freshwater Alga. Environment Canada, Conservation and

Protection. Ottawa, Ontario. Report EPS 1/RM/25, 2nd ed. (March 2007).

SAMPLE IDENTIFICATION

TEST RESULTS

Statistical Method Effect Value 95% Confidence Limits

IC25 (Growth) >90.91%

The results reported relate only to the sample tested.

ZINC (AS ZINC SULPHATE) REFERENCE TOXICANT DATA

Nonlinear Regression (CETIS)^a 2012-04-24 Statistical Method: Date Tested:

Historical Mean IC25: 11.0 µg/L Organism Batch: Ps12-04 Warning Limits ( $\pm$  2SD):  $4.3 - 28.6 \,\mu g/L$ Test Duration: 72 hours VC/SM IC25 Growth:

 $11.5 \mu g/L$ Analyst(s):

95% Confidence Limits: 7.9 - 15.5 μg/L

The reference toxicity test was performed under the same experimental conditions as those used with the test sample.

# **TEST CONDITIONS**

Control/Dilution Water: Millipore Milli-Q (no chemicals added) Pseudokirchneriella subcapitata Test Organism:

Organism Batch: Ps12-04 Test Vessel: U-shaped polystyrene microplate

220 µL Strain Number: UTEX (1648) Volume per Replicate: Number of Control Replicates: 10 Source: In-house culture 4 University of Texas, Austin TX, USA Number of Test Replicates: Culture Origin:

Concentrations Tested: 10 + Control 5 days (in exponential growth) Age (at start of test):

Continuous light, 4196 - 4368 lux Photoperiod / Light Intensity: pH Adjustment: None

None Mean Test Temperature (± SD):  $23.0^{\circ}C (\pm 0.0)$ Hardness Adjustment: 72 hours Sample Pre-aeration: None Test Duration:

Sample Filtration: 0.45 µm preconditioned filter Test Method Deviation(s): None

Volume Filtered: ≥10 mL

## **COMMENTS**

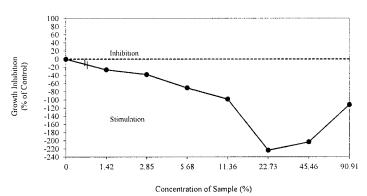
- •All test validity criteria as specified in the test method cited above were satisfied.
- •No unusual appearance or treatment of culture prior to testing.
- •Algal growth curve is determined at least twice per year as required by the test method cited above.

2 of 2

Work Order: Sample Number:

221257 33962

# Pseudokirchneriella subcapitata Growth Inhibition



### **CELL ENUMERATION AT 72-HOURS**

Initiation Date:

2012-04-25

Sample pH (at 0 hours):

8.5

Initiated By:

AS

Control pH (at 0 hours):

6.5

Completion/Enumeration Date: 2012-04-28

Control pH (at 72 hours):

6.5 10864 cells/mL per microplate well

Enumerated By:

CL

Initial Cell Density at 0-h: Inoculum Prepared:

00:50 h prior to test initiation

Enumeration Technique: Control Cell Increase Factor:

27.7 times growth

Manual (hemocytometer)

# Cell Concentration (x 10000 cells/mL)

# Cell Yield (x 10000 cells/mL)

Concentration	n			Replicate					Mean	Standard	CV (%)	Stimulation
(%)	1	2	3	4	7	8	9	10		Deviation		(% of control)**
Control	30.0	29.5	29.0	32.5	29.5	34.5	25.5	30.0	28.98	2.62	9.06	_
0.18	_	_		person.					_		_	Name of the last o
0.35	_	_	_						_	_	_	_
0.71	_	_	_	nom.					_		_	
1.42	34.0	42.0	37.0	-					36.58	4.04	11.0	_
2.85	35.5	48.5	39.0	estro.					39.91	6.73	16.9	37.7
5.68	53.0	48.5	50.0						49.41	2.29	4.6	70.5
11.36	62.5	59.5	53.5	-					57.41	4.58	8.0	98.1
22.73	98.5	91.0	95.0						93.75	3.75	4.0	223.5
45.46	86.0	81.5	99.5						87.91	9.37	10.7	203.4
90.91	56.5	62.5	68.5						61.41	6.00	9.8	111.9

**Significant stimulation compared to control, according to ANOVA - Dunnett's Test (CETIS)^a,  $\alpha$ =0.05. NOTES:

- •Control replicates 5 and 6 used for pH measurement.
- •The Mann-Kendall test shows that there is no inhibitory gradient ( $\alpha$ =0.05).
- •No outlying data points were detected according to Grubbs Test (CETIS)^a.

"-" = not enumerated/not required

Data Reviewed By:

# REFERENCES

^a CETIS, © 2001-2007. Comprehensive Environmental Toxicity Information System. Tidepool Scientific Software, McKinleyville CA 95519. [Program on disk and printed User's Guide].

Approved By:



# **Attachment 2**

### EBA Engineering Consultants Ltd. operating as EBA, A Tetra Tech Company

Oceanic Plaza, 9th Floor, 1066 West Hastings Street Vancouver, BC V6E 3X2 CANADA p. 604.685.0275 f. 604.684.6241

# TECHNICAL MEMO

**ISSUED FOR USE** 

TO: Rick Hoos DATE: April 18, 2012

C: MEMO NO.:

FROM: David Morantz EBA FILE: V15101007.004

**SUBJECT:** Background Water Quality Assessment – Avalon Thor Lake Project

# **Water Quality Summary**

The table below provides a summary of the baseline water quality data for total metals in Drizzle, Murky, and Thor lakes. For comparison, the table also shows laboratory detection limits, CCME guideline levels, and the modelled values of chemical parameters in each of the lakes due to discharges from the Tailings Management Facility (TMF) for year 20 following mine startup. The year-20 value represents a worst case situation since modelled values for years one to twenty progressively increase with time due to the recycling and subsequent concentration of metals in the recycle stream. All data have been summarized from Appendix F in the Stantec (2011) Environmental Baseline Report, Volume 3-Aquatics and Fisheries.

Mean and standard deviation values in the table are based on a summary of analysis results from the following sampling periods:

Drizzle Lake: September 2009; April 2010; June 2010; September 2010; October 2010.

**Murky Lake**: March 2008; October 2008; March 2009; June 2009; September 2009; April 2010; June 2010; September 2010; October 2010.

**Thor Lake**: March 2008; October 2008; March 2009; June 2009; September 2009; April 2010; June 2010; September 2010; October 2010.

# **Comments**

The water quality summary indicates that mean background values of all parameters, except iron in Drizzle and Murky lakes, are less than CCME guideline levels. Some metals were consistently less than the laboratory detection limits. The high mean iron values in Drizzle and Murky lakes are skewed due to very high concentrations of this metal found in samples collected under the ice in March and April (as indicated by a standard deviation that is more than twice the mean). In all cases, iron values are less than CCME guideline levels during open water periods. This can be explained by the fact that iron is released from the sediments under anoxic conditions, which exist under the ice in Drizzle and Murky lakes. Iron levels in Thor Lake did not exceed CCME guideline levels at any time during the study period.

The predicted concentrations of metals in each of the lakes during mine operation, under worst case conditions, are consistently within background levels. Based on modelling results, the operation of the mine is therefore not anticipated to affect significantly the metals concentrations in any of the lakes.



# Mean and standard deviationⁱ of selected baseline water quality parameters and predicted (modelled) values resulting from discharges from the Tailings Management Area in Year 20.

	00145	Datastian	I	Orizzle Lake		ı	Murky Lak	е		Thor Lak	ке
Parameter	CCME Guideline	Detection Limit ⁱⁱ	Mean	S.D.	Modelled Value	Mean	S.D.	Modelled Value	Mean ⁱⁱⁱ	S.D.	Modelled Value
Aluminum	100	5.0	8.30	9.10	10.0	7.20	3.90	8.53	3.30	0.54	3.77
Arsenic	5.0	0.1	0.92	0.23	0.93	1.29	0.51	1.29	0.77	0.06	0.77
Cadmium	0.052	0.017	0.01	0.002	0.01	0.02	0.01	0.02	0.02	0.01	0.02
Chromium	8.9	0.5	0.25	0	0.25	0.25	0	0.25	0.28	0.02	0.28
Copper	2-4	0.3	0.25	0.09	0.26	0.36	0.13	0.36	0.36	0.15	0.36
Iron	300	30	1091	2322	1093	3054	4948	3055	69.50	25.40	69.93
Lead	1-7	0.05	0.028	0.01	0.030	0.03	0.01	0.031	0.05	0.05	0.050
Mercury	0.026	0.05	0.005	0	0.005	0.01	0.01	0.01	0.01	0	0.01
Molybdenum	73	0.05	1.27	0.48	1.40	1.42	0.53	1.52	2.10	0.03	2.14
Nickel	25-150	0.5	0.25	0	0.27	0.25	0	0.27	0.25	0	0.26
Selenium	1.0	1.0	0.50	0	0.50	0.50	0	0.50	0.50	0	0.50
Silver	0.1	0.01	0.01	0.01	0.01	0.01	0.02	0.01	0.01	0	0.01
Thallium	0.8	0.1	0.05	0	0.05	0.05	0	0.05	0.05	0	0.05
Zinc	30	1.0	0.90	0.60	0.92	2.30	1.10	2.32	1.43	0.50	1.44

i All concentration values in μg/L.

Where sample values were less than the detection limit, half the value of the detection limit was used for computations; standard deviation values of 0 imply that all measurements were less than the detection level.

Mean of mean values for four sampling locations in Thor Lake.



# **Attachment 3**

### EBA Engineering Consultants Ltd. operating as EBA, A Tetra Tech Company

MEETING NOTES

Oceanic Plaza, 9th Floor, 1066 West Hastings Street Vancouver, BC V6E 3X2 CANADA p. 604.685.0275 f. 604.684.6241

**ISSUED FOR USE** 

**ATTENDEES:** Nathen Richea, Paul Green – Aboriginal Affairs

DATE:

May 1, 2012

and Northern Development Canada Anne Wilson – Environment Canada

Rick Hoos-EBA David Morantz-EBA

**EBA FILE:** V15101007.004

**MEETING TITLE:** Avalon Thor Lake Site Specific Water Quality Objectives

The telecon meeting was held to discuss water quality issues related to the possible development of site specific water quality objectives (SSWQO) for the Avalon Rare Metals Inc. Thor Lake Project. The following summarizes the topics that were raised during the meeting:

- In advance of specific discussions related to water quality issues, EBA indicated that a complete copy of the Developer's Assessment Report (DAR) would be sent (preferably electronically) to Nathen and Anne, since the MVEIRB Registry website provides sections of the DAR as separate files. At the end of the call, EBA indicated that all necessary tables from various documents and IR responses would be provided as part of the meeting notes, for submission to the MVEIRB.
- Some questions were raised concerning tabular data that was recently sent in the April 18, 2012 memo to AANDC and EC, regarding baseline and modeled WQ information for Drizzle, Murky, and Thor lakes. In particular, it was noted that natural seasonal baseline iron levels were shown to be considerably higher than Canadian Council of Ministers of the Environment (CCME) guideline limits for the protection of aquatic life. Also, the reported iron concentrations were quite variable, which produced a very large standard deviation (twice the value of the mean).
  - EBA indicated that the mean values for iron were skewed due to very high concentrations found under the ice in winter, due to anoxic conditions. The data presented as an Appendix to the DAR showed that such high values only occurred in winter in shallow lakes, where anoxic conditions resulted in the mobilization of iron from the substrate. EBA further suggested that SSWQOs should not be related to natural extreme events, rather they should be related to more normal concentrations measured during the open water period.
- AANDC made an inquiry into what the worst case scenario presented in the Table provided in the April 18, 2012 memo represented EBA identified that the worst case referred to the predicted effluent quality after 20 years of operation. These values did not consider any influence of wet years or dry years. The values presented in the April 18 Table represented the expected water quality within the entire lake at Year 20.
- It was questioned whether the estimated discharge rate and quality (from the Tailings Management Facility (TMF), polishing pond, or processing plant) shown in the DAR are likely to change. In response, EBA indicated that the design of the TMF and the operation were continuing to be



optimized in the current feasibility study and that such refinements are anticipated to result in further improvements, in particular possible reductions in the decant discharge volumes.

- There was discussion concerning nutrient (nitrate, phosphorous, etc.) discharges from the TMF, which were not included in the recent (April 18) table of baseline and modeled water quality parameters that was sent to AANDC and EC. It was indicated that information regarding nutrient concentrations in the effluent has been provided in responses to specific AANDC and EC Information Requests, which indicated that phosphorous levels in treated sewage discharged to the TMF will be very low due to the state-of-the-art treatment facility that will be used at the mine site. However, it was agreed that a summary of information concerning phosphorous levels in the mine discharge will be provided by the end of April.
- There was a discussion about the lack of information on the background concentrations of Rare Earth Elements (REE). EBA indicated that further groundwater sampling is currently being conducted, which will provide information on background REE concentrations. The expected delivery date for this sampling event is the end of April, 2012. In addition, chronic toxicity analyses are presently being run using standard EC bioassay tests. The results, including characterization of the effluent used in the tests, will be provided to the MVEIRB, AANDC and EC when they become available (acute results are expected at the end of April). These results, including the chronic toxicity information, which is not expected for a few more weeks, will contribute to the consideration of site specific water quality objectives.
- Table 1 of the February 20 memo (Response to AANDC IR #2.1) presented data which indicated that Cerium and Lanthanum concentrations in the 5 Day Decant may be potentially toxic to Hyallela and Rainbow Trout (lanthanum only), based on a review of U.S. Environmental Protection Agency information. However, EBA indicated that these data are very conservative and reflect worst case conditions since they were based on toxicity tests using dissolved Rare Earth Metals (REE) in waters having low hardness concentrations (toxicity is inversely related to hardness; the hardness of water in the Thor Lake drainage is considerably higher than in water used for the reported toxicity tests).
  - Toxicity information for REEs are contained in a comprehensive document prepared by Wilfred Laurier University. EBA agreed to provide a digital copy of this document to AANDC and EC. It was noted by EBA that this report indicates that REEs generally have a low solubility and low toxicity levels. AANDC agreed to review this new document and get back to EBA is they had additional questions.
- Additional discussion occurred regarding chronic toxicity and its relationship to mixing zones. AANDC acknowledged that Avalon has suggested using Drizzle Lake as the mixing zone. It was noted that Drizzle Lake is very shallow and has not been shown to support fish. Modeled values for metals in Drizzle Lake in year 20 of mine operation are considerably lower than the existing CCME guidelines, except for iron in winter. These predicted concentrations may form the basis for discussions regarding the setting of SSWQOs for this project. There was some discussion whether CCME guideline levels are appropriate as SSWQOs. AANDC noted CCMEs non-degradation approach and pollution prevention principles in the recent Land and Water Board Water and Effluent Quality Policy. Environment Canada noted that CCME guidelines do not exist for the REEs.

- AANDC indicated that the outlet of Drizzle Lake may be the appropriate location for establishing the
  discharge point for monitoring of SSWQOs. It was noted that the water quality model indicates that
  effluent will be of relatively high quality.
- Discussion then moved to Environmental Effects Monitoring (EEM), which is mandated by the Metal Mining Effluent Regulations (MMER). AANDC noted that EEM monitoring would be required as a minimum but that an even more rigorous monitoring program may be required per its AEMP Guidelines.
- Although AANDC has noted that Avalon has requested to use CCME Guidelines as the primary basis for establishing SSWQOs at the outlet of Drizzle Lake, it was concerned that CCME Water Quality Guidelines do not exist for all parameters of concern. AANDC also indicated that given the predicted high quality of the decant discharge from the Nechalacho Tailings Management Facility to the downstream receiving environment, the adoption of CCME criteria for the establishment of SSWQOs for this project could be a good initial starting point for further discussions.
- Discussions concluded with a consensus that the meeting was productive, and that future meetings to resolve issues related to the setting of SSWQOs for other parameters for which CCME criteria do not yet exist would be valuable, particularly when the results of REEE and acute toxicity were available (end of April).

Attached to this Technical Memo are copies of the Tables of predicted effluent/downstream water quality that were previously provided in response to various AANDC and EC Information Requests.

# Table 1 is excerpted from the February 2012 Avalon response to AANDC Information Request #2:

	SON OF PREDICTED RAR DATA – THOR LAKE	E EARTH ELEMENT	CONCENTRATION	S WITH AVAILABLE
Element	Day 5 Decant Concentration in Tailings Discharge (mg/L)	Max. Predicted Concentration Years 1-20 (mg/L)	LC 50 Concentration (mg/L)	Aquatic Organism Affected
Cerium (Ce)	1.39E-1	1E-4	0.032	Hyalella Azteca (amphipod crustacean)
Dysprosium (Dy)	2.52E-3	1.9E-6	0.162	Hyalella Azteca
Erbium (Er)	5.81E-4	4.4E-7	0.191	Hyalella Azteca
Europium (Eu)	1.09E-3	8.3E-7	0.112	Hyalella Azteca
Gallium (Ga)	2.86E-3	2.2E-6	>1.0	Hyalella Azteca
Gadolinium (Gd)	9.37E-3	7.1E-6	0.150	Hyalella Azteca
Holmium (Ho)	3.12E-4	2.4E-7	n/a	
Lanthanum (La)	6.88E-2	5.2E-5	0.018 0.020	Hyalella Azteca Oncorhynchus mykiss (Rainbow trout)
Lutetium (Lu)	3.3E-5	2.5E-8	0.029	Hyalella Azteca
Niobium (Nb)	2.57E-3	1.9E-6	0.026	Hyalella Azteca
Neodymium (Nd)	6.16E-2	4.7E-5	0.143	Hyalella Azteca
Praseodymium (Pr)	1.73E-2	1.3E-5	0.035	Hyalella Azteca
Scandium (Sc)	3.39E-3	2.6E-6	0.029	Hyalella Azteca
Samarium (Sm)	1.10E-2	8.3E-6	0.074	Hyalella Azteca
Tantalum (Ta)	2.30E-4	1.7E-7	0.002	Hyalella Azteca
Terbium (Tb)	8.19E-4	6.2E-7	0.084	Hyalella Azteca
Thulium (Tm)	4.6E-5	3.5E-8	n/a	
Ytterbium (Yb)	3.24E-4	2.5E-7	0.069	Hyalella Azteca
Zirconium (Zr)	3.29E-3	2.5E-6	>1.0	Hyalella Azteca

Table 3 is excerpted from the February 2012 Avalon response to AANDC IR #6.2/3:

	TABLE 3: MAXIMUM CONCENTRATION IN THE THOR LAKE SYSTEM AND WATER QUALITY GUIDELINES FOR THE PARAMETERS OF CONCERN												
	Backgr	ound Conce	ntration	Thor Murky		Murky Drizzle CCME	CCME	MMER					
Species	Thor Lake	Murky Lake	Drizzle Lake	Lake	Lake	Lake	Water Quality Guideline	Effluent Criteria					
TDS (mg/L)	191	224	169	191.3	224.9	170.1	-	-					
Cl (mg/L)	4.35	5.30	3.60	4.39	5.39	3.72	-	-					
SO ₄ (mg/L)	0.28	0.30	< 0.5	0.351	0.514	< 0.771	-	-					
F (mg/L)	1.11	1.04	0.893	1.11	1.05	0.91	-	-					
NO ₃ * (as N mg/L)	0.053	0.014	-	0.053	0.014	0.014	2.9	-					
Ammonia (as N mg/L)	< 0.02	0.70	0.71	< 0.02	0.70	0.71	-	-					

The background concentration value for  $NO_3$  in Drizzle Lake was missing and assumed to have the same value as the background concentration in Murky Lake.

Table 20 is excerpted from the January 2012 Avalon response to the Environment Canada IR #13.1.

Tailings S	olids Com	ponent ²			Tailings Liquid Component ^{1,2}	PP1 TIs Decant Day 5 ²	Regulat	ions and Guidelines
Parameter	Units	PP1 TIs	Parameter	Units	PP1 TIs	PP1 Tis	MMER	CCME Guideline for the Protection of Aquatic Life
SiO ₂	%	60.2	Initial pH	units	9.28			
Al ₂ O ₃	%	13.2	Final pH	units	8.81			
Fe ₂ O ₃	%	10.7	Radionucl	ide Analyses				
MgO	%	2.43	226Ra	Bq/L	<0.01	< 0.01	0.37	
CaO	%	0.85	228Ra	Bq/L	< 0.3	0.3		
Na ₂ O	%	3.35	210Pb	Bq/L	<0.1	<0.1		
K ₂ O	%	6.05	General and	Metals Analyses				
TiO ₂	%	0.03	рН	units	7.95	8.20		6.5-9.0
P2O ₅	%	0.04	Alkalinity	mg/L as CaCO ₃		119		
MnO	%	0.09	EMF	mV		284		
Cr ₂ O ₃	%	<0.01	Conductivity	μS/cm		617		
$V2O_5$	%	< 0.01	TDS	mg/L		400		
LOl	%	1.54	TSS	mg/L		14		
Sum	%	98.5	Cl	mg/L	3.6	44		
$Nb_2O_5$	%	0.18	$SO_4$	mg/L		100		
ZrO ₂	%	1.52	F	mg/L	1.83	4.43		
			TOC	mg/L		12.2		
			Hg	mg/L	< 0.0001	< 0.0001		0.000026
			As	mg/L	0.0199	0.0022	0.5	0.005
			Ca	mg/L	21.0	43.7		
			Cu	mg/L	0.0010	0.0023	0.30	Minimum ³ 0.002

# Table 20 continued...

Parameter	Units	Tailings Liquid Component ^{1,2}	PP1 TIs Decant Day 5 ²	Regulation s and Guidelines	Parameter
		PP1 TIs	PP1 TIs	MMER	CCME Guideline for the Protection of Aquatic Life
Fe	mg/L	0.041	0.570		0.3
K	mg/L	8.76	28.8		
Mg	mg/L	3.20	9.14		
Mn	mg/L		0.0788		
Na	mg/L	13.4	70.4		
Ni	mg/L	0.0059	0.0070	0.50	Minimum ³ 0.025
Pb	mg/L		0.00060	0.20	Minimum ³ 0.001
Se	mg/L	< 0.001	< 0.001		0.001
Si	mg/L	4.72	8.10		
Th	mg/L	0.000832	0.000694		
U	mg/L	0.00535	0.00880		0.015
Y	mg/L		0.00877		
Zn	mg/L	0.003	0.007	0.50	0.03

^{1.} CCME guidelines are not typically applied to tailings facilities.

^{2.} Source: SGS Minerals Services. August 30, 2011. Environmental Characterisation of Ore, Concentrate and Tailings from the Nechalacho Rare Earth Element Project – Phase #2. Prepared for Avalon Rare Metals Inc. (Project 11806-007) (Tables 10, 20, and 24)

^{3.} The minimum guidelines have been expressed in this table. The CWQG copper and lead equations determine specific guidelines based on water hardness. [copper concentration guidelines =  $e^{0.8545[ln(hardness)]-1.465} * 0.2 \mu g/L$ ; nickel concentration guidelines =  $e^{0.76[ln(hardness)]+1.06} \mu g/L$ .] [lead concentration guidelines =  $e^{0.76[ln(hardness)]+1.06} \mu g/L$ .]

# The following table is excerpted from the April 18, 2012 memo addressed to Anne Wilson and Nathen Richea.

Mean and standard deviationⁱ of selected baseline water quality parameters and predicted (modelled) values resulting from discharges from the Tailings Management Area in Year 20.

Parameter	CCME Guideline	Detection Limit ⁱⁱ		Drizzle Lake			Murky Lake			Thor Lake	
			Mean	S.D.	Modelled Value	Mean	S.D.	Modelled Value	Mean ⁱⁱⁱ	S.D.	Modelled Value
Aluminum	100	5.0	8.30	9.10	10.0	7.20	3.90	8.53	3.30	0.54	3.77
Arsenic	5.0	0.1	0.92	0.23	0.93	1.29	0.51	1.29	0.77	0.06	0.77
Cadmium	0.052	0.017	0.01	0.002	0.01	0.02	0.01	0.02	0.02	0.01	0.02
Chromium	8.9	0.5	0.25	0	0.25	0.25	0	0.25	0.28	0.02	0.28
Copper	2-4	0.3	0.25	0.09	0.26	0.36	0.13	0.36	0.36	0.15	0.36
Iron	300	30	1091	2322	1093	3054	4948	3055	69.50	25.40	69.93
Lead	1-7	0.05	0.028	0.01	0.030	0.03	0.01	0.031	0.05	0.05	0.050
Mercury	0.026	0.05	0.005	0	0.005	0.01	0.01	0.01	0.01	0	0.01
Molybdenum	73	0.05	1.27	0.48	1.40	1.42	0.53	1.52	2.10	0.03	2.14
Nickel	25-150	0.5	0.25	0	0.27	0.25	0	0.27	0.25	0	0.26
Selenium	1.0	1.0	0.50	0	0.50	0.50	0	0.50	0.50	0	0.50
Silver	0.1	0.01	0.01	0.01	0.01	0.01	0.02	0.01	0.01	0	0.01
Thallium	0.8	0.1	0.05	0	0.05	0.05	0	0.05	0.05	0	0.05
Zinc	30	1.0	0.90	0.60	0.92	2.30	1.10	2.32	1.43	0.50	1.44

i All concentration values in μg/L.

Where sample values were less than the detection limit, half the value of the detection limit was used for computations; standard deviation values of 0 imply that all measurements were less than the detection level.

iii Mean of mean values for four sampling locations in Thor Lake.



# **Attachment 4**

# Review of Aquatic Effects of Lanthanides and Other Uncommon Elements

**Final Project Report** 

March 31, 2011

# Prepared for the

EC Contribution Agreement with the CNTC for 2010/2011

Tania Ng, D. Scott Smith, Anthony Straus and James C. McGeer

Wilfrid Laurier University

# **Table of Contents**

Summary3
Background and Overall Objectives:
Progress Against Milestones for February 1, 2010 - March 31, 2011:4
Sources of Lanthanides5
Concentrations in the Environment6
Solubility & Speciation7
Potential Precipitates in Toxicity Tests
Geochemical Speciation9
Bioavailability9
Biological Effects on Aquatic Organisms 10
i). Acute Toxicity
ii). Chronic Toxicity
Bioaccumulation by Aquatic Organisms
Data Gaps & Recommendations for Toxicity Testing15
i). Test Elements
ii). Test Systems
iii). Test Species
iv). Test Methods 16
v). Test Topics
References
Appendix A

# **Project Final Report**

Reporting Period: Feb. 7 – Mar. 31, 2011

# **Summary**

**Project Title:** Review of Aquatic Effects of Lanthanides & Other Uncommon Elements

**Participants:** Dr. Jim McGeer (PI), Department of Biology, Wilfrid Laurier University, Waterloo. Email: jmcgeer@wlu.ca. Phone: 519 884-0710 ext. 3537.

Dr. Scott Smith (co-PI), Department of Chemistry, Wilfrid Laurier University, Waterloo. Email: ssmith@wlu.ca. Phone: 519 884-0710 ext. 3046.

Dr. Tania Ng (Investigator), Department of Biology, Wilfrid Laurier University, Waterloo. Email: ngtania2002@yahoo.com. Phone: 519 884-0710.

Anthony Straus (Research Assistant), Department of Biology, Wilfrid Laurier University, Waterloo. Email: anthonystraus@gmail.com. Phone: 519 884-0710.

	MILESTONE DESCRIPTIONS	MILESTONES PROGRESS		
1.	Bibliographic listing of primary or secondary literature on lanthanides to March 31, 2011, with annotations.	Completed. In year 1, Dr. Gheorghiu has searched a list of literature articles on lanthanides. In year 2, Dr. Ng added "bioavailability" to the search items and included secondary literature articles on the list. A total of 6359 papers were found from the search. Of these, 629 papers, with annotations (complete author listing, abstracts, source) were retained for review. Search strategies and results are summarized in Table 1. The complete reference list, in annotated form, is also attached.		
2.	Establish an EndNote data-base on the ecotoxicology of lanthanides.	Completed. A comprehensive database comprising 629 peer-reviewed papers and some government reports has been compiled. Articles were categorized into groups using keywords: bioaccumulation, bioavailability, birds, fish, invertebrates, mammals, plants, solubility, speciation, toxicity. The EndNote library is attached as a separate file.		
3.	Full review of the lanthanides literature, evaluation of data, identification of data gaps, and recommendations for aquatic ecotoxicity testing program.	Completed. In year 2, articles from the EndNote database were critically reviewed. Data are collected for evaluation on sources (Table 2); concentrations in the environment (Fig. 1); speciation & solubility (Table 3, Fig 2,3); geochemical speciation; bioavailability; toxicity (Table 4 – 6) and bioaccumulation of lanthanides (Table 7 - 8).		

# Final Report: Review of Aquatic Effects of Lanthanides & Other Uncommon Elements

# **Background and Overall Objectives:**

The lanthanides (also called Rare Earth Elements – REE, or Lanthanoids) are the chemical elements found in Row 6 of the periodic table between Groups 3 and 4, beginning with lanthanum (La), which accounts for the group name. The lanthanides consist of the following metals: lanthanum (La), cerium (Ce), praseodymium (Pr), neodymium (Nd), promethium (Pm), samarium (Sm), europium (Eu), gadolinium (Gd), terbium (Tb), dysprosium (Dy), holmium (Ho), erbium (Er), thulium (Tm), ytterbium (Yb), and lutetium (Lu).

The present knowledge concerning the eco-toxicological effects of lanthanides is limited, with few studies having addressed the biological effects of the lanthanides on aquatic plants and animals. The **overall objective** of this project is to collect, collate and review the existing literature on the eco-toxicity of the lanthanide elements. The compiled a database on the toxicology of the lanthanides was reviewed with the goal of identifying gaps in our understanding on the chemistry, bioavailability, and toxicity potential of the lanthanides to aquatic organisms.

# Progress Against Milestones for February 1, 2010 - March 31, 2011:

A database on the lanthanides (Milestone 1) was established by conducting literature searches using the PubMed [(US National Library of Medicine (NLM), National Institute of Health (NIH)], ISI Web of Science (Thomson Reuters) and TOXNET (NLM-NIH) databases. Searches on the lanthanides in general, and the individual lanthanides, were conducted using the following keywords: rare earth elements, lanthanides, cerium, praseodymium, neodymium, promethium, samarium, europium, gadolinium, terbium, dysprosium, holmium, erbium, thulium, ytterbium, and lutetium. Searches were then narrowed using these keywords in combination with the additional terms: ecotoxicity, environmental toxicity, toxicity, speciation, bioaccumulation, fish, invertebrates, plants. This search was completed in phase 1 of the project which was led by

Wilkie et al. and delivered in March 2010. This search was updated at the beginning of phase 2 (February 2011). Since "bioavailability" is a key topic area of interest, it was added as one of the search keywords in year 2. Web of Science was only used to run this search because based on previous search experience it provides the maximum number of relevant articles in aquatic science. Results for each search combination are presented in Table 1. From these searches, there were a total of 6,359 "hits" requiring further screening, which led to the identification of 862 articles. These were assessed for availability and retained for further screening. After further screening, and elimination of duplicates, a total of 577 peer-reviewed published articles and government reports plus 52 additional articles (primary and secondary – not accessible or written in other languages) cited in literature, resulting to a total of 629 articles entered into an EndNote database (Milestone 2; library attached). The smart group feature of EndNote was used to generate subfolders for the following topics: speciation, bioaccumulation, bioavailability, birds, fish, invertebrates, mammals, plants, solubility, speciation, toxicity. A complete annotated reference list (authors, year, abstracts, journal) of the articles in the EndNote database is attached with the report. Articles where an electronic version (pdfs) was available were used for a critical review of: sources; concentrations in the environment; solubility, speciation and partitioning; bioavailability; toxicity; and bioaccumulation of lanthanides (Milestone 3; summary of evaluation provided below).

# **Sources of Lanthanides**

Contrary to their designation as rare earth elements, the lanthanides are relatively abundant in the Earth's crust (except for promethium, a radioactive artificial element which is a byproduct of spontaneous uranium decay). REEs are usually found together in minerals (Table 2) and about 160 mineral ores are known to contain REE at levels of up to 60 % (Slooff et al., 1993). Approximately 80 % of known REE mineral supplies/desposits are found in China (Annema, 1990). Highly pure, low-cost rare earths can be industrially produced mainly through ion-exchange and solvent extraction processes. Lanthanides are used in different industries, most commonly as petroleum refining catalysts or catalytic converters, as permanent magnets; in glass polishing and ceramics, the production of sunglass lenses; laser industries; as chemical fertilizers, or trace supplements in agriculture; as misch metal to remove oxygen and to enrich steel; as luminophores; and as high-temperature superconductors (Palasz and Czekaj, 2000). In addition, radioisotopes of lanthanides are used in the anticancer therapy (Alberts et al., 1997).

### **Concentrations in the Environment**

Due to its co-existence with other minerals, natural concentration of lanthanides are similar to essential elements such as iodine, cobalt and selenium (Palasz and Czekaj, 2000). Compared to the non-essential metals, concentrations of La, Ce and Nd tend to be higher than Pb and even concentrations of the less abundant REEs, (e.g. Tm and Lu), are typically higher than Cd in the earth crust (Weltje, 2002). An exception is Pm, a by-product from decay of uranium and not naturally present in the earth crust. Natural concentrations of other REEs in the earth crust display a typical saw-tooth pattern with a log-linear decrease of concentration with atomic number (Z). Additionally, the odd-numbered elements have lower concentrations than the even-numbered elements (Weltje, 2002) (Fig. 1). In general, the highest concentration is found for Ce, the first even-numbered lanthanide, while the lowest concentrations are found for the last two odd-numbered lanthanides, Tm and Lu, whose concentrations are about two orders of magnitude lower than those of Ce.

REEs have low solubility (see "Solubility &, Speciation" section) and dissolved (passing  $\leq$  0.45  $\mu$ m filter) Ce and Lu only ranged from 2.9 – 714 ng L⁻¹ and 0.04 – 7 ng L⁻¹ respectively in the pristine freshwater environments (Weltje, 2002). Dissolved concentrations of other REEs are between these ranges. Pore water often has slightly higher dissolved REEs than surface water due to its lower water pH that increases solubility (Weltje et al., 2002b). In contrast to dissolved REEs, concentrations in sediment are much higher (Sneller et al., 2000), reflecting the high  $K_p$  sediment-water partition coefficient (see "Geochemical Speciation" section). For example, concentrations of Ce, La, Pr and Nd in unpolluted sediment of Tsurumin River, Japan are 45.1, 22.3, 6.2 and 21.1 mg kg⁻¹ respectively (Mohiuddin et al., 2010). Due to the low solubility of many inorganic lanthanide salts, the high affinity for sediment and suspended matter, and their tendency to form complexes with both inorganic and organic ligands, the free ion concentrations of lanthanides in natural freshwaters is very low (pM to nM range). Compared to freshwater sediments (0.3 – 68.9 mg kg⁻¹), background concentration of REE elements in the sediment of saltwater environments (0.3 – 92.6 mg kg⁻¹) tend to be higher (Sneller et al., 2000).

The main anthropogenic emissions of REEs are to surface water (Slooff et al., 1993) and subsequently the majority of lanthanides end up in sediment. Depositions associated with

phosphate fertilizer and discharges from catalyst producing industries have resulted in contaminated sediment in the Rhine R. estuary (Netherlands) where concentrations reach up to 170 mg kg⁻¹ Ce, 80 mg kg⁻¹ La, 80 mg kg⁻¹ Nd, 30 mg kg⁻¹ Pr and 20 mg kg⁻¹ Sm (Sneller et al., 2000). The lanthanum-modified clay Phoslock®, which is used to remove phosphate from wastewater in Australia, results in total La concentrations as high as 400 μg L⁻¹ after one application (Flapper, 2003; McIntosh, 2007) with dissolved lanthanum concentrations being up to 12 μg L⁻¹ after several applications over 24-48 h after which it then dropped rapidly to below 1 μg L⁻¹ (Haghseresht, 2006).

# **Solubility & Speciation**

Generally, solubility of REEs is low due to complex formation and the low solubility product (K_{sp}) of the complex. Dominant species of inorganic and organic lanthanide complexes in solution are given in Table 3. Often the simple fluorides, oxalates, phosphates, and carbonates are sparingly soluble while the complex carbonates, sulfates, and fluorides are readily soluble (Moeller and Vincenti, 1965; Cantrell and Byrne, 1987). Since carbonates, phosphates, hydroxides are ubiquitous ligands in aquatic environments they strongly limit aqueous lanthanide concentration. The K_{sp}'s of most complexes are very low e.g., K_{sp} of REE-phosphates can be as low as  $10^{-25} \, \text{mol}^2 \, \text{l}^{-2}$  (Liu and Byrne, 1997);  $K_{sp}$  of REE-carbonates and hydroxides are the lowest as 10⁻³⁰ mol⁵ l⁻⁵ and 10⁻²⁴ mol⁴ l⁻⁴ at 25 °C respectively (Martell and Smith, 1997-2001). However, solubility of REEs is also strongly dependent on pH and temperature. Increase in temperature increases solubility. An increase in pH can result in precipitation of REEs as hydroxides or carbonates, exchange of REEs for H⁺ on accessible mineral exchange sites, or adsorption of the REEs onto the surfaces of minerals (Humphris, 1984). Each of these processes reduces the dissolved REE concentration. Generally, REEs can be divided into 2 groups, the first of which is composed of more soluble elements (La-Gd), and the second group consists of the less soluble elements (Tb-Lu) (Sneller et al., 2000).

Speciation of REEs which determines solubility and bioavailability, depend on pH, salinity and the presence of anions (Sneller et al., 2000). As a consequence, speciation of REEs in salt water is different from fresh water. In salt water, between 70 and 96 % of the REEs is present as carbonate complexes, depending on the dissolved organic carbon (DOC) level, whereas in freshwater, humate complexes play a dominant role (Maas and Botterweg, 1993). At moderate to

high pH values, complexation of REEs with carbonate species dominates and that heavy REEs (Tb-Lu) form stronger complexes with carbonate ions than light REEs due to an increase in stability constants with atomic number (Cantrell and Byrne, 1987; Millero, 1992; Lee and Bryne, 1993). In addition, heavy REEs are generally more sensitive to pH alternations than light members due to their lower basicity (Tu et al., 1994). With reference to Table 3 and Brookins (1989), in acidic to neutral pH, the REE³⁺ ions may be the dominant form. At very low pH, REE³⁺ and (REE)SO₄⁺ are the most important. The halides are of minor importance even at very low pH. At neutral to slightly basic pH, the carbonate complexes become important with the light lanthanides (La-Gd) and one heavy lanthanide Tb, preferring REE(CO₃)⁺. The other heavy lanthanides preferring REE(CO₃)². As the pH approaches 10, hydrolysis products REE(OH)₂⁺, REE(OH)₃(aq), and REE(OH)₄ become important. Unlike other REEs, behaviour of Ce and Eu is different, due to their occurrence of more than one redox-states (Ce³⁺, Ce⁴⁺; Eu²⁺, Eu³⁺). Under normal oxic circumstances, REEs are present in the trivalent oxidation states (Cotton, 1991), but Ce³⁺ is oxidized to Ce⁴⁺, therefore forming CeO₂ which is insoluble in water. A different behaviour occurs for Eu in porewater in which Eu³⁺ is reduced to Eu²⁺, resulting to increasing dissolved concentrations in porewater (Weltje et al., 2002b).

# **Potential Precipitates in Toxicity Tests**

In natural systems lanthanides will possibly be undersaturated with respect to solid phases but at the higher concentrations often used for toxicity testing it is important to understand the potential for precipitation. It is well established that nominal and measured lanthanide concentrations do not necessarily match (see Test Methods section below) with measured values as much as a factor of two lower than nominal values. The likely solid phases in laboratory toxicity tests include hydroxide and/or carbonate phases. To investigate the likelihood of precipitation several equations were derived using National Institute of Standards and Technology (NIST) certified equilibrium constant values. The details of the equation derivations are given in Appendix A and the results of these calculations are shown for hydroxide in Figure 2 and for carbonate in Figure 3. Lanthanum hydroxide is most soluble at acidic pH and becomes less soluble as pH is increased. Alternatively, lanthanum carbonate goes through a solubility minimum around pH 8.2. The comparison acute toxicity values shown on Figure 2 demonstrate that for the measured La concentrations in toxicity tests (taken from Table 4) most studies were undersaturated with respect to La(OH)_{3(s)}. This is not the case for carbonate precipitation where most studies were

actually performed in the regime where La₂(CO₃)_{3(s)} is expected to precipitate. It should be noted that these calculations assumed a minimum amount of carbonate in that the only source of CO₂ was atmospheric. In general waters with even moderate alkalinity will have much greater carbonate concentrations and greater tendency to precipitate carbonate species. Thus, future toxicity testing on La elements should consider the possibility of carbonate precipitation and always utilize measured concentrations (see Test Methods below).

# **Geochemical Speciation**

REEs have relatively high affinity to sediment (Sneller et al., 2000). Maas and Botterweg (1993) reported that  $\log K_p$  (sm/w) of REEs in suspended matter and water in rivers is about  $3 \text{ l kg}^{-1}$ , where  $K_p$  is the partition or distribution coefficient. In oceans, this value is around 4, with the exception of Ce, for which the value is around 5 because of low solubility of the main form of Ce, CeO₂ in ocean, in which Ce is present as Ce⁴⁺, as compared to trivalent state of other REEs. Therefore large amount of REEs are bound to sediment or suspended matter in the aquatic environment. In addition,  $\log K_p$  (sediment to porewater) values for REEs are relatively higher compared to heavy metals such as Cu, Zn, Cd and Pb ( $\log K_p$  sed/pw:  $2.5 - 4 \text{ l kg}^{-1}$ ). In general, the variability in salinity, pH, composition of suspended matter, organic carbon content and Feand Mn-hydroxides, caused by the tidal movement, results in constantly changing equilibrium between sediment, pore water and surface water, and thus continuously changing REE concentrations in each partition (Bakkenist and Van De Wiel, 1995).

# **Bioavailability**

Since most of lanthanides (99 %) are present in or bound to suspended matter and sediment, they are considered largely unavailable. Only a minor fraction is dissolved in sediment pore or surface water (Weltje, 2002). Among the dissolved species, bioavailability decreases in the following order: free ion, inorganic complex, and finally organic complex (Sun et al., 1997). Changes in ligand or ion concentration in the environment can change bioavailability of REEs, for example, uptake of Gd by the marine algae *Ulva lactua* was strongly reduced by increasing carbonate concentrations (Stanley and Byrne, 1990). Uptake of La by the freshwater algae *Scenedesmus pannonicus* was reduced at a lower pH, also at a higher Ca concentration (Demon et al., 1988; 1989). In addition, Marang et al. (2008) has shown that competitive interactions with H⁺ inorganic species, and major cations (Ca²⁺, Mg²⁺) could influence Eu transport and

bioavailability. Due to the tremendous use of lanthanides in industrial processes (see "Sources of Lanthanides" section), methylated lanthanides may be produced during the synthesis processes. For example, use of methylation to increase water solubility of lanthanum complex in optical industry (Spyroulias et al., 1998; Syproulias et al., 1998); cerium triflate as alternative for oxidizing aromatic hydrocarbons to synthesize useful organic chemicals (Molander, 1992). Methylmercury is known to have a better bioavailability than mercury, but biovailability of methylated lanthanides to aquatic organisms is not yet studied.

Although sediment-bound lanthanide is usually considered not to be available, the bioavailability may change when the equilibrium is disturbed. For instance, when sediment particles are ingested by organisms, the lower pH in digestive system of the organisms may shift the equilibrium, thus increases concentration of dissolved lanthanides. Sediment-rooting plants may also exude protons or chelating substances (e.g., malate, citrate) to increase availability of another trivalent metal, the essential but poorly soluble iron (Mori, 1999). Therefore sediment may become a source of bioavailable lanthanide in water. Lanthanides present in biota may also become available (trophic transfer) to other organisms feeding on them. By using sequential extraction, bioavailability of lanthanides from the sediment can be estimated. In the sediment, among the water-soluble, acid-extractable and organic / sulphide bound species, water soluble species of La, Gd, Sm, Ce, Y has the strongest correlation with the accumulation in the algae *Chlorella vulgaris* (Sun et al., 1998). It suggests that the REE released from the sediment is an important factor affecting the bioavailability to the algae.

# **Biological Effects on Aquatic Organisms**

# i). Acute Toxicity

Data from literature about acute and chronic <u>t</u>oxicity of lanthanides to aquatic organisms are summarized in Table 4, 5 (freshwater) and Table 6 (saltwater). In general, most toxicity studies assessed effects of total/nominal lanthanides in the aquatic environments, fewer measured the dissolved concentrations, which is the most bioavailable form for biota. Due to the low solubility of lanthanides, it is not surprising that in freshwater (Table 4), the lower acute toxicity values (43 – 1232  $\mu$ g REE ·L⁻¹) are found for studies measuring dissolved lanthanide concentration and the higher is for studies using nominal lanthanide concentrations (450 – 4,069,767  $\mu$ g REE ·L⁻¹). There is an exception for the *Daphnia magna*, rainbow trout (*Oncorhynchus mykiss*) and the

shrimp, *Thamnocephalus platyurus* which have extremely high toxicity values (> 44,000 µg REE ·L⁻¹) of dissolved lanthanides. Watson-Leung (2009)'s study on *D. magna* and rainbow trout tested on La leached from Phoslock®, with unclear water chemistry of the leachate other than hardness and pH, therefore it is difficult to explain for their resistance to La, compared to toxicity from salts. In general, among different species of cladocerans (Daphnia magna, Daphnia carinata, Ceriodaphnia dubia), increasing hardness from 22 to 210 mg·L⁻¹ as CaCO₃ can reduce acute toxicity of lanthanides from 5 (total REEs) – 27 folds (dissolved REEs) (Table 4), and similar effect has been reported within D. carinata for La (Barry and Meehan, 2000). Freshwater algae, cladoceran and cnidarian have similar sensitivity to different REEs, with algae toxicity values from  $450 - 4{,}400 \,\mu g \, \text{REE} \cdot \text{L}^{-1}$ ; cladoceran toxicity values from  $43.2 - 24{,}000 \,\mu g \, \text{REE} \cdot \text{L}^{-1}$ and cnidarian toxicity values from  $44 - 4.400 \,\mu g \, REE \cdot L^{-1}$  in water of all hardness and all forms of REEs (dissolved / total). Toxicity of lanthanum to algae can be indirect associated with the removal of phosphate from the growth medium through the formation of LaPO₄ (Stauber, 2000). The zebrafish *Danio rerio* is ranked second for the sensitivity to lanthanides with acute toxicity values of total REEs from 19,000 – 25,000 in water hardness of 210 mg ⋅L⁻¹ as CaCO₃, and sensitivity of them is comparable to the high ends of cladocerans in the same water hardness (Table 4). There is not sufficient information to compare toxicity among the fish species when Watson-Leung (2009) is not considered. The shrimp, T. platyurus is the most resistant to lanthanide. Since toxicity information is the most obtained for cladocerans, comparison is made for toxicity among REEs within them. Acute toxicity of total REEs to cladocerans in the freshwater environment (water hardness: 210 mg ·L⁻¹ as CaCO₃) is ranked as Nd > Gd > Sm > Pr ~ Dy > Ce > La (Table 4). Acute toxicity of REEs on sediment dwelling organisms is less studied. The EC₅₀s (movement inhibition) of different REEs on the oligochaete *Tubifex tubifex* were similar and they averaged to 78.1 mM (Filipi et al., 2007). This article was written in other language, water chemistry and form of lanthanide (total / dissolved) are unclear, therefore it is not included in Table 4.

There is scarce data on toxicity of lanthanides to saltwater organisms, with limited information on acute toxicity only (Table 6). Some of the data available is from a secondary source and written in another language, making the evaluation on toxicity to saltwater organisms difficult. In general, all REEs (light REEs vs heavy REEs; odd-numbered REEs vs even-numbered REEs) have about the same extent of toxicity to the marine algae *Skeletonema costatum* (nominal toxicity values:  $4055 - 5009 \,\mu g \,REE \cdot L^{-1}$ ) (Tai et al., 2010), but in the copepod, *Acartia tonsa*,

toxicity of REEs is ranked as Ce > Sm > Gd > Nd > Pr > La > Dy and in the fish, *Poecilia reticulate*, toxicity is ranked as Pr > Nd > Sm > Gd > Ce > Dy > La (Table 6). Among the three saltwater species, copepod has the highest sensitivity to REEs, algae is the intermediate and the fish is the least sensitive.

Sneller et al. (2000) reported that, in general, acute toxicity of REEs is higher to marine organisms than freshwater organisms. From our literature data (Table 4, 6), algae in freshwater has a larger variation in responses (toxicity values:  $450 - 15,470 \,\mu g$  total REE ·L⁻¹) to acute exposure of REEs than in saltwater (toxicity values:  $4055 - 5009 \,\mu g$  total REE ·L⁻¹) which may be explained by the higher sensitivity of freshwater algae in water with low hardness. Saltwater crustacean (toxicity values:  $150 - 3,600 \,\mu g$  total REE ·L⁻¹) are more susceptible to lanthanides than freshwater crustacean (toxicity values:  $1,400 - 4,069,767 \,\mu g$  total REE ·L⁻¹) (Table 4, 6), although only one species of saltwater crustacean is compared. Data on different species of freshwater and saltwater fish are limited for evaluation.

# ii). Chronic Toxicity

Chronic toxicity of lanthanides to freshwater organisms show similar pattern as the acute toxicity (Table 5). Increasing hardness reduces chronic toxicity to the cladocerans and a less conclusive effect is reported for the amphipods. Water hardness increasing from 40-48 mg·L⁻¹ to 210 mg·L⁻¹ as CaCO₃ almost reduces toxicity of total lanthanides to the cladocerans by 5 fold (Table 5). When water hardness increased from 18 – 124 mg·L⁻¹ as CaCO₃, toxicity values increased from  $0.01 - 191 \mu g REE \cdot L^{-1}$  to  $278 - 1665 \mu g REE \cdot L^{-1}$  in *Hyallela azteca*. However, the latter range is a nominal value with no information on relative amount of dissolved REEs in the medium. Amphipod (toxicity values:  $0.01 - 191 \mu g$  dissolved REE ·L⁻¹, hardness at  $18 \text{ mg} \cdot \text{L}^{-1}$  as CaCO₃) and cladoceran (toxicity values:  $8.7 - 842 \mu g$  dissolved REE·L⁻¹, hardness at  $85 - 160 \text{ mg} \cdot \text{L}^{-1}$  as CaCO₃) have the highest sensitivity, the midge *Chironomus dilutus* is the intermediate (NOEC of total REE > 880  $\mu$ g·L⁻¹, hardness at 138 – 179 mg·L⁻¹ as CaCO₃) whereas the zebrafish *D. rerio* may have the lowest sensitivity (NOEC of total REEs  $> 2.600 \,\mathrm{ug}\cdot\mathrm{L}^{-1}$ , hardness at 210 mg·L⁻¹ as CaCO₃) (Table 5). Chronic toxicity (7 d LC₅₀) of each dissolved REE to *H. azteca* (Borgmann et al., 2005) is ranked as the follow: Tm > La > Nd > Lu > Ce > Pr > Sm > Tb > Yb > Eu > Ho >Gd > Dy > Er. Among all the chronic toxicity endpoints, brood size, length and weight of cladoceran neonates are more sensitive to lanthanides than survival and growth (Table 5). Main

route of lanthanum uptake by daphnia may be through the carapace because Ca is constantly taken up during moult cycle to harden carapace, but La is absorbed and binds to the Ca binding sites (Das et al., 1988), thus interfere moulting cycle (Barry and Meehan, 2000).

There are also studies on microorganisms and physiological impacts of REEs on aquatic animals. Gd at nominal concentrations of 15.7 mg·L⁻¹ and 47.2 mg·L⁻¹ affected the population size of bacteria E. coli and protozoan T. thermophila respectively in a microcosm after they were exposed for from 7 d to over 100 d (Fuma et al., 2001). Sm, Er and Ho can increase the toxicity of sediment and sediment elutriate (certified reference material E7 sediment, National Water Research Institute, Burlington, ON) from 20 to over 4000-fold based on the microbial assays – Microtox and Luminotox solid phase assays (Blaise et al., 2008). Physiologically, Sm and Er caused cytotoxicity (activity of enzyme to reduce 3-(4,5-Dimethylthiazol-2-yl)-2,5diphenyltetrazolium bromide, test for cell viability and growth) of hepatocytes in rainbow trout (48 h threshold effect nominal concentrations:  $Sm > 43 \text{ mg} \cdot L^{-1}$ ;  $Er = 0.44 - 4.4 \text{ mg} \cdot L^{-1}$ ) (Blaise et al., 2008). Gd also caused cytotoxicity (nominal 24 h  $EC_{50} = 151.5 \text{ mg} \cdot \text{L}^{-1}$ ) and inhibited EROD activity (nominal 24 h  $EC_{50} = 58.4 \text{ mg} \cdot \text{L}^{-1}$ ) of primary rainbow trout hepatocytes (Laville et al., 2004). Glutamate-pyruvate transaminase activity in goldfish (Carassius auratus) liver was stimulated at 0.05 mg·L⁻¹ Yb³⁺ and inhibited at higher concentrations. In addition, REEs affect antioxidant enzymatic activities in the fish. Activity of superoxide dismutase in the goldfish was stimulated at Yb higher than 0.05 mg·L⁻¹ and catalase was strongly inhibited after 40 d of exposure, whereas glutathione S-transferase and glutathione peroxidise were stimulated at 0.05 mg·L⁻¹ and inhibited at 0.1 mg·L⁻¹ Yb (Guo et al., 2002).

# **Bioaccumulation by Aquatic Organisms**

Lanthanides can be taken up by aquatic organisms from surface water, pore water and sediment. Bioaccumulation of lanthanides by aquatic organisms is summarized in Table 7 (freshwater) and 8 (saltwater). Among the freshwater bioaccumulation studies, only studies with sufficient data on BCF or both tissue and exposure concentrations are included in the tables. Note that, all the exposure concentrations of REEs in the freshwater studies are nominal values, therefore bioconcentration factor (BCF) may be under-estimated. Bioconcentration factor (BCF) in *Cyprinus carpio* is the highest in internal organs where is the major site for metabolism and detoxification, the intermediate is gill and the lowest is in muscle and skeleton (Tu et al., 1994;

Sun et al., 1996). However, BCF or tissue concentration of La in *C. carpio* in Sun et al. (1996) is much higher than in Tu et al. (1994) under an exposure at the same water hardness. It may be explained by the inverse relationship between exposure concentration and BCF (exposure concentration in Tu et al. 94 is higher than Sun et al. 96) (McGeer et al., 2003). In general, bioaccumulation by the duckweed is higher than the fish, *C. carpio* (Table 7). As reported by Weltje et al. (2002b), BCF of REEs (La – Lu) in freshwater plants and mollusc is between 10,000 and 100,000 L· kg⁻¹ dry weight (duckweed *Lemna minor*: 10,000; pondweed *Potamogeton pectinatus*: 5,000 – 300,000; snail soft tissue: 5,000 – 200,000; bivalve soft tissue: 3,000 -30,000). There was only a low extent of biomagnifications (biomagnification factor: 5.5) from plant (food) to snail tissue due to a similarity between BCF of pondweed and snail tissue (Weltje et al., 2002b).

In salt water, most of the studies only measured tissue concentration of organisms collected from the field, but no REE concentration or salinity were reported (Table 8). There is no significant difference in tissue concentration of REEs among the cephalopod, mussel and scallop from the unpolluted sites (Lobel et al., 1991; Riget et al., 1996; Bustamante and Miramand, 2005; Pernice et al., 2009), but there is a variation of distribution of REEs among different tissues in the body. When the sites are polluted by REEs, invertebrates tend to distribute most of REEs in the digestive gland, following the gill, gonad and kidney, and the least is in the muscle (Table 8). There was 61 - 79% Ce, 72 - 81%, La, 55 - 75% Nd in the digestive glands of the two *Nautilus* species (Pernice et al., 2009). It highlighted that digestive gland is the important site for bioaccumulation of REEs. Stronkhorst and Yland (1998) also reported the BCF of an estuarine amphipod, Corophium volutator decreases with increasing atomic number of REEs (La: 28,840; Ce: 48,978; Pr: 38,905; Nd: 29,512; Sm: 17,783; Eu: 11,220; Gd: 13,183; Tb: 12,882; Dy: 9,550; Ho: 8,511; Er: 7,413; Tm: 6,310; Yb: 6,607; Lu: 4,786; all in 1 kg⁻¹). Since this data set is from a secondary reference written in other language, and information is lacking such as BCF unit expression (wet or dry weight), it is not included in Table 7 or 8. However, a trend of higher BCF is observed in the amphipod than the freshwater fish, C. carpio.

# **Data Gaps & Recommendations for Toxicity Testing**

# i). Test Elements

Lighter REEs (La – Gd) may have a higher priority than heavy lanthanides (Tb – Lu) for toxicity testing because they are naturally more abundant. In addition, they are commonly applied in industries, e.g., La (fertilizers); Ce, Sm (fuel additive, nanoparticles); Gd (pharmaceutical products) (Barry and Meehan, 2000; Laville et al., 2004; Blaise et al., 2008; Gaiser et al., 2009). Ligher REEs also have a higher solubility which suggests they are more bioavailable to biota. Among the lighter REEs, Ce and Eu need to be particularly concerned because they have different physicochemical behaviour from other REEs. Pm can be neglected for testing due to the absence of their natural occurrence, except for studying effects of their radioisotopes.

# ii). Test Systems

Freshwater environment is relatively more studied than saltwater environment, with both acute and chronic data available for fresh water, but only acute data available for salt water. Sneller et al. (2000) reported that saltwater organisms in general are more susceptible to lanthanides than freshwater organisms. Our literature also demonstrates this trend in the crustaceans (Table 4 & 6), otherwise data is limited from a few species of each system to support this statement. Toxicity of lanthanides in pore water which has a lower pH, thus a higher dissolved concentration, and the sediment which is the largest reservoir for bioavailable lanthanides in water are rarely studied (only Borgmann et al. 2005; Watson-Leung, 2009, Table 5). Although sediment bound lanthanides are considered to be not bioavailable, bioavailability may change when the equilibrium is disturbed (e.g., change in pH) in the sediment and biota systems. In general, data is still lacking on different areas (see sections below) of each aquatic system.

# iii). Test Species

A lot of tests have been conducted on the cladocerans, but tests on other species from different families are insufficient. US EPA (Stephan et al., 1985; EPA, 1994) requires toxicity data from a suite of aquatic organisms for developing the water quality criteria and conducting the ecological risk assessment. They are algae; fish (salmonid & non-salmonid); planktonic crustacean; benthic crustacean; insect and annelid or mollusk. Among the toxicity tests from the literature, we are lack of studies on the freshwater salmonids e.g., rainbow trout (only Watson-Leung, 2009, Table 4) and sensitive non-salmonid species. Fathead minnow is a recommended sensitive non-

salmonid species for toxicity testing on their life cycle embryo-larval/teratogenicity (chronic) by EPA (1994). There are also insufficient data on the freshwater amphipod (only Borgmann et al., 2005, Table 5), different REEs on freshwater insect (only La in Watson-Leung, 2009), annelid worms and mollusks. The oligochaete *Lumbriculus* or *Tubifex tubifex* is recommended to be the tested freshwater annelid because toxicity tests have been well developed for them (Chapman et al., 1999), whereas the pond snail *Lymnaea stagnalis* as the tested freshwater mollusk as it is very sensitive to pollutants (e.g., Grosell et al., 2006). Very limited toxicity data is available for acute tests on saltwater organisms (one species of algae, fish and invertebrates only, Table 6) and chronic toxicity tests on all saltwater species are lacking. Suggested chronic tests on saltwater organisms are survival and development of marine bivalve larvae, sea urchin fertilization, mysid survival, growth and fecundity which are all standard toxicity tests of EPA (1994).

### iv). Test Methods

About 50 % of toxicity values from literature are nominal or total REEs concentrations. However, dissolved lanthanide concentrations can deviate from nominal values of more than 50 % (Lürling and Tolman, 2010) since a lot of them can be adsorbed to containers or precipitate within 24 h of exposure (Barry and Meehan, 2000). Since only dissolved lanthanide can cause toxicity to the biota, it is important to measure concentrations in dissolved form. Keep the containers for equilibrium with lanthanides for 24 h before exposure starts, and measure the dissolved concentrations regularly during the toxicity tests. Daily renewal of the medium or flow-through system is also important to maintain the lanthanide concentrations. Using nominal or total lanthanide concentration for calculating BCF is also a concern in bioaccumulation studies because the ratio of lanthanides taken up from the water is under-estimated.

### v). Test Topics

Water chemistry has been demonstrated to play an important role on metal toxicity. Some abiotic ligands (Ca; Mg; carbonates; pH) have already been shown to affect uptake of lanthanides by algae, therefore effects of a wide range of toxicity modifying factors (Ca, Mg, Na, pH, carbonates, hydroxides, DOC, salinity) are of importance for further investigations especially in the fish and invertebrates. In order to fulfill the goals, water chemistry parameters need to be measured throughout the toxicity tests. In freshwater system, since DOC plays an important role on speciation of lanthanides (Maas and Botterweg, 1993), quality and quantity of dissolved organic matter on toxicity and bioavailability of lanthanides are of interest for investigations. Biotic ligand e.g., gill in fish, is the site of toxic action, similar to metals, physiological binding

constants of lanthanides (Log K,  $B_{max}$ ) can be quantified on the fish gill or whole body of invertebrates (Di Toro et al., 2001). In addition, short-term lanthanide burden (3-24 h) can be measured to determine if there is a relationship between short-term burden and acute or chronic toxicity on the organisms. These binding constants do not exist for lanthanides. Collection of all these data will help to develop a new model, or add data to an existing model e.g., Biotic Ligand Model for predicting toxicity of lanthanides, as well as metals. Toxicity of lanthanides in different form, e.g., methylated or organic lanthanides should not be neglected for testing as bioavailability of them compared to from salts is less known. In addition, as dietary toxicity of pollutants, e.g., metals, are getting more interests, effects of lanthanide from diet through trophic transfer is not studied yet.

Table 1: Search strategies, keywords & results

	ategies, keywords & resui	Results					
	Keywords	# of papers / abstracts	# papers retained for				
		screened ("hits")	further screening				
	environmental toxicity	1464	20				
Rare earth elements	in fish	135	10				
	bioavailability	59	3				
	toxicity	614	20				
Lanthanides	environment toxicity	578	67				
	bioavailability	11	3				
	ecotoxicity	0	0				
	environmental toxicity	28	15				
	speciation	14	2				
Lanthanum	bioaccumulation	6	6				
	in fish	79	16				
	in invertebrates	294	35				
	bioavailability	24	2				
	ecotoxicity	0	0				
	environmental toxicity	17	14				
	toxicity	221	126				
	speciation	9	7				
	bioaccumulation	7	7				
Cerium	physiology	134	0				
	in fish	138	20				
	in invertebrates	37	14				
	in plants	249	36				
	bioavailability	17	1				
	ecotoxicity	0	0				
	environmental toxicity	0	0				
	toxicity	19	13				
	speciation	0	0				
Praseodymium	bioaccumulation	0	0				
	in fish	1	1				
	in invertebrates	2	0				
	in plants	8	5				
	bioavailability	4	0				
NT 1 '	ecotoxicity	0	0				
Neodymium	environmental toxicity	0	0				

	toxicity	31	14		
	speciation	1	0		
	bioaccumulation	0	0		
	in fish	1	1		
	in invertebrates	0	0		
	in plants	17	8		
	bioavailability	6	0		
	ecotoxicity	0	0		
	environmental toxicity	0	0		
	toxicity	9	7		
	speciation	0	0		
Promethium	bioaccumulation	0	0		
	in fish	0	0		
	in invertebrates	0	0		
	in plants	4	1		
	bioavailability	1	0		
	ecotoxicity	1	1		
	environmental toxicity	3	1		
	toxicity	85	18		
	speciation	0	0		
Samarium	bioaccumulation	0	0		
	in fish	1	1		
	in invertebrates	9	3		
	in plants	11	9		
	bioavailability	10	0		
	ecotoxicity	0	0		
	environmental toxicity	0	0		
	toxicity	26	25		
	physiology	812	104		
Europium	speciation	24	12		
Europium	bioaccumulation	9	9		
	in fish	19	3		
	in invertebrates	23	9		
	in plants	40	17		
	bioavailability	7	0		
	ecotoxicity	0	0		
Gadolinium	environmental toxicity	35	9		
	toxicity	519	32		

	speciation	13	2
	bioaccumulation	1	1
	in fish	28	7
	in invertebrates	52	6
	in plants	74	3
	bioavailability	10	0
	ecotoxicity	0	0
	environmental toxicity	6	5
	toxicity	19	10
	speciation	0	0
Terbium	bioaccumulation	0	0
	in fish	29	3
	in invertebrates	16	0
	in plants	14	7
	bioavailability	4	0
	ecotoxicity	0	0
	environmental toxicity	2	1
	toxicity	16	12
	speciation	0	0
Dysprosium	bioaccumulation	0	0
	in fish	0	0
	in invertebrates	1	1
	in plants	18	2
	bioavailability	5	0
	ecotoxicity	0	0
	environmental toxicity	2	2
	toxicity	21	9
	speciation	0	0
Holmium	bioaccumulation	0	0
	in fish	1	1
	in invertebrates	1	1
	in plants	3	3
	bioavailability	2	0
	ecotoxicity	1	1
	environmental toxicity	1	1
Erbium	toxicity	9	4
	speciation	0	0
	bioaccumulation	0	0

In invertebrates		in fish	1	1
Display		in invertebrates	4	3
Cotoxicity   0		in plants	6	2
Convironmental toxicity		bioavailability	1	0
Thulium		ecotoxicity	0	0
Speciation		environmental toxicity	0	0
Thulium   bioaccumulation   0   0   0   0   in fish   0   0   0   0   0   in fish   0   0   0   0   0   0   0   0   0		toxicity	4	3
In fish		speciation	1	1
in invertebrates   0	Thulium	bioaccumulation	0	0
In plants   4		in fish	0	0
Description		in invertebrates	0	0
Cotoxicity   Cot		in plants	4	2
Provided Heat		bioavailability	2	0
Toxicity   22   10		ecotoxicity	0	0
Speciation   1		environmental toxicity	2	1
Ytterbium         bioaccumulation         0         0           in fish         10         7           in invertebrates         1         1           in plants         22         3           bioavailability         2         0           ecotoxicity         0         0           environmental toxicity         0         0           toxicity         42         11           speciation         4         3           bioaccumulation         1         1           in fish         0         0           in invertebrates         1         1           in plants         3         2           bioavailability         3         0           Total         6359         862           # papers retained for detailed review         577           (excluding duplicates)         577		toxicity	22	10
in fish 10 7 in invertebrates 1 1 1 in plants 22 3 bioavailability 2 0 ecotoxicity 0 0 0 environmental toxicity 0 0 0 toxicity 42 11 speciation 4 3 bioaccumulation 1 1 1 in fish 0 0 0 in invertebrates 1 1 1 in plants 3 2 bioavailability 3 0  Total 6359 862 # papers retained for detailed review (excluding duplicates)	Ytterbium	speciation	1	1
in invertebrates		bioaccumulation	0	0
in plants   22   3		in fish	10	7
bioavailability   2		in invertebrates	1	1
ecotoxicity		in plants	22	3
environmental toxicity		bioavailability	2	0
Lutetium       toxicity       42       11         speciation       4       3         bioaccumulation       1       1         in fish       0       0         in invertebrates       1       1         in plants       3       2         bioavailability       3       0         Total       6359       862         # papers retained for detailed review (excluding duplicates)       577		ecotoxicity	0	0
Speciation		environmental toxicity	0	0
Lutetium         bioaccumulation         1         1           in fish         0         0           in invertebrates         1         1           in plants         3         2           bioavailability         3         0           Total         6359         862           # papers retained for detailed review (excluding duplicates)         577		toxicity	42	11
in fish   0   0       in invertebrates   1   1     in plants   3   2     bioavailability   3   0      Total   6359   862     # papers retained for detailed review (excluding duplicates)   577		speciation	4	3
in invertebrates	Lutetium	bioaccumulation	1	1
in plants bioavailability 3 0  Total 6359 862 # papers retained for detailed review (excluding duplicates)		in fish	0	0
bioavailability 3 0  Total 6359 862  # papers retained for detailed review 577  (excluding duplicates)		in invertebrates	1	1
Total 6359 862  # papers retained for detailed review 577  (excluding duplicates)		in plants	3	2
# papers retained for detailed review 577 (excluding duplicates)		bioavailability	3	0
(excluding duplicates)			6359	862
		niled review		577
# additional articles from literature 52				
	# additional articles from	literature		52
Total papers entered into EndNote database for review 629  Note: "Biograpia bility" search was only run in Web of Science database and only articles relevant to aquatic				

Note: "Bioavailability" search was only run in Web of Science database and only articles relevant to aquatic organisms were retained.

Table 2. Rank of abundance of lanthanides in earth crust and their associated minerals

Lanthanide	Abundance in Earth's crust (rank)	Mineral where found
Lanthanum (La)	28	monazite; bastnasite
Cerium (Ce)	26	monazite; bastnasite; allanite.
Praseodymium (Pr)	37	monazite; bastnasite
Neodymium (Nd)	27	monazite; bastnasite
Samarium (Sm)	40	monazite, bastnasite, samarskite
Europium (Eu)	50	monazite, bastnaesite
Gadolinium (Gd) 41		samarskite, gadolinite, monazite,some varieties of Norwegian ytterspar
Terbium (Tb)	58	cerite, gadolinite, monazite, xenotime, euxenite
Dysprosium (Dy)	42	xenotime, fergusonite, gadolinite, euxenite, polycrase, blomstrandine, monazite, bastnäsite
Holmium (Ho)	55	gadolinite, monazite
Erbium (Er)	43	monazite
Thulium (Tm)	61	euxenite, gadolinite, blomstrandine
Ytterbium (Yb)	29	monazite, euxenite, xenotime.
Lutetium (Lu)	59	gadolinite, monazite, and xenotime

^{*}Promethium is a by-product from decay of uranium and it is not naturally present in the earth crust.

Table 3. A list of dominant species of lanthanide complex in solution (Moeller and Vincenti, 1965; Cantrell and Byrne, 1987; Brookins, 1989).

Inorganic	Remarks
$REE(X)^{2+}$	X= F, Cl, Br, I
$REE(X_2)^+$	X=Cl, Br
REE(ClO ₄ ) ²⁺	
REE(NO ₃ ) ²⁺	
$REE(P_2O_7)_n^{(3-4n)+}$	n=1, 2
$REE(SO_4)^+$	
$REE(SO_4)^{2-}$	
$REE(CO_3)^+$	
REE(CO ₃ ) ²	
Organic	
$REE(C_2H_3O_2)_n^{(3-n)+}$	n= 1-3
REE(HOCH ₂ COO) _n (3-n)+	n= 1-4
REE(EDTA)	
REE(NTA) _n (3-n)+	n=1,2
REE(HEDTA)(IMDA) ²⁻	
REE(HEDTA)(OH)	
REE(b-diketon) _n (3-n)+	n= 1-3
$REE(PDC)_n^{(3-2n)+}$	n=1-3
REE(NO ₃ ) ₃ ·3TBP	

Table 4. Acute toxicity of lanthanides to freshwater organisms

Pseudokirchneriella   Sm   Sm ₃ O ₃     S   D ³     6-6.8     72 h IC25   Growth   430 - 4,300 ⁵	(Blaise et al., 2008) (Blaise et al., 2008) (Blaise et al., 2008)
Algae   Subcapitata   Er   Er ₂ O ₃     S   D ^a     6-6.8     72 h IC25   Growth   440 - 4,400 ^b	(Blaise et al., 2008) (Blaise et al.,
Algae   Subcapitata   Er   ErO3     S   D ^a     6-6.8     72 h IC25   Growth   440 - 4,400 ^b	(Blaise et al.,
Algae   subcapitata   Ho   Ho ₂ O ₃     S   D ^b     6-6.8     72 h IC25   Growth   440 - 4,400 ^b	,
Pseudokirchneriella   CeO   nanoparticles   CeO   nanoparticles   Cells ml¹   S   N, T   24.2   7.4   25   72 h IC50   Growth   8,302 - 15,470	2008)
Algae   subcapitata   Ce   nanoparticles   cells ml ⁻¹   S   N, T   24.2   7.4   25   72 h IC50   Growth   8,302 - 15,476	(Van Hoecke et
Algae         capricornutum         La         LaCl ₃ S         N, T         <10          24         72 h IC50         Growth         450           Cladoceran         Daphnia carinata         La         LaCl ₃ neonate         SR         M, D         160         7.5         20         24 h EC50         Survival         1232.4           Cladoceran         Daphnia carinata         La         LaCl ₃ neonate         SR         M, D         160         7.5         20         24 h EC50         Survival         1232.4           Cladoceran         Daphnia carinata         La         LaCl ₃ neonate         SR         M, D         160         7.5         20         24 h EC50         Survival         1180           Cladoceran         Daphnia carinata         La         LaCl ₃ neonate         SR         M, D         22         7.5         20         24 h EC50         Survival         48.5           Cladoceran         Daphnia magna         Ce         neonate         SR         M, D         22         7.5         20         48 h EC50         Survival         43.2           Cladoceran         Daphnia magna         Ce         neonate<	oc al., 2009)
Cladoceran         Daphnia carinata         La         LaCl ₃ neonate         SR         M, D         98         7.8         20         48 h EC50         Survival         49           Cladoceran         Daphnia carinata         La         LaCl ₃ neonate         SR         M, D         160         7.5         20         24 h EC50         Survival         1232.4           Cladoceran         Daphnia carinata         La         LaCl ₃ neonate         SR         M, D         160         7.5         20         24 h EC50         Survival         1180           Cladoceran         Daphnia carinata         La         LaCl ₃ neonate         SR         M, D         22         7.5         20         24 h EC50         Survival         484.5           Cladoceran         Daphnia carinata         La         LaCl ₃ neonate         SR         M, D         22         7.5         20         24 h EC50         Survival         48.5           Cladoceran         Daphnia magna         Ce         bulk particles         s         N, T         84.9           96 h LC50         Survival         NOEC: 8139.           Cladoceran         Daphnia magna         L	(NICNAS,
Cladoceran         Daphnia carinata         La         LaCl ₃ neonate         SR         M, D         160         7.5         20         24 h EC50         Survival         1232.4           Cladoceran         Daphnia carinata         La         LaCl ₃ neonate         SR         M, D         160         7.5         20         48 h EC50         Survival         1180           Cladoceran         Daphnia carinata         La         LaCl ₃ neonate         SR         M, D         22         7.5         20         24 h EC50         Survival         484.5           Cladoceran         Daphnia carinata         La         LaCl ₃ neonate         SR         M, D         22         7.5         20         24 h EC50         Survival         484.5           Cladoceran         Daphnia magna         Ce         Daphnia magna         Ce         Daphnia magna         Ce         Survival         N, T         84.9           96 h LC50         Survival         NOEC: 8139.2           Cladoceran         Daphnia magna         Ce         nanoparticles         <24 h         S         M, T         210         6.5-8.2          48 h EC50         Survival         24,000	2001)
Cladoceran         Daphnia carinata         La         LaCl ₃ neonate         SR         M, D         160         7.5         20         24 h EC50         Survival         1232.4           Cladoceran         Daphnia carinata         La         LaCl ₃ neonate         SR         M, D         160         7.5         20         48 h EC50         Survival         1180           Cladoceran         Daphnia carinata         La         LaCl ₃ neonate         SR         M, D         22         7.5         20         24 h EC50         Survival         484.5           Cladoceran         Daphnia carinata         La         LaCl ₃ neonate         SR         M, D         22         7.5         20         24 h EC50         Survival         484.5           Cladoceran         Daphnia magna         Ce         Daphnia magna         Ce         Daphnia magna         Ce         Survival         N, T         84.9           96 h LC50         Survival         NOEC: 8139.2           Cladoceran         Daphnia magna         Ce         nanoparticles         <24 h         S         M, T         210         6.5-8.2          48 h EC50         Survival         24,000	(Barry and
Cladoceran         Daphnia carinata         La         LaCl ₃ neonate         SR         M, D         160         7.5         20         48 h EC50         Survival         1180           Cladoceran         Daphnia carinata         La         LaCl ₃ neonate         SR         M, D         22         7.5         20         24 h EC50         Survival         484.5           Cladoceran         Daphnia carinata         La         LaCl ₃ neonate         SR         M, D         22         7.5         20         48 h EC50         Survival         43.2           Cladoceran         Daphnia magna         Ce         CeO ₂ nano & bulk particles         neonate         S         N, T         84.9           96 h LC50         Survival         NOEC: 8139.2           Cladoceran         Daphnia magna         Ce         nanoparticles         <24 h	Meehan, 2000)
Cladoceran         Daphnia carinata         La         LaCl ₃ neonate         SR         M, D         160         7.5         20         48 h EC50         Survival         1180           Cladoceran         Daphnia carinata         La         LaCl ₃ neonate         SR         M, D         22         7.5         20         24 h EC50         Survival         484.5           Cladoceran         Daphnia carinata         La         LaCl ₃ neonate         SR         M, D         22         7.5         20         48 h EC50         Survival         43.2           Cladoceran         Daphnia magna         Ce         CeO ₂ nano & bulk particles         neonate         S         N, T         84.9           96 h LC50         Survival         NOEC: 8139.2           Cladoceran         Daphnia magna         Ce         nanoparticles         <24 h	(Barry and
Cladoceran         Daphnia carinata         La         LaCl ₃ neonate         SR         M, D         22         7.5         20         24 h EC50         Survival         484.5           Cladoceran         Daphnia carinata         La         LaCl ₃ neonate         SR         M, D         22         7.5         20         48 h EC50         Survival         43.2           Cladoceran         Daphnia magna         Ce         bulk particles         neonate         S         N, T         84.9           96 h LC50         Survival         NOEC: 8139.           Cladoceran         Daphnia magna         Ce         nanoparticles         <24 h	Meehan, 2000)
Cladoceran         Daphnia carinata         La         LaCl ₃ neonate         SR         M, D         22         7.5         20         24 h EC50         Survival         484.5           Cladoceran         Daphnia carinata         La         LaCl ₃ neonate         SR         M, D         22         7.5         20         48 h EC50         Survival         43.2           Cladoceran         Daphnia magna         Ce         bulk particles         neonate         S         N, T         84.9           96 h LC50         Survival         NOEC: 8139.           Cladoceran         Daphnia magna         Ce         nanoparticles         <24 h	(Barry and
Cladoceran         Daphnia carinata         La         LaCl ₃ neonate         SR         M, D         22         7.5         20         48 h EC50         Survival         43.2           Cladoceran         Daphnia magna         Ce         bulk particles bulk particles         neonate         S         N, T         84.9           96 h LC50         Survival         NOEC: 8139.           Cladoceran         Daphnia magna         Ce         nanoparticles         <24 h         S         N, T         249         7.4         20         48 h EC50         Survival         NOEC: 8139.           Cladoceran         Daphnia magna         La          <24 h         S         M, T         210         6.5-8.2          48 h EC50         Survival         24,000           Cladoceran         Daphnia magna         Ce          <24 h         S         M, T         210         6.2-8.0          48 h EC50         Survival         22,000           Cladoceran         Daphnia magna         Pr          <24 h         S         M, T         210         6.2-8.2          48 h EC50         Survival         9,000	Meehan, 2000)
Cladoceran         Daphnia carinata         La         LaCl ₃ neonate         SR         M, D         22         7.5         20         48 h EC50         Survival         43.2           Cladoceran         Daphnia magna         Ce         bulk particles bulk particles         neonate         S         N, T         84.9           96 h LC50         Survival         NOEC: 8139.           Cladoceran         Daphnia magna         Ce         nanoparticles         <24 h         S         N, T         249         7.4         20         48 h EC50         Survival         NOEC: 8139.           Cladoceran         Daphnia magna         La          <24 h         S         M, T         210         6.5-8.2          48 h EC50         Survival         24,000           Cladoceran         Daphnia magna         Ce          <24 h         S         M, T         210         6.2-8.0          48 h EC50         Survival         22,000           Cladoceran         Daphnia magna         Pr          <24 h         S         M, T         210         6.2-8.2          48 h EC50         Survival         9,000	(Barry and
Cladoceran         Daphnia magna         Ce bulk particles         neonate         S         N, T         84.9           96 h LC50         Survival         NOEC: 8139.           Cladoceran         Daphnia magna         Ce         nanoparticles         <24 h	Meehan, 2000) (Barry and
Cladoceran         Daphnia magna         Ce bulk particles         neonate         S         N, T         84.9           96 h LC50         Survival         NOEC: 8139.           Cladoceran         Daphnia magna         Ce         nanoparticles         <24 h	Meehan, 2000)
Cladoceran         Daphnia magna         Ce         bulk particles         neonate         S         N, T         84.9           96 h LC50         Survival         NOEC: 8139.           Cladoceran         Daphnia magna         Ce         nanoparticles         <24 h	(Gaiser et al.,
Cladoceran         Daphnia magna         Ce         CeO2 nanoparticles         <24 h         S         N, T         249         7.4         20         48 h EC50         Survival         NOEC: 8139: N	,
Cladoceran         Daphnia magna         Ce         nanoparticles         <24 h         S         N, T         249         7.4         20         48 h EC50         Survival         NOEC: 8139: NOEC:	(Van Hoecke et
Cladoceran         Daphnia magna         La          <24 h         S         M, T         210         6.5-8.2          48 h EC50         Survival         24,000           Cladoceran         Daphnia magna         Ce          <24 h	`
Cladoceran         Daphnia magna         Ce          <24 h         S         M, T         210         6.2-8.0          48 h EC50         Survival         22,000           Cladoceran         Daphnia magna         Pr          <24 h	(Den-Ouden, 1995) ^{SD}
Cladoceran         Daphnia magna         Pr          <24 h         S         M, T         210         6.2-8.2          48 h EC50         Survival         9,000	(Den-Ouden,
	1995) ^{SD}
	(Den-Ouden, 1995) ^{SD}
Cladoceran         Daphnia magna         Nd          <24 h         S         M, T         210         6.4-8.3          48 h EC50         Survival         1,400	(Den-Ouden.
	1995) ^{SD}
Cladoceran         Daphnia magna         Sm          <24 h         S         M, T         210         6.3-8.2          48 h EC50         Survival         7,600	(Den-Ouden, 1995) ^{SD}
Cladoceran         Daphnia magna         Gd          <24 h         S         M, T         210         6.5-8.2          48 h EC50         Survival         6,800	(Den-Ouden, 1995) ^{SD}
	(Den-Ouden,
Cladoceran         Daphnia magna         Dy          <24 h         S         M, T         210         6.5-8.1          48 h EC50         Survival         9,100	1995) ^{SD}
Cladoceran         Daphnia magna         La         dPhoslock®         <24 h         S         M, D         192         6.8-8.3         21         48 h LC50         Survival         NOEC > 632'	(Watson-Leung, 2009)
Cladoceran Ceriodaphnia dubia La dephoslock® <24 h S M, D 84.9 7 25 48 h EC50 Survival 80	(Stauber, 2000)
Ciauoceian Cerioaapinia auoia La Pilosiock® <24 ii S IVI, D 84.9 / 25 48 ii EC30 Survival 80	(NICNAS,
Cladoceran Ceriodaphnia dubia La LaCl ₃ <24 h S M, T 40-48 7 25 48 h EC50 Survival 5,000	

Table 4. Acute toxicity of lanthanides to freshwater organisms (con't)

Organism	Species	Base element	Compound	Organism age or Size	Exposure method*	Element conc.*	Hardness (mg CaCO ₃ l ⁻¹ )	pН	Temp	Endpoint	Effect	Toxicity value (µg REE l ⁻¹ )	Reference*
								6-	20 –				(Blaise et al.,
Cnidarian	Hydra attenuata	Sm	$Sm_2O_3$	adult	S	D ^a	130	6.8	24	96 h EC50	Morphology	$430 - 4,300^{b}$	2008)
Cnidarian	Hydra attenuata	Er	Er ₂ O ₃	adult	S	$\mathbf{D}^{\mathrm{a}}$	130	6- 6.8	20 – 24	96 h EC50	Morphology	$440 - 4,400^{b}$	(Blaise et al., 2008)
Cnidarian	Hydra attenuata	Но	Ho ₂ O ₃	adult	S	Da	130	6- 6.8	20 – 24	96 h EC50	Morphology	44 – 440 ^b	(Blaise et al., 2008)
Cilidarian	Thamnocephalus	110	110203	20-22 h instar	S	D	130	6-	24	90 II EC30	Wiorphology	44 - 440	(Blaise et al.,
Shrimp	platyurus	Sm	Sm ₂ O ₃	(stage II-III)	S	$\mathbf{D}^{\mathrm{a}}$		6.8	25	24 h LC50	Survival	> 43,000 ^b	2008)
1	Thamnocephalus		2-3	20-22 h instar				6-				- ,	(Blaise et al.,
Shrimp	platyurus	Er	$Er_2O_3$	(stage II-III)	S	$\mathbf{D}^{\mathrm{a}}$		6.8	25	24 h LC50	Survival	$>44,000^{b}$	2008)
Shrimp	Thamnocephalus platyurus	Но	Ho ₂ O ₃	20-22 h instar (stage II-III)	S	$\mathbf{D}^{\mathrm{a}}$		6- 6.8	25	24 h LC50	Survival	> 44,000 ^b	(Blaise et al., 2008)
1	Thamnocephalus		CeO ₂	20-22 h instar								,	(Van Hoecke
Shrimp	platyurus	Ce	nanoparticles	(stage II-III)	S	N, T		7.4	25	24 h LC50	Survival	4,069,767	et al., 2009)
Fish	Melanotaenia duboulayi	La	dPhoslock®	juvenile	S	M, D	84.9	6.9		96 h EC50	Immobilization	NOEC: 127	(Stauber, 2000)
	Melanotaenia			, , , , , ,		,		6.5-	23.4 -				(NICNAS,
Fish	duboulayi	La	LaCl ₃	juvenile	S	M, D	40-48	8.1	24.5	96 h EC50	Immobilization	NOEC < 600	2001)
Fish	Danio rerio	Ce	CeO ₂ nanoparticles	<30 min after spawning	S	N, T	209	7.4	28	72 h EC10	Hatching	NOEC: 162791	(Van Hoecke et al., 2009)
Fish	Danio rerio	La			SR	M. T	210	6.4- 8.2		96 h LC50	Survival	23.000	(Den-Ouden, 1995) ^{SD}
					~	, -		6.3-					(Den-Ouden,
Fish	Danio rerio	Ce			SR	M, T	210	8.1		96 h LC50	Survival	22,000	1995) ^{SD}
Fish	Danio rerio	Pr			SR	M, T	210	6.2- 8.2		96 h LC50	Survival	25,000	(Den-Ouden, 1995) ^{SD}
Fish	Danio rerio	Nd			SR	M, T	210	6.5- 8.4		96 h LC50	Survival	21,000	(Den-Ouden, 1995) ^{SD}
1 1011	Danio Terro	110			511	1,1, 1	210	6.3-		70112000	Surviva	21,000	(Den-Ouden,
Fish	Danio rerio	Sm			SR	M, T	210	8.0		96 h LC50	Survival	22,000	1995) ^{SD}
Fish	Danio rerio	Gd			SR	M, T	210	6.5- 8.0		96 h LC50	Survival	19,000	(Den-Ouden, 1995) ^{SD}
Fish	Danio rerio	Dy			SR	M. T	210	6.2- 8.0		96 h LC50	Survival	25,000	(Den-Ouden, 1995) ^{SD}
1 1011	Oncorhynchus	Dy			DIC	171, 1	210	7.1-		75 II EC30	Survivui	NOEC >	(Watson-
Fish	mykiss	La	dPhoslock®	juvenile	S	M, D	128	8.4	15	96 h LC50	Survival	63270	Leung, 2009)

^{*}S = static; SR = static-renewal; M = measured; N = nominal; D = dissolved (pass through ≤ 0.45 µm filter); T = total; SD = secondary data, whereas unmarked references are primary data

Dissolved lanthanide concentration is calculated from passing the exposure medium through a 0.22 mm preweighed filter to get the insoluble concentration (Blaise et al. 2008)

^bExact endpoint values are not reported in the article, instead a range is reported according to the EU Directive 93/67/EEC

^cEC50 range is given for nanoparticles with different sizes (14, 20, 29 nm) (Van Hoecke et al. 2009) ^dPhoslock® is a lanthanum modified clay for water treatment

 $\label{thm:change} \textbf{Table 5. Chronic toxicity of lanthanides to freshwater organisms} \\$ 

Organism	Species	Base element	Compound	Element conc.*	Hardness (mg CaCO ₃ l ⁻¹ )	pН	Temp (°C)	Endpoint	Effect	Toxicity value (μg REE l ⁻¹ )	Reference*
Duckweed	Lemna minor	La	LaCl ₃	N, T	54.8	5.1	25	9 d EC50	Growth	NOEC: 1.39	(Weltje et al., 2002a)
Cladoceran	Daphnia carinata	La	LaCl ₃	M, D	160	7.5	20	Life cycle LC50	Survival	LOEC: 39	(Barry and Meehan, 2000)
Cladoceran	Daphnia carinata	La	LaCl ₃	M, D	160	7.5	20	Life cycle EC50	Age at maturity	LOEC: 39	(Barry and Meehan, 2000)
Cladoceran	Daphnia carinata	La	LaCl ₃	M, D	160	7.5	20	Life cycle EC50	Brood size	LOEC: 30	(Barry and Meehan, 2000)
Cladoceran	Daphnia magna	La	^a Phoslock®	M, D	88	7.6	20	5 d EC50	Weight	8.7 ^b	(Lürling and Tolman, 2010)
Cladoceran	Daphnia magna	La	^a Phoslock®	M, D	88	7.6	20	5 d EC50	Length	15.6 ^b	(Lürling and Tolman, 2010)
Cladoceran	Daphnia magna	La	La(NO ₃ ) ₃	M, D	88	7.6	20	Life cycle LC50	Survival	NOEC = 1,001	(Lürling and Tolman, 2010)
Cladoceran	Daphnia magna	La	La(NO ₃ ) ₃	M, D	88	7.6	20	Life cycle EC50	Brood size	NOEC = 1,001	(Lürling and Tolman, 2010)
Cladoceran	Daphnia magna	La	La(NO ₃ ) ₃	M, D	88	7.6	20	Life cycle EC50	Growth	NOEC = 1,001	(Lürling and Tolman, 2010)
Cladoceran	Daphnia magna	Ce	CeO ₂ nanoparticles	N, T	249	7.4	20	21 d LC50	Survival	30035 - 57872°	(Van Hoecke et al., 2009)
Cladoceran	Daphnia magna	Се	CeO ₂ nanoparticles	N, T	249	7.4	20	21 d EC50	Brood size	16686 – 34756°	(Van Hoecke et al., 2009)
Cladoceran	Daphnia magna	Nd		M, T	210	7.6-8.7		21 d LC50	Survival	NOEC: 1,600	(Den-Ouden, 1995) ^{SD}
Cladoceran	Daphnia magna	Nd		M, T	210	7.6-8.7		21 d EC50	Fitness	NOEC: 1,600	(Den-Ouden, 1995) ^{SD}
Cladoceran	Daphnia magna	Dy		M, T	210	-		21 d EC50	Reproduction ^d	NOEC < 200	(Den-Ouden, 1995) ^{SD}
Cladoceran	Daphnia magna	Dy		M, T	210	7.9-8.5		21 d LC50	Survival	NOEC > 2,100	(Den-Ouden, 1995) ^{SD}
Cladoceran	Daphnia magna	Dy		M, T	210	7.9-8.5		21 d EC50	Fitness	NOEC > 2,100	(Den-Ouden, 1995) ^{SD}
Cladoceran	Ceriodaphnia dubia	La	^a Phoslock®	M, D	84.9	7.9	25	7 d LC50	Survival	842	(NICNAS, 2001)
Cladoceran	Ceriodaphnia dubia	La	^a Phoslock®	M, D	84.9	7.9	25	7 d EC50	Brood size	154	(NICNAS, 2001)
Cladoceran	Ceriodaphnia dubia	La	LaCl ₃	M, T	40-48	7.9	25	7 d LC50	Survival	510	(Borgmann et al., 2005)
Cladoceran	Ceriodaphnia dubia	La	LaCl ₃	M, T	40-48	7.9	25	7 d EC50	Brood size	430	(Borgmann et al., 2005)
Amphipod	Hyallela azteca	Ce	CeO ₂	N, T	124	7.2-9.0	24 - 25	7 d LC50	Survival	651	(Borgmann et al., 2005)
Amphipod	Hyallela azteca	Ce	CeO ₂	M, D	18	6.4-8.7	24 - 25	7 d LC50	Survival	32	(Borgmann et al., 2005)
Amphipod	Hyallela azteca	Dy	Dy ₂ O ₃	N, T	124	7.2-9.0	24 - 25	7 d LC50	Survival	897	(Borgmann et al., 2005)
Amphipod	Hyallela azteca	Dy	Dy ₂ O ₃	M, D	18	6.4-8.7	24 - 25	7 d LC50	Survival	162	(Borgmann et al., 2005)
Amphipod	Hyallela azteca	Er	Er ₂ O ₃	N, T	124	7.2-9.0	24 - 25	7 d LC50	Survival	929	(Borgmann et al., 2005)
Amphipod	Hyallela azteca	Er	Er ₂ O ₃	M, D	18	6.4-8.7	24 - 25	7 d LC50	Survival	191	(Borgmann et al., 2005)
Amphipod	Hyallela azteca	Eu	Eu ₂ O ₃	N, T	124	7.2-9.0	24 - 25	7 d LC50	Survival	717	(Borgmann et al., 2005)
Amphipod	Hyallela azteca	Eu	Eu ₂ O ₃	M, D	18	6.4-8.7	24 - 25	7 d LC50	Survival	112	(Borgmann et al., 2005)
Amphipod	Hyallela azteca	Gd	Gd ₂ O ₃	N, T	124	7.2-9.0	24 - 25	7 d LC50	Survival	599	(Borgmann et al., 2005)
Amphipod	Hyallela azteca	Gd	$Gd_2O_3$	M, D	18	6.4-8.7	24 - 25	7 d LC50	Survival	150	(Borgmann et al., 2005)

Table 5. Chronic toxicity of lanthanides to freshwater organisms (con't)

Organism	Species	Base element	Compound	Element conc.*	Hardness (mg CaCO ₃ l ⁻¹ )	pН	Temp (°C)	Endpoint	Effect	Toxicity value (μg REE I ⁻¹ )	Reference*
Amphipod	Hyallela azteca	Но	Ho ₂ O ₃	N, T	124	7.2-9.0	24 - 25	7 d LC50	Survival	755	(Borgmann et al., 2005)
Amphipod	Hyallela azteca	Но	Ho ₂ O ₃	M, D	18	6.4-8.7	24 - 25	7 d LC50	Survival	143	(Borgmann et al., 2005)
Amphipod	Hyallela azteca	La	La ₂ O ₃	N, T	124	7.2-9.0	24 - 25	7 d LC50	Survival	1665	(Borgmann et al., 2005)
Amphipod	Hyallela azteca	La	La ₂ O ₃	M, D	18	6.4-8.7	24 - 25	7 d LC50	Survival	18	(Borgmann et al., 2005)
Amphipod	Hyallela azteca	Lu	Lu ₂ O ₃	N, T	124	7.2-9.0	24 - 25	7 d LC50	Survival	1054	(Borgmann et al., 2005)
Amphipod	Hyallela azteca	Lu	Lu ₂ O ₃	M, D	18	6.4-8.7	24 - 25	7 d LC50	Survival	29	(Borgmann et al., 2005)
Amphipod	Hyallela azteca	Nd	Nd ₂ O ₃	N, T	124	7.2-9.0	24 - 25	7 d LC50	Survival	511	(Borgmann et al., 2005)
Amphipod	Hyallela azteca	Nd	Nd ₂ O ₃	M, D	18	6.4-8.7	24 - 25	7 d LC50	Survival	55	(Borgmann et al., 2005)
Amphipod	Hyallela azteca	Pr	Pr ₆ O ₁₁	N, T	124	7.2-9.0	24 - 25	7 d LC50	Survival	441	(Borgmann et al., 2005)
Amphipod	Hyallela azteca	Pr	Pr ₆ O ₁₁	M, D	18	6.4-8.7	24 - 25	7 d LC50	Survival	35	(Borgmann et al., 2005)
Amphipod	Hyallela azteca	Sm	Sm ₂ O ₃	N, T	124	7.2-9.0	24 - 25	7 d LC50	Survival	846	(Borgmann et al., 2005)
Amphipod	Hyallela azteca	Sm	Sm ₂ O ₃	M, D	18	6.4-8.7	24 - 25	7 d LC50	Survival	74	(Borgmann et al., 2005)
Amphipod	Hyallela azteca	Tb	Tb ₄ O ₇	N, T	124	7.2-9.0	24 - 25	7 d LC50	Survival	693	(Borgmann et al., 2005)
Amphipod	Hyallela azteca	Tb	Tb ₄ O ₇	M, D	18	6.4-8.7	24 - 25	7 d LC50	Survival	84	(Borgmann et al., 2005)
Amphipod	Hyallela azteca	Tm	$Tm_2O_3$	N, T	124	7.2-9.0	24 - 25	7 d LC50	Survival	739	(Borgmann et al., 2005)
Amphipod	Hyallela azteca	Tm	Tm ₂ O ₃	M, D	18	6.4-8.7	24 - 25	7 d LC50	Survival	0.01	(Borgmann et al., 2005)
Amphipod	Hyallela azteca	Yb	Yb ₂ O ₃	N, T	124	7.2-9.0	24 - 25	7 d LC50	Survival	278	(Borgmann et al., 2005)
Amphipod	Hyallela azteca	Yb	Yb ₂ O ₃	M, D	18	6.4-8.7	24 - 25	7 d LC50	Survival	69	(Borgmann et al., 2005)
Amphipod	Hyallela azteca	La	^a Phoslock®	M, D	145-208	8.0-8.6	23	14 d	Survival	NOEC: 7	(Watson-Leung, 2009)
Amphipod	Hyallela azteca	La	^a Phoslock®	M, D	145-208	8.0-8.6	23	14 d	Growth	NOEC: 7	(Watson-Leung, 2009)
Insect	Chironomus dilutus	La	^a Phoslock®	M, T	138-179	8.1-9.1	21-23	10 d	Survival	NOEC: 880	(Watson-Leung, 2009)
Insect	Chironomus dilutus	La	^a Phoslock®	M, T	138-179	8.1-9.1	21-23	10 d	Weight	NOEC: 880	(Watson-Leung, 2009)
Insect	Hexagenia spp.	La	^a Phoslock®	M, D	101-125	8.1-8.5	22	21 d	Survival	NOEC < 3	(Watson-Leung, 2009)
Insect	Hexagenia spp.	La	^a Phoslock®	M, D	101-125	8.1-8.5	22	21 d	Weight	NOEC < 3	(Watson-Leung, 2009)
Fish	Danio rerio	Dy		M, T	210			30 d EC50	Weight	NOEC: 3,000	(Den-Ouden, 1995) ^{SD}
Fish	Danio rerio	Dy		M, T	210	6.7-8.4		30 d LC50	Survival	NOEC: 2,600	(Den-Ouden, 1995) ^{SD}
Fish	Danio rerio	Dy		M, T	210	6.7-8.4		30 d EC50	Fitness	NOEC: 3,800	(Den-Ouden, 1995) ^{SD}

^{*}M = measured; N = nominal; D = dissolved (pass through  $\leq$  0.45  $\mu m$  filter), T = total; SD = secondary data, whereas unmarked references are primary data

^aPhoslock® is a lanthanum modified clay for water treatment

^bEC50 of La is calculated from the measured percentage of La (0.001 %) leached from Phoslock® (Lürling and Tolman, 2010)

^cEC50 range is given for nanoparticles with different sizes (14, 20, 29 nm) (Van Hoecke et al. 2009) ^dNo information is given for what reproduction endpoint the study assessed.

Table 6. Acute toxicity of lanthanides to saltwater organisms

Organism	Species	Base element	Compound	Organism age or Size	Exposure method*	Element conc.*	Salinity (‰)	Temp (°C)	Endpoint	Effect	Toxicity value (μg REE I ⁻¹ )	Reference*
Algae	Skeletonema costatum	La	LaCl ₃	10 ⁴ cells ml ⁻¹	S	N, T	32-35	25	72 h EC50	Growth	4054.5	(Tai et al., 2010)
Algae	Skeletonema costatum	Ce	Ce(NO ₃ ) ₃	10 ⁴ cells ml ⁻¹	S	N, T	32-35	25	72 h EC50	Growth	4155.2	(Tai et al., 2010)
Algae	Skeletonema costatum	Nd	NdCl ₃	10 ⁴ cells ml ⁻¹	S	N, T	32-35	25	72 h EC50	Growth	4375	(Tai et al., 2010)
Algae	Skeletonema costatum	Sm	SmCl ₃	10 ⁴ cells ml ⁻¹	S	N, T	32-35	25	72 h EC50	Growth	4313.5	(Tai et al., 2010)
Algae	Skeletonema costatum	Eu	Eu(NO ₃ ) ₃	10 ⁴ cells ml ⁻¹	S	N, T	32–35	25	72 h EC50	Growth	4432.3	(Tai et al., 2010)
Algae	Skeletonema costatum	Gd	Gd(NO ₃ ) ₃	10 ⁴ cells ml ⁻¹	S	N, T	32–35	25	72 h EC50	Growth	4686	(Tai et al., 2010)
Algae	Skeletonema costatum	Tb	TbCl ₃	10 ⁴ cells ml ⁻¹	S	N, T	32-35	25	72 h EC50	Growth	4536.3	(Tai et al., 2010)
Algae	Skeletonema costatum	Dy	DyCl ₃	10 ⁴ cells ml ⁻¹	S	N, T	32-35	25	72 h EC50	Growth	4593.9	(Tai et al., 2010)
Algae	Skeletonema costatum	Но	HoCl ₃	10 ⁴ cells ml ⁻¹	S	N, T	32-35	25	72 h EC50	Growth	4829.9	(Tai et al., 2010)
Algae	Skeletonema costatum	Er	Er(NO ₃ ) ₃	10 ⁴ cells ml ⁻¹	S	N, T	32–35	25	72 h EC50	Growth	4962.1	(Tai et al., 2010)
Algae	Skeletonema costatum	Tm	TmCl ₃	10 ⁴ cells ml ⁻¹	S	N, T	32–35	25	72 h EC50	Growth	4866	(Tai et al., 2010)
Algae	Skeletonema costatum	Yb	YbCl ₃	10 ⁴ cells ml ⁻¹	S	N, T	32–35	25	72 h EC50	Growth	4935.7	(Tai et al., 2010)
Algae	Skeletonema costatum	Lu	LuCl ₃	10 ⁴ cells ml ⁻¹	S	N, T	32-35	25	72 h EC50	Growth	5008.5	(Tai et al., 2010)
Copepod	Acartia tonsa	Се	_	6-8 d	S	_	28	_	48 h LC50	Survival	150	(Bowmer et al., 1992) ^{SD}
Copepod	Acartia tonsa	Dy	_	6-8 d	S	=	28	_	48 h LC50	Survival	3600	(Bowmer et al., 1992) ^{SD}
Copepod	Acartia tonsa	Gd	-	6-8 d	S	_	28	-	48 h LC50	Survival	520	(Bowmer et al., 1992) ^{SD}
Copepod	Acartia tonsa	La	_	6-8 d	S	-	28	_	48 h LC50	Survival	1040	(Bowmer et al., 1992) ^{SD}
Copepod	Acartia tonsa	Nd	_	6-8 d	S	-	28	_	48 h LC50	Survival	850	(Bowmer et al., 1992) ^{SD}
Copepod	Acartia tonsa	Pr	_	6-8 d	S	-	28	_	48 h LC50	Survival	920	(Bowmer et al., 1992) ^{SD}
Copepod	Acartia tonsa	Sm	=	6-8 d	S	=	28	=	48 h LC50	Survival	420	(Bowmer et al., 1992) ^{SD}
Fish	Poecilia reticulata	Ce	_	0.15 g, 1.9 cm	SR	_	28	21-25	96 h LC50	Survival	11200	(Hooftman et al., 1992) ^{SD}
Fish	Poecilia reticulata	Dy	_	0.15 g, 1.9 cm	SR	_	28	21-25	96 h LC50	Survival	15400	(Hooftman et al., 1992) ^{SD}
Fish	Poecilia reticulata	Gd		0.15 g, 1.9 cm	SR		28	21-25	96 h LC50	Survival	10800	(Hooftman et al., 1992) ^{SD}
Fish	Poecilia reticulata	La	_	0.15 g, 1.9 cm	SR	_	28	21-25	96 h LC50	Survival	47000	(Hooftman et al., 1992) ^{SD}
Fish	Poecilia reticulata	Nd	_	0.15 g, 1.9 cm	SR	=	28	21-25	96 h LC50	Survival	9600	(Hooftman et al., 1992) ^{SD}
Fish	Poecilia reticulata	Pr	_	0.15 g, 1.9 cm	SR	_	28	21-25	96 h LC50	Survival	4500	(Hooftman et al., 1992) ^{SD}
Fish	Poecilia reticulata	Sm		0.15 g, 1.9 cm	SR	-	28	21-25	96 h LC50	Survival	10600	(Hooftman et al., 1992) ^{SD}

^{*}S = static; SR = static-renewal; N = nominal; T = total; SD = secondary data, whereas unmarked references are primary data

Table 7. Bioaccumulation of lanthanides by freshwater organisms

Organism	Species	Base element	Compound	Hardness (mg CaCO ₃ l ⁻¹ )	Duration	Tissue	рH	Exposure conc. ^a (µg l ⁻¹ )	Tissue conc. ^b (μg REE g ⁻¹ )	BCF ^c (L kg ⁻¹ )	Reference*
Organism	Species	Cicincia	Сотроина	(ing cacosi )	Duration	Whole	PII	( <b>PS</b> 1 )	(pg REE g )		
Duckweed	Lemna minor	La	LaCl ₃	54.8	48 h	plant	5.1-5.6	1.39	1.7	826.3 ^d	(Weltje et al., 2002a)
Fish	Cyprinus carpio	La	La(NO ₃ ) ₃	53-60	45 d	Skeleton	6	500	2.8	5.6	(Tu et al., 1994)
Fish	Cyprinus carpio	La	La(NO ₃ ) ₃	53-60	45 d	Muscle	6	500	1.3	2.6	(Tu et al., 1994)
Fish	Cyprinus carpio	La	La(NO ₃ ) ₃	53-60	45 d	Gill	6	500	7	13.9	(Tu et al., 1994)
Fish	Cyprinus carpio	La	La(NO ₃ ) ₃	53-60	45 d	Internal organs	6	500	38.9	77.8	(Tu et al., 1994)
Fish	Cyprinus carpio	Gd	Gd(NO ₃ ) ₃	53-60	45 d	Skeleton	6	500	2.3	4.6	(Tu et al., 1994)
Fish	Cyprinus carpio	Gd	Gd(NO ₃ ) ₃	53-60	45 d	Muscle	6	500	1.6	3.2	(Tu et al., 1994)
Fish	Cyprinus carpio	Gd	Gd(NO ₃ ) ₃	53-60	45 d	Gill	6	500	5.3	10.7	(Tu et al., 1994)
Fish	Cyprinus carpio	Gd	Gd(NO ₃ ) ₃	53-60	45 d	Internal organs	6	500	42.3	84.6	(Tu et al., 1994)
Fish	Cyprinus carpio	Ce	Ce(NO ₃ ) ₃	53-60	43 d	Muscle	6	270	0.05	0.2	(Sun et al., 1996)
Fish	Cyprinus carpio	Ce	Ce(NO ₃ ) ₃	53-60	43 d	Skeleton	6	270	1.6	5.9	(Sun et al., 1996)
Fish	Cyprinus carpio	Ce	Ce(NO ₃ ) ₃	53-60	43 d	Gill	6	270	3.5	12.8	(Sun et al., 1996)
Fish	Cyprinus carpio	Ce	Ce(NO ₃ ) ₃	53-60	43 d	Internal organs	6	270	164.2	608	(Sun et al., 1996)
Fish	Cyprinus carpio	La	La(NO ₃ ) ₃	53-60	43 d	Muscle	6	300	0.2	0.8	(Sun et al., 1996)
Fish	Cyprinus carpio	La	La(NO ₃ ) ₃	53-60	43 d	Skeleton	6	300	1.1	3.7	(Sun et al., 1996)
Fish	Cyprinus carpio	La	La(NO ₃ ) ₃	53-60	43 d	Gill	6	300	4.1	13.5	(Sun et al., 1996)
Fish	Cyprinus carpio	La	La(NO ₃ ) ₃	53-60	43 d	Internal organs	6	300	180.6	602	(Sun et al., 1996)
Fish	Cyprinus carpio	La	SmCl ₃	53-60	43 d	Muscle	6	250	0.3	1.1	(Sun et al., 1996)
Fish	Cyprinus carpio	La	SmCl ₃	53-60	43 d	Skeleton	6	250	1.4	5.5	(Sun et al., 1996)
Fish	Cyprinus carpio	La	SmCl ₃	53-60	43 d	Gill	6	250	4	16	(Sun et al., 1996)
Fish	Cyprinus carpio	La	SmCl ₃	53-60	43 d	Internal organs	6	250	176.3	705	(Sun et al., 1996)
Fish	Danio rerio	Ce	CeO ₂ nanoparticles	97.7	7 d	Liver	7.2	500	1350°	2700°	(Johnston et al., 2010)

^{*}All references are primary data

^aConcentrations are nominal and total of lanthanide in the exposure medium ^bTissue concentration is expressed in wet weight, except otherwise described

^cBCF is calculated from total or nominal REE concentrations in the water and expressed in wet weight, except otherwise described ^dBCF is dynamic i.e., accumulation not in equilibrium yet, is calculated as described in Weltje et al (2002) and expressed based on fresh weight of plant.

^eValues are calculated from dry weight of tissue.

 $\label{thm:constraints} \textbf{Table 8. Bioaccumulation of lanthanides by saltwater organisms in field sites} \\$ 

Organism	Species	Base element	Polluted / Unpolluted	Tissue	Tissue conc. (μg REE g ⁻¹ dw)	Reference*
Cephalopod	Nautilus macromphalus	La	Unpolluted	Digestive Gland	0.3	(Pernice et al., 2009)
Cephalopod	Nautilus macromphalus	La	Unpolluted	Pericardial appendages	0.1	(Pernice et al., 2009)
Cephalopod	Nautilus macromphalus	Ce	Unpolluted	Digestive Gland	0.3	(Pernice et al., 2009)
Cephalopod	Nautilus macromphalus	Ce	Unpolluted	Pericardial appendages	0.2	(Pernice et al., 2009)
Cephalopod	Nautilus macromphalus	Nd	Unpolluted	Digestive Gland	0.2	(Pernice et al., 2009)
Cephalopod	Nautilus macromphalus	Nd	Unpolluted	Pericardial appendages	0.2	(Pernice et al., 2009)
Cephalopod	Nautilus pompilius	La	Unpolluted	Digestive Gland	1	(Pernice et al., 2009)
Cephalopod	Nautilus pompilius	La	Unpolluted	Pericardial appendages	0.2	(Pernice et al., 2009)
Cephalopod	Nautilus pompilius	Ce	Unpolluted	Digestive Gland	1.6	(Pernice et al., 2009)
Cephalopod	Nautilus pompilius	Ce	Unpolluted	Pericardial appendages	0.4	(Pernice et al., 2009)
Cephalopod	Nautilus pompilius	Nd	Unpolluted	Digestive Gland	0.9	(Pernice et al., 2009)
Cephalopod	Nautilus pompilius	Nd	Unpolluted	Pericardial appendages	0.3	(Pernice et al., 2009)
Mussel	Mytilus edulis	La	Unpolluted	Kidney	0.4	(Lobel et al., 1991)
Mussel	Mytilus edulis	La	Unpolluted	Digestive Gland	0.3	(Lobel et al., 1991)
Mussel	Mytilus edulis	La	Unpolluted	Gills	0.2	(Lobel et al., 1991)
Mussel	Mytilus edulis	La	Unpolluted	Mantle	0.2	(Lobel et al., 1991)
Mussel	Mytilus edulis	La	Unpolluted	Foot	0.1	(Lobel et al., 1991)
Mussel	Mytilus edulis	Ce	Unpolluted	Kidney	0.5	(Lobel et al., 1991)
Mussel	Mytilus edulis	Ce	Unpolluted	Digestive Gland	0.5	(Lobel et al., 1991)
Mussel	Mytilus edulis	Ce	Unpolluted	Gills	0.2	(Lobel et al., 1991)
Mussel	Mytilus edulis	Ce	Unpolluted	Mantle	0.2	(Lobel et al., 1991)
Mussel	Mytilus edulis	Ce	Unpolluted	Foot	0.1	(Lobel et al., 1991)
Mussel	Mytilus edulis	La	Unpolluted	Soft tissue	3.7	(Riget et al., 1996)
Mussel	Mytilus edulis	Ce	Unpolluted	Soft tissue	4.7	(Riget et al., 1996)
Mussel	Mytilus edulis	Eu	Unpolluted	Soft tissue	0.02	(Riget et al., 1996)
Scallop	Chlamys varia	Ce	Polluted	Digestive Gland	10.6	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	Ce	Polluted	Kidney	1.9	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	Ce	Polluted	Gills	5.4	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	Ce	Polluted	Gonad	5.7	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	Ce	Polluted	Muscle	0.3	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	Ce	Polluted	Non-organ tissue	4	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	Ce	Unpolluted	Digestive Gland	2.2	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	Ce	Unpolluted	Kidney	0.2	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	Ce	Unpolluted	Gills	0.1	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	Ce	Unpolluted	Gonad	1.5	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	Ce	Unpolluted	Muscle	0.04	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	Ce	Unpolluted	Non-organ tissue	0.6	(Bustamante and Miramand, 2005)

 $Table \ 8. \ Bioaccumulation \ of \ lanthanides \ by \ saltwater \ organisms \ in \ field \ sites \ (con't)$ 

Organism	Species	Base element	Polluted / Unpolluted	Tissue	Tissue conc. (μg REE g ⁻¹ dw)	Reference*
Scallop	Chlamys varia	La	Polluted	Digestive Gland	7.9	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	La	Polluted	Kidney	1.8	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	La	Polluted	Gills	4	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	La	Polluted	Gonad	5.1	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	La	Polluted	Muscle	0.3	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	La	Polluted	Non-organ tissue	2.8	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	La	Unpolluted	Digestive Gland	0.2	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	La	Unpolluted	Kidney	0.2	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	La	Unpolluted	Gills	0.4	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	La	Unpolluted	Gonad	0.7	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	La	Unpolluted	Muscle	0.03	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	La	Unpolluted	Non-organ tissue	0.2	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	Nd	Polluted	Digestive Gland	5.4	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	Nd	Polluted	Kidney	0.7	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	Nd	Polluted	Gills	2	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	Nd	Polluted	Gonad	3.5	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	Nd	Polluted	Muscle	0.1	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	Nd	Polluted	Non-organ tissue	1.8	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	Nd	Unpolluted	Digestive Gland	0.9	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	Nd	Unpolluted	Kidney	0.1	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	Nd	Unpolluted	Gills	0.3	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	Nd	Unpolluted	Gonad	1.1	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	Nd	Unpolluted	Muscle	0.01	(Bustamante and Miramand, 2005)
Scallop	Chlamys varia	Nd	Unpolluted	Non-organ tissue	0.3	(Bustamante and Miramand, 2005)

^{*}All references are primary data

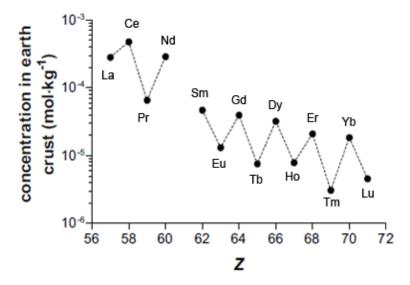


Fig. 1. Concentration (mol kg⁻¹) of lanthanides in the earth crust versus atomic number (Z). Data is obtained from Lide (Lide, 1994) and graph is modified from Weltje (Weltje, 2002). Pm has no stable or long-lived isotopes, hence no natural concentration is shown on the graph.

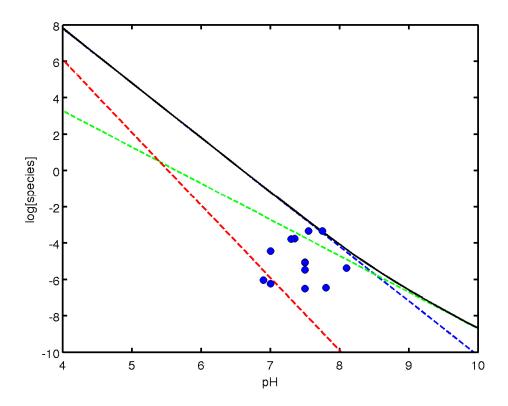


Fig. 2. Solubility of  $La(OH)_{3(s)}$ . The black line corresponds to equilibrium between  $La(OH)_{3(s)}$  and aqueous solution. Above this line Lanthanum hydroxide is supersaturated. The blue lines corresponds to  $[La^{3+}]$  in equilibrium with solid, green line is  $LaOH^{2+}$  and the red line is  $La_2OH_2^{2+}$  in equilibrium with the solid phase. The blue dots are La toxicity endpoints and pH values from Table 4 of this report.

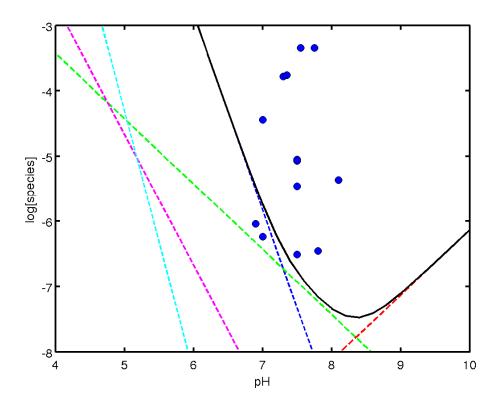


Fig. 3. Solubility of  $La_2(CO3)_{3(s)}$ . The black line corresponds to equilibrium between  $La_2(CO_3)_{3(s)}$  and aqueous solution. Above this line Lanthanum carbonate is supersaturated. The blue lines corresponds to  $[La^{3+}]$  in equilibrium with solid, green line is  $[LaCO_3^+]$  and the red line is  $[La(CO_3)_2^-]$  in equilibrium with the solid phase. The magenta line corresponds to  $[LaHCO_3^{2+}]$  and the cyan line corresponds to  $[La_2CO_3^{4+}]$ . The blue dots are La toxicity endpoints and pH values from Table 4 of this report.

### References

- Alberts, A.S., Smit, B.J., Louw, W.K., van Rensburg, A.J., van Beek, A., Kritzinger, V., Nel, J.S., 1997. Dose response relationship and multiple dose efficacy and toxicity of samarium-153-EDTMP in metastatic cancer to bone. Radiother Oncol. 43, 175-179.
- Annema, J.A., 1990. Zeldzame aarden. Stichting Natuur en Milieu. Report No. SNM-MB 90/1 Utrecht
- Bakkenist, S., Van De Wiel, B., 1995. Bepaling van het verspreidingspatroon van de lozingen van Hydro Agri en Kemira Permis met lanthaniden als tracer, Stageverslag Landbouwuniversiteit Wageningen
- Barry, M.J., Meehan, B.J., 2000. The acute and chronic toxicity of lanthanum to Daphnia carinata. Chemosphere. 41, 1669-1674.
- Blaise, C., Gagne, F., Ferard, J.F., Eullaffroy, P., 2008. Ecotoxicity of selected nano-materials to aquatic organisms. Environ Toxicol. 23, 591-598.
- Bowmer, C.T., Kauffman-Van Bommel, J.A., Degeling, C., 1992. Semi-static acute toxicity test with metal-chloride salts and the marine copepod Acartia tonsa. Report No. TNO report IMW 92/342 t/m 349
- Brookins, D.G., 1989. Aqueous geochemistry of rare earth elements. in: Lipin, B.R., McKay, G.A. (Eds.) Reviews in Mineralogy. Mineralogical Society of America, Washington D.C., USA, pp. 201-222.
- Bustamante, P., Miramand, P., 2005. Subcellular and body distributions of 17 trace elements in the variegated scallop Chlamys varia from the French coast of the Bay of Biscay. Sci Total Environ. 337, 59-73.
- Cantrell, K.J., Byrne, R.H., 1987. Rare earth element complexation by carbonate and oxalate ions. Geochem Cosmochim Acta. 51, 591-606.
- Chapman, K.K., Benton, M., Brinkhurst, R.O., Scheuerman, P.R., 1999. Use of the aquatic oligochaetes *Lumbriculus variegatus* and *Tubifex tubifex* for assessing the toxicity of copper and cadmium in a spiked-artificial sediment toxicity test. Environ. Toxicol. 14, 271-278.
- Cotton, S., 1991. Lanthanides and actinides, MacMillan Education Ltd., London, UK.
- Das, T., Sharma, A., Talukder, G., 1988. Effects of lanthanum in cellular systems. A review. Biol Trace Elem Res. 18, 201-228.
- Den-Ouden, A.C.P.H., 1995. Aquatisch ecotoxicologisch onderzoek van zeldzame aardmetalen. Report No. TNO-MW, Delft
- Di Toro, D.M., Allen, H.E., Bergman, H.L., Meyer, J.S., Paquin, P.R., Santore, R.C., 2001. Biotic ligand model of the acute toxicity of metals. 1. Technical basis. Environ. Toxicol. Chem. 20, 2383-2396.
- EPA, 1994. ECO Update Catalogue of Standard Toxicity Tests for Ecological Risk Assessment, US Environmental Protection Agency
- Filipi, R., Nesmerak, K., Rucki, M., Roth, Z., Hanzlikova, I., Tichy, M., 2007. Acute toxicity of rare earth elements and their compounds. CHEM LISTY. 101, 793-798.
- Flapper, T., 2003. Research and Development Program. Project completion report Phoslock trial at Fyshwick STP Lagoon. Report No. Ecowise No. A011389
- Fuma, S., Takeda, H., Miyamoto, K., Yanagisawa, K., Inoue, Y., Ishii, N., Sugai, K., Ishii, C., Kawabata, Z., 2001. Ecological evaluation of gadolinium toxicity compared with other heavy metals using an aquatic microcosm. Bull Environ Contam Toxicol. 66, 231-238.

- Gaiser, B.K., Fernandes, T.F., Jepson, M., Lead, J.R., Tyler, C.R., Stone, V., 2009. Assessing exposure, uptake and toxicity of silver and cerium dioxide nanoparticles from contaminated environments. Environ Health. 8 Suppl 1, S2.
- Grosell, M., Gerdes, R.M., Brix, K.V., 2006. Chronic toxicity of lead to freshwater invertebrates *Branchionus calyciflorus, Chironomus tetans*, and *Lymnaea stagnalis*. Environ. Toxicol. Chem. 25, 97-104.
- Guo, H.Y., Chen, L., Wang, X.R., Chen, Y., 2002. Physiological responses of Carassius auratus to ytterbium exposure. Ecotoxicol Environ Saf. 53, 312-316.
- Haghseresht, F., 2006. The availability of lanthanum ions from Phoslock: Residual concentrations following Phoslock applications & implications for consumption Report No. Report No. PLT-06-2
- Hooftman, R.N., Roza, P., Stofberg, E.M., 1992. Semi-static acute toxicity test with metal-chloride salts and the fish species Poecilia reticulata (OECD Guideline no. 203, 96 h). Report No. TNO IMW-R 92/361 t/m 368
- Humphris, S.E. (Ed.) 1984. The mobility of the rare earths in the crust, Vol. Elsevier Science Publishing, Amsterdam, Germany
- Johnston, B.D., Scown, T.M., Moger, J., Cumberland, S.A., Baalousha, M., Linge, K., van Aerle, R., Jarvis, K., Lead, J.R., Tyler, C.R., 2010. Bioavailability of Nanoscale Metal Oxides TiO(2), CeO(2), and ZnO to Fish. Environ Sci Technol. 44, 1144-1151.
- Laville, N., Ait-Aissa, S., Gomez, E., Casellas, C., Porcher, J.M., 2004. Effects of human pharmaceuticals on cytotoxicity, EROD activity and ROS production in fish hepatocytes. Toxicology. 196, 41-55.
- Lide, D.R. (Ed.) 1994. Handbook of Chemistry and Physics, Vol. CRC Press, Boca Raton, FL, USA
- Liu, X.W., Byrne, R.H., 1997. Rare earth and yttrium phosphate solubilities in aqueous solution. Geochim Cosmochim Acta. 61, 1625-1633.
- Lobel, P.B., Longerich, H.P., Jackson, S.E., Belkhode, S.P., 1991. A major factor contributing to the high degree of unexplained variability of some elements concentrations in biological tissue: 27 elements in 5 organs of the mussel Mytilus as a model. Arch Environ Contam Toxicol. 21, 118-125.
- Lürling, M., Tolman, Y., 2010. Effects of lanthanum and lanthanum-modified clay on growth, survival and reproduction of Daphnia magna. Water Res. 44, 309-319.
- Maas, J.L., Botterweg, J., 1993. Milieubezwaarlijkheid van zeldzame aardmetalen (lanthaniden, yttrium, scandium) in oppervlaktewater. Report No. RIZAnota 93.018
- Marang, L., Reiller, P.E., Eidner, S., Kumke, M.U., Benedetti, M.F., 2008. Combining spectroscopic and potentiometric approaches to characterize competitive binding to humic substances. Environ Sci Technol. 42, 5094-5098.
- Martell, A.E., Smith, R.M. (Eds.), 1997-2001. Critically selected stability constants of metal complexes database, Vol. National Institute of Standards and Technology (NIST), Gaithersburg, MD, USA
- McIntosh, J., 2007. Phoslock application Lake Okareka Final Report. Report No. Environment Bay of Plenty Environmental Publication 2007/23, ISSN: 1175 9372
- Moeller, T., Vincenti, G., 1965. Observations on the rare earths LXXVII. J Inorg Nuc Chem. 27, 1477-1482.
- Molander, G.A., 1992. Application of lanthanide reagents in organic synthesis. Chem Rev. 92, 29-68.
- Mori, S., 1999. Iron acquisition by plants. Curr Opin Plant Biol. 2, 250-253.
- NICNAS, 2001. National Industrial Chemicals Notification and Assessment Scheme, Full Public Report Lanthanum Modified Clay. Report No. NA/899

- Pernice, M., Boucher, J., Boucher-Rodoni, R., Joannot, P., Bustamante, P., 2009. Comparative bioaccumulation of trace elements between Nautilus pompilius and Nautilus macromphalus (Cephalopoda: Nautiloidea) from Vanuatu and New Caledonia. Ecotox Environ Saf. 72, 365-371.
- Riget, F., Johansen, P., Asmund, G., 1996. Influence of length on element concentrations in blue mussels (Mytilus edulis). Mar Pollut Bull. 32, 745-751.
- Slooff, W., Bont, P.F.H., Van den Hoop, M.A.G.T., Janus, J.A., Annema, J.A., 1993. Exploratory report rare earth elements and their compounds. Report No. RIVM 710401025, National Institute of Public Health and the Environment, Bilthoven, The Netherlands
- Sneller, F.E.C., Kalf, D.F., Weltje, L., Van Wezel, A.P., 2000. Maximum Permissible Concentrations and Negligible Concentrations for Rare Earth Elements (REEs). Report No. RIVM 601501011, National Institute of Public Health and the Environment, Bilthoven
- Spyroulias, G.A., de Montauzon, D., Maisonat, A., Poilblanc, R., Coutsolelos, A.G., 1998. Synthesis, UV-visible and electrochemical studies of lipophilic and hydrophilic lanthanide (III) bis(porphyrinates). Inorgan Chim Acta. 275-276, 182-191.
- Stanley, J.K.J.R., Byrne, R.H., 1990. The influence of solution chemistry on REE uptake by Ulva lactuca L. in seawater. Geochim Cosmochim Acta. 54, 1587-1596.
- Stauber, J.L., 2000. Toxicity testing of modified clay leachates using freshwater organisms. Report No. ET/IR26R, Centre for Advanced Analytical Chemistry Energy Technology, CSIRO
- Stephan, C.E., Mount, D.I., Hansen, D.J., Gentile, J.H., Chapman, G.A., Brungs, W.A., 1985. Guidelines for Deriving Numeric National Water Quality Criteria for the Protection of Aquatic Organisms and Their Uses. Report No. EPA 882/R-85-100, Office of Research and Development, Environmental Research Laboratories, US Environmental Protection Agency, Duluth, MN, Narragansett, RI, Corvallis, OR
- Stronkhorst, J., Yland, E.M.L., 1998. De biologische beschikbaarheid van zeldzame aardmetalen in Nieuwe Maas-sediment voor de Slijkgarnaal Corophium volutator. Report No. RIKZwerkdocument AB-98.103x
- Sun, H.A.O., Wang, X.R., Hua, Z.Z., Wu, C.H., Wang, L.S., Dai, L.M., Li, Z., Chen, Y.J., 1996. Bioconcentration and elimination of five light rare earth elements in carp (Cyprinus carpio L.). Chemosphere. 33, 1475-1483.
- Syproulias, G.A., de Montauzon, D., Maisonat, A., Poilblanc, R., Coutsolelos, A.G., 1998. Synthesis, UV-visible and electrochemical studies of lipophilic and hydrophilic lanthanide (III) bis(porphyrinates). Inorgan Chim Acta. 275-276, 182-191.
- Tai, P.D., Zhao, Q., Su, D., Li, P.J., Stagnitti, F., 2010. Biological toxicity of lanthanide elements on algae. Chemosphere. 80, 1031-1035.
- Tu, Q., Wang, X.R., Tian, L.Q., Dai, L.M., 1994. Bioaccumulation of the rare earth elements lanthanum, gadolinium and yttrium in carp (Cyprinus carpio). Environ Pollut. 85, 345-350.
- Watson-Leung, T., 2009. Phoslock Toxicity Testing with Three Sediment Dwelling Organisms (Hyalella azteca, Hexagenia spp. and Chironomus dilutus) and Two Water Column Dwelling Organisms (Rainbow Trout and Daphnia magna), Aquatic Toxicology Unit, Laboratory Services Branch, Ontario Ministry of the Environment Etobicoke
- Weltje, L., 2002. Bioavailability of lanthanides to freshwater organisms: Speciation, Accumulation and Toxicity. PhD, Delft University of Technology
- Weltje, L., Brouwer, A.H., Verburg, T.G., Wolterbeek, H.T., De Goeu, J.J.M., 2002a. Accumulation and elimination of lanthanum by duckweed (Lemna minor L.) as influenced by organism growth and lanthanum sorption to glass. Environ Toxicol Chem. 21, 1483-1489.

Weltje, L., Heidenreich, H., Zhu, W., Wolterbeek, H.T., Korhammer, S., de Goeij, J.J., Markert, B., 2002b. Lanthanide concentrations in freshwater plants and molluscs, related to those in surface water, pore water and sediment. A case study in The Netherlands. Sci Total Environ. 286, 191-214.

# Appendix A.

**Derivation of solubility curves for La** 

The potential precipitation of two solid phases, hydroxide and carbonate, are investigated. The calculations presented below use thermodynamic constants from NIST (Martell and Smith, 2004). In all cases the lowest available ionic strength value was selected and temperature of 25 °C. The calculations assume the solid phase is in equilibrium with all dissolved species. Lanthanum is selected as the test lanthanide for this analysis.

### 0.1 Lanthanum Hydroxide Precipitation

For the dissolution reaction:

$$La(OH)_{3(g)} = La^{3+} + 3OH^{-}$$

with known solubility product  $(K_{sp})$  and dissociation constant for water  $(K_w)$ , the concentration of La³⁺ versus pH can be calculated as follows:

$$\begin{split} K_{sp} &= [\text{La}^{3+}][\text{OH}^{-}]^{3} \\ K_{sp} &= \frac{[\text{La}^{3+}]Kw^{3}}{[\text{H}]^{3}} \\ \log K_{sp} &= \log [\text{La}^{3+}] + 3\log Kw - 3\log [\text{H}^{+}] \\ \log [\text{La}^{3+}] &= \log K_{sp} - 3\log Kw + 3\log [\text{H}^{+}] \\ \log [\text{La}^{3+}] &= \log K_{sp} - 3\log Kw - 3pH \end{split}$$

This is a linear equation determining log [La³⁺] as a function of pH with slope -3 and intercept log  $K_{sp}$  –  $3 \log Kw$ .

Similarly the concentration of logLaOH²⁺ can be calculated from the association constant ( $\beta_1$ )

$$\beta_1 = \frac{[\text{LaOH}^{2+}]}{[\text{La}^{3+}][\text{OH}^{-}]} = = \frac{[\text{LaOH}^{2+}][\text{H}^{+}]}{[\text{La}^{3+}]K_w}$$
$$\log[\text{LaOH}^{2+}] = \log \beta_1 + \log Kw + \log[\text{La}^{3+}] + pH$$

after substitution of  $\log [\mathrm{La}^{3+}]$  expression

$$\log [\text{LaOH}^{2+}] = \log \beta_1 + \log Kw + (\log K_{sp} - 3\log Kw - 3pH) + pH$$
  
 $\log [\text{LaOH}^{2+}] = \log \beta_1 + \log K_{sp} - 2\log Kw - 2pH$ 

The final NIST reported lanthanum hydrolysis product is  $La_2OH_2^{4+}$ . The symbol for the hydroxide formation constant can be given as  $\beta_{22}$  and the concentration can be derived as a function of pH:

$$\log [\text{LaOH}^{2+}] = \log \beta_{22} + 2 \log K_{sp} - 4 \log Kw - 4pH$$

Figure (XX) in the body of the report was derived using these equations and the following equilibrium constant values:

symbol	reaction	logK	ionic strength (M)
$K_{sp}$	La(OH) _{3(s)} =La ³⁺ +3OH -	-22.2	0
$\beta_1$	$La^{3+} + OH^{-} = LaOH^{2+}$	5.5	2.0
B22	$2La^{3+} + 2OH^{-} = La_2OH_2^{4+}$	10.5	2.0

Table 1: Hydrolysis constants for Lanthanides. Values from NIST (Martell and Smith, 2004). All values 25 °C.

#### 0.2 Lanthanum Carbonate Precipitation

For the dissolution reaction:

$${\rm La_2(CO_3)_{3(s)}} = 2 \, {\rm La^{\,3+}} + 3 {\rm CO_3^{\,2-}}$$

with known solubility product  $(K_{sp})$  and dissociation constant for water  $(K_w)$ , the concentration of La³⁺ versus pH can be calculated as follows:

$$\log\left[\mathrm{La^{3+}}\right] = \frac{\log K_{sp} - 3\log\left[\mathrm{CO_3^{2-}}\right]}{2}$$

Carbonate concentration can be calculated as a function of pH by assuming a fixed partial pressure of  $CO_{2(g)}$  and associated carbonate equilibria:

$$\log[CO_3^{(2-)}] = \log K_{a1} + \log K_{a2} + \log K_H + \log P_{CO_a} + 2pH$$

Using these La $^{3+}$  and CO $_{3}^{2-}$  relationshipps, additional carbonate complexes can be calculated as ...

$$\begin{split} \log \left[ \mathrm{LaCO_3^+} \right] &= \log \beta_1 + \log \left[ \mathrm{La^{3+}} \right] + \log \left[ \mathrm{CO_3^{2-}} \right] \\ \\ \log \left[ \mathrm{La(CO_3)_2^-} \right] &= \log \beta_2 + \log \left[ \mathrm{La^{3+}} \right] + 2 \log \left[ \mathrm{CO_3^{2-}} \right] \\ \\ \log \left[ \mathrm{La_2(CO_3)^{4+}} \right] &= \log \beta_{21} + 2 * \log \left[ \mathrm{La^{3+}} \right] + \log \left[ \mathrm{CO_3^{2-}} \right] \end{split}$$

Bicarbonate complex can be calculated using the following derived relationships ...

$$\begin{split} \log\left[\mathrm{LaHCO_3^{2+}}\right] &= \log\beta_{H1} + \log\left[\mathrm{La^{3+}}\right] + \log\left[\mathrm{HCO_3^{-}}\right] \\ &\log\left[\mathrm{HCO_3^{-}}\right] = \log K_{a1} + \log K_{H} + \log P_{\mathrm{CO_2}} + pH \end{split}$$

symbol	reaction	logK	ionic strength (M)
$K_{sp}$	$La_3CO_{32(s)} = 3La^{3+} + 2CO_3^{2-}$	-34.4	0.0
$\beta_1$	$La^{3+} + CO_3^{2-} = LaCO_3^+$	6.98	0.0
$\beta_2$	$La^{3+} + 2CO_3^{2-} = La(CO_3)^{-}$	11.86	0.0
$\beta_{H1}$	$La^{3+} + HCO_3^- = LaHCO_3^{2+}$	1.41	3.0
$\beta_{21}$	$2La^{3+} + CO_3^{2-} = La_2CO_3^{4+}$	6.92	3.0
$K_H$	$CO_2+H_2O=H_2CO_3$	-1.47	0.0
$K_{a1}$	$H_2CO_3=H^++HCO_3^-$	-6.35	0.0
$K_{a2}$	$HCO_3^- = H^+ + CO_3^{2-}$	-10.33	0.0

Table 2: Carbonate equilibria and Lanthanide. Values from NIST (Martell and Smith, 2004). All values 25 °C.

## References

Martell, A. E., Smith, R. M., 2004. NIST standard reference database 46 version 8.0. Database software developed by R. J. Motekaitis, gaithersburg, MD 20899, USA.